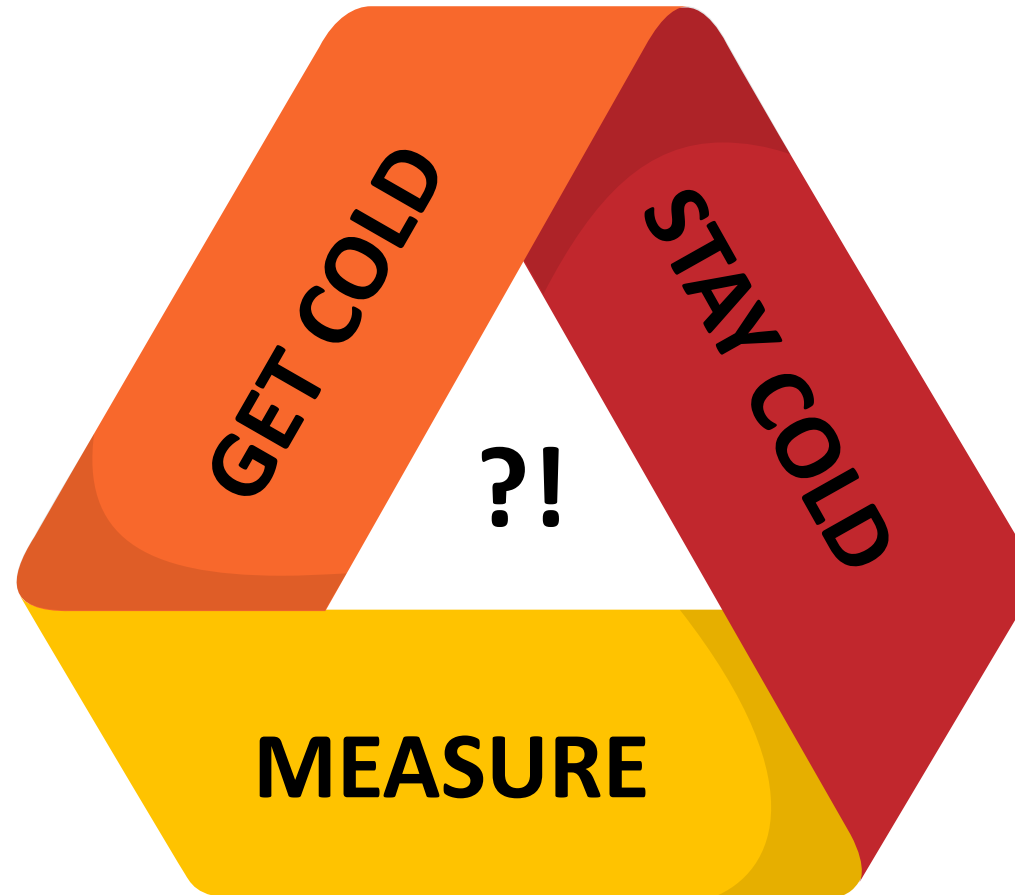




Cryostat design below 1K



Initial statement of problem

Choose a refrigerator that can get colder than the temperature needed by your experiment T_E

Cool the sample through some thermal link:

- sample must cool in a reasonable time
- thermal contact must be good enough that Q_E doesn't warm the experiment above T_E

GET COLD

?!

STAY COLD

MEASURE

Choose required measurement setup and materials (optics, particle beam, AC, DC wiring, magnetic field, gravity...)

The refrigerator must have sufficient cooling power to absorb the power dissipated by the measurement Q_E and still maintain T_E

Choose appropriate thermometer (resolution, stability, transferability, thermal contact, power dissipation of the thermometer)

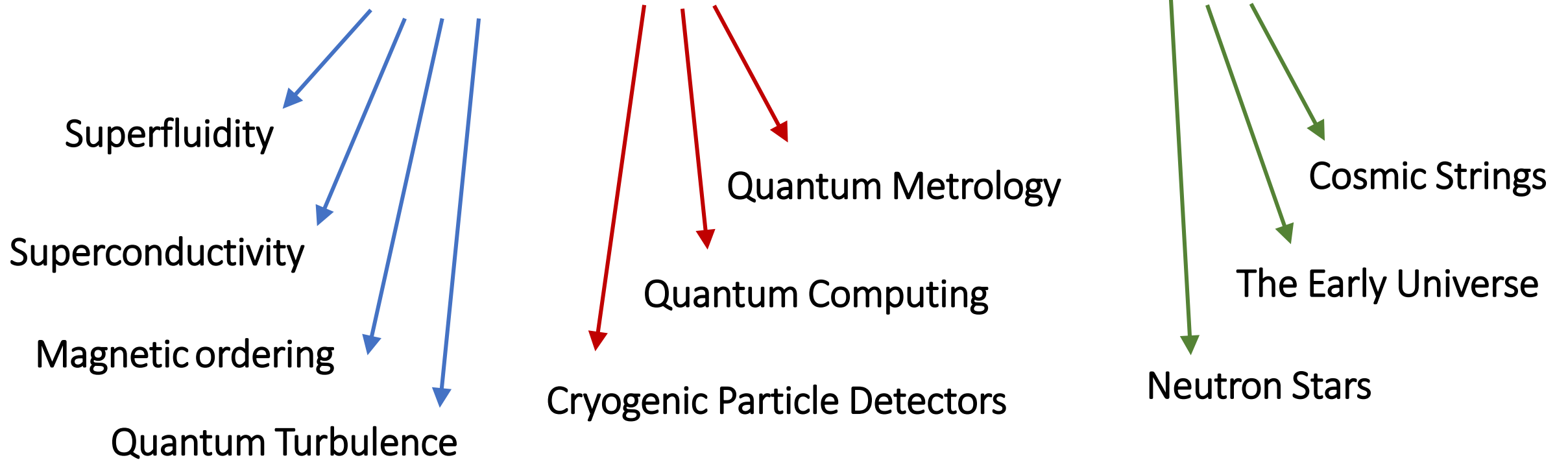
Reasons to reach low temperatures

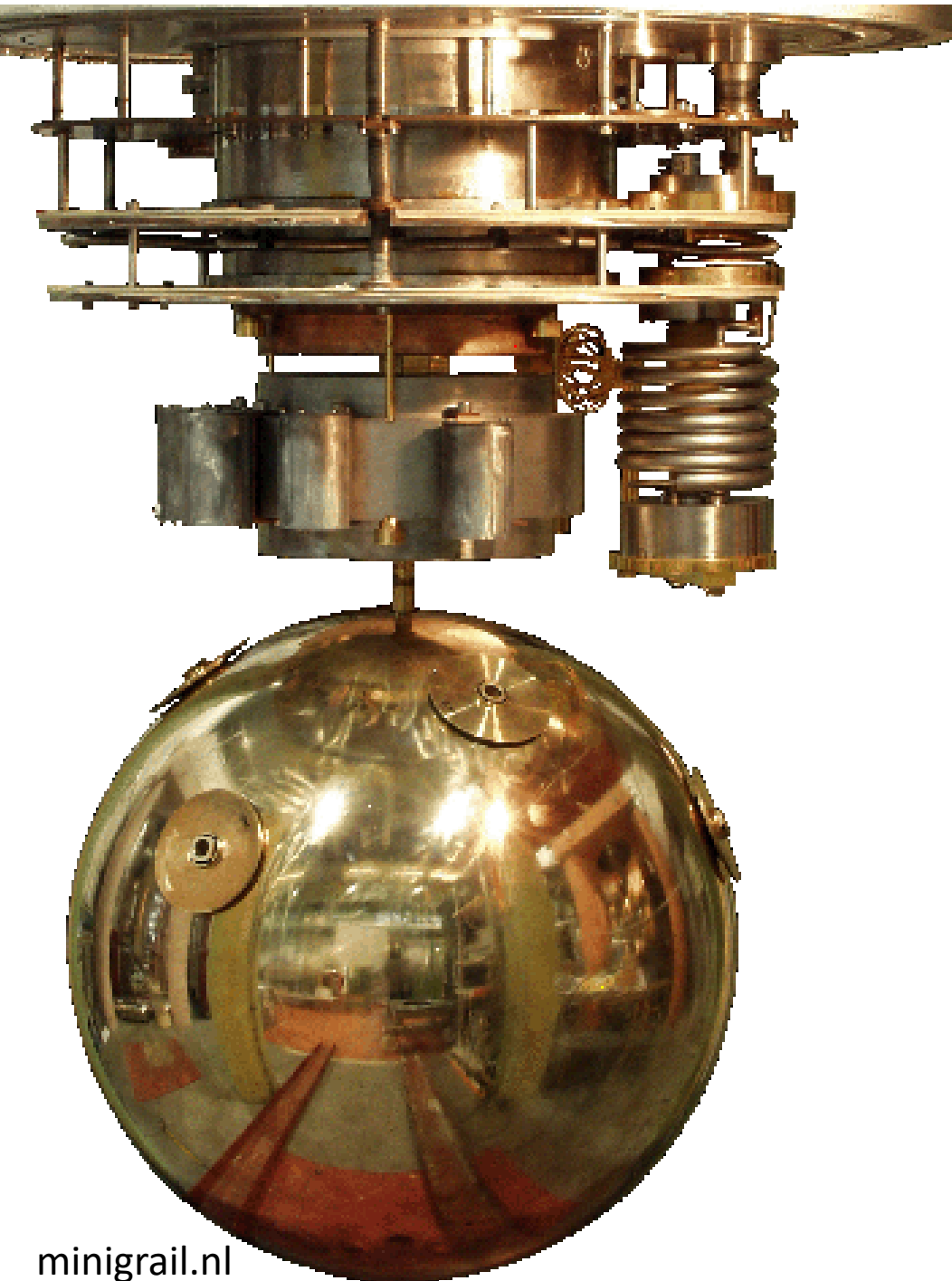
selective freezing of degrees of freedom

systems with small energies

low noise measurements

new phenomena, new technologies and modelling inaccessible systems

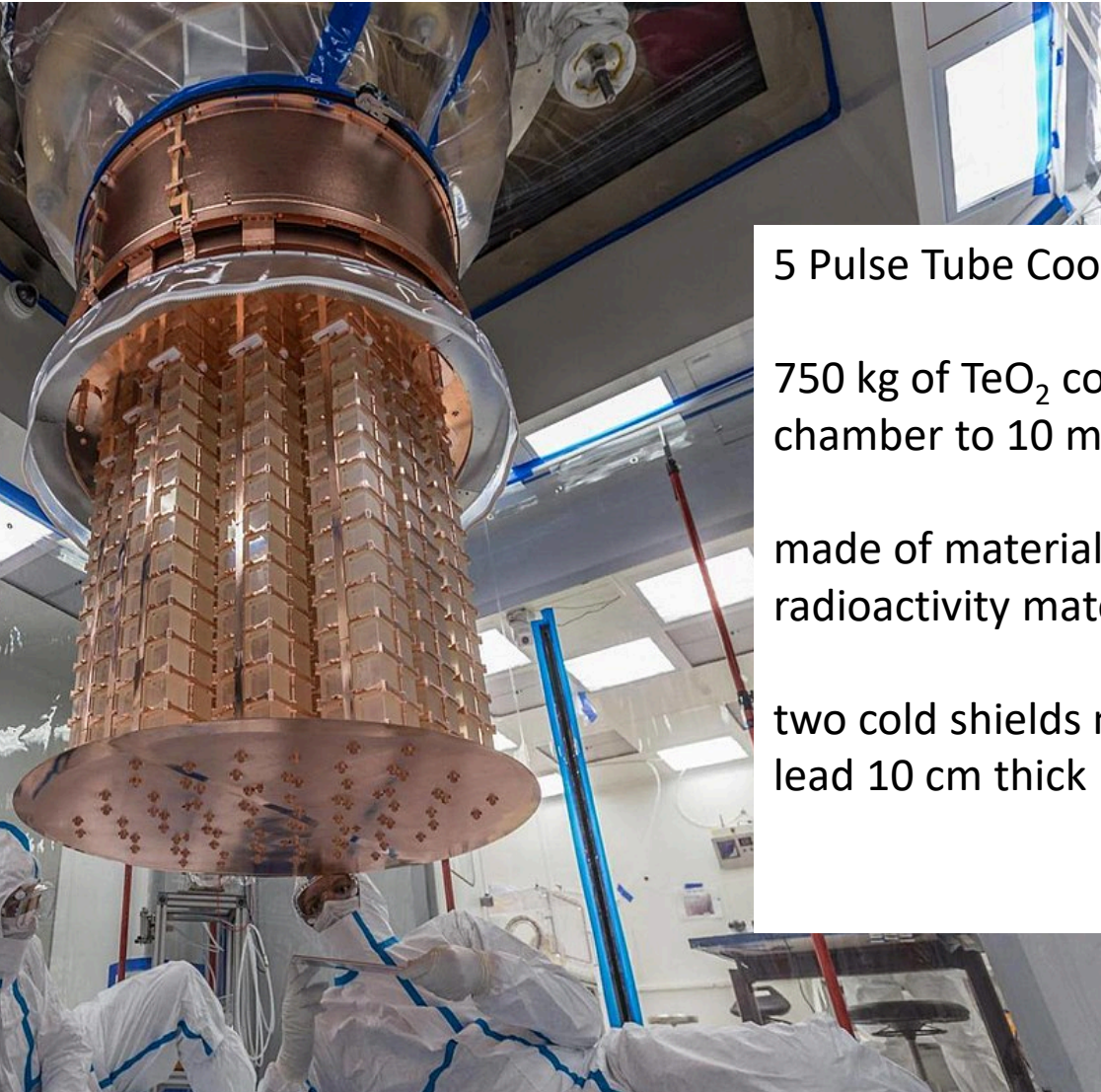




MiniGRAIL - Gravitational Radiation Antenna In Leiden

- The MiniGRAIL detector is a 68 cm diameter spherical gravitational wave antenna made of CuAl (6%) alloy.
- Mass of 1400 Kg.
- The antenna will operate at a temperature of 20 mK.

Cuore - Cryogenic Underground Observatory for Rare Events

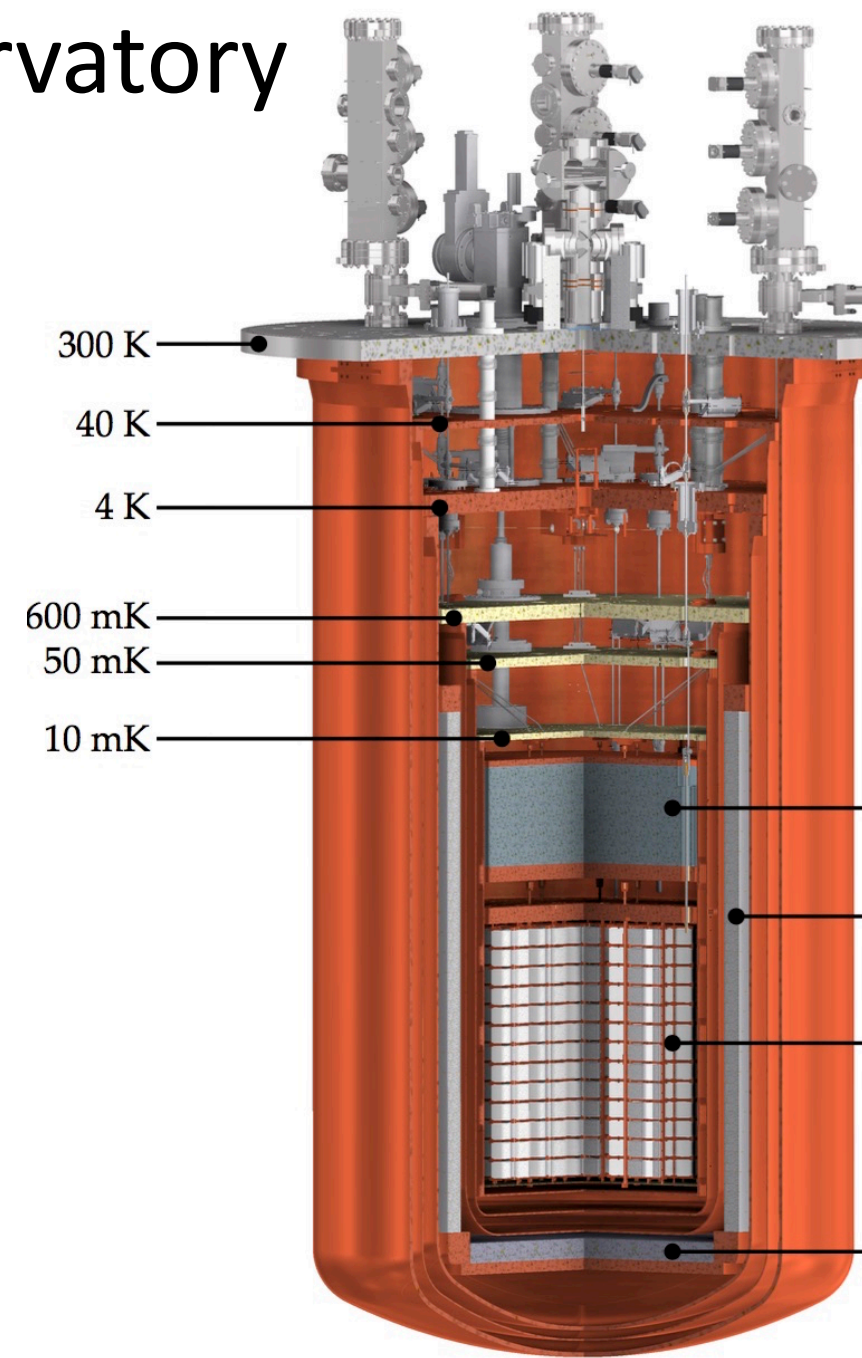


5 Pulse Tube Coolers

750 kg of TeO_2 cooled by the mixing chamber to 10 mK

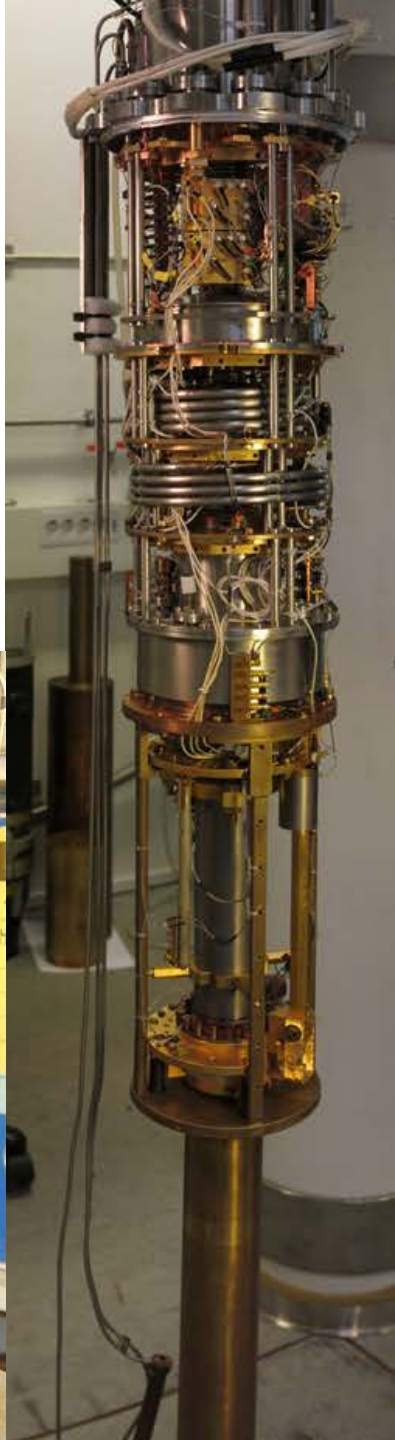
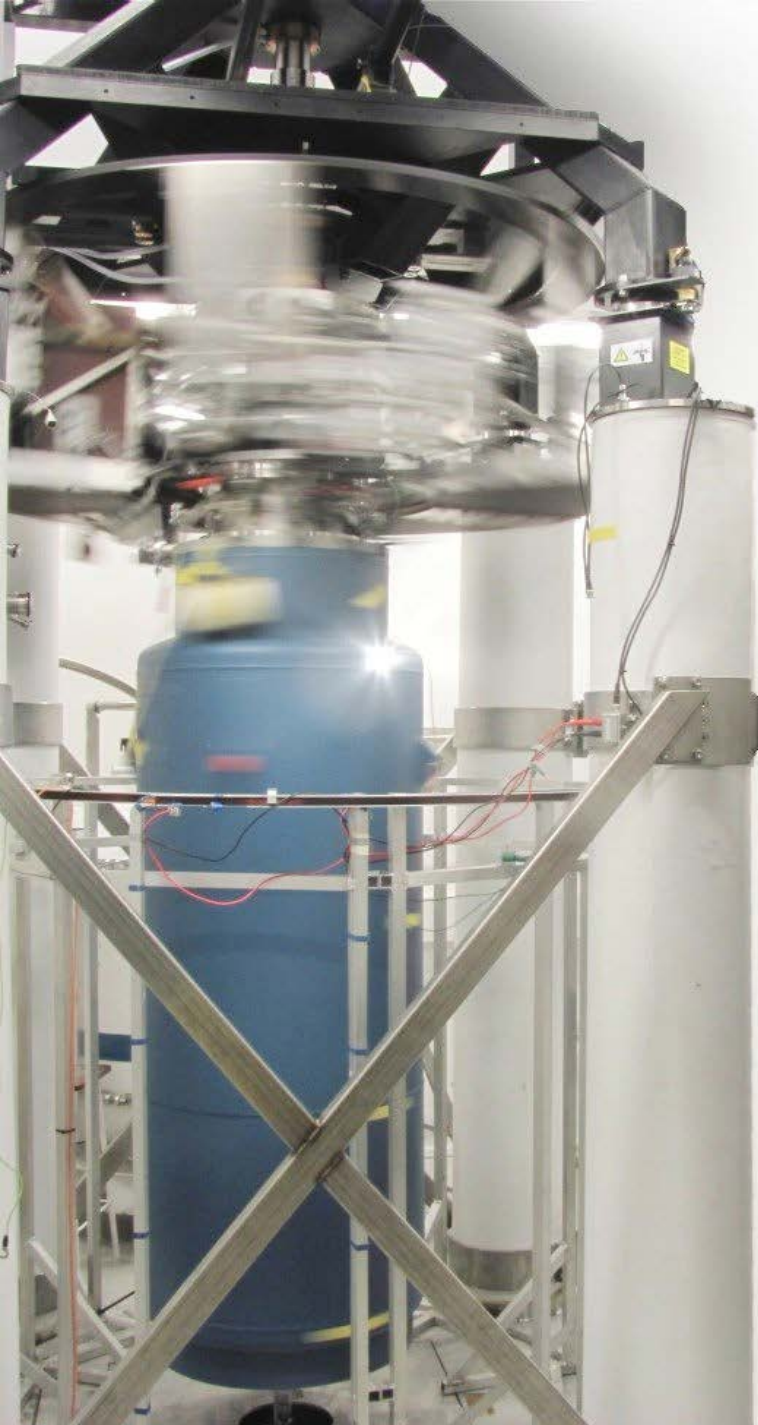
made of materials with low level radioactivity materials

two cold shields made of ancient roman lead 10 cm thick



ROTA – rotating cryostat for superfluid ^3He research

- Adiabatic Nuclear Demagnetisation Cryostat (combination of dilution refrigerator and nuclear cooling stage)
- Superfluid ^3He temperature of $0.15T_c$ ($\sim 140\mu\text{K}$ at zero bar pressure)
- Rotation velocity of $\sim 4\text{ rad/s}$
- Has a cryopump to collect ^3He - ^4He mixture during rotation



The cryostat for a 50 qubit quantum computer



An IBM quantum computer cryostat (Image: Andy Aaron, IBM Research/Flickr)

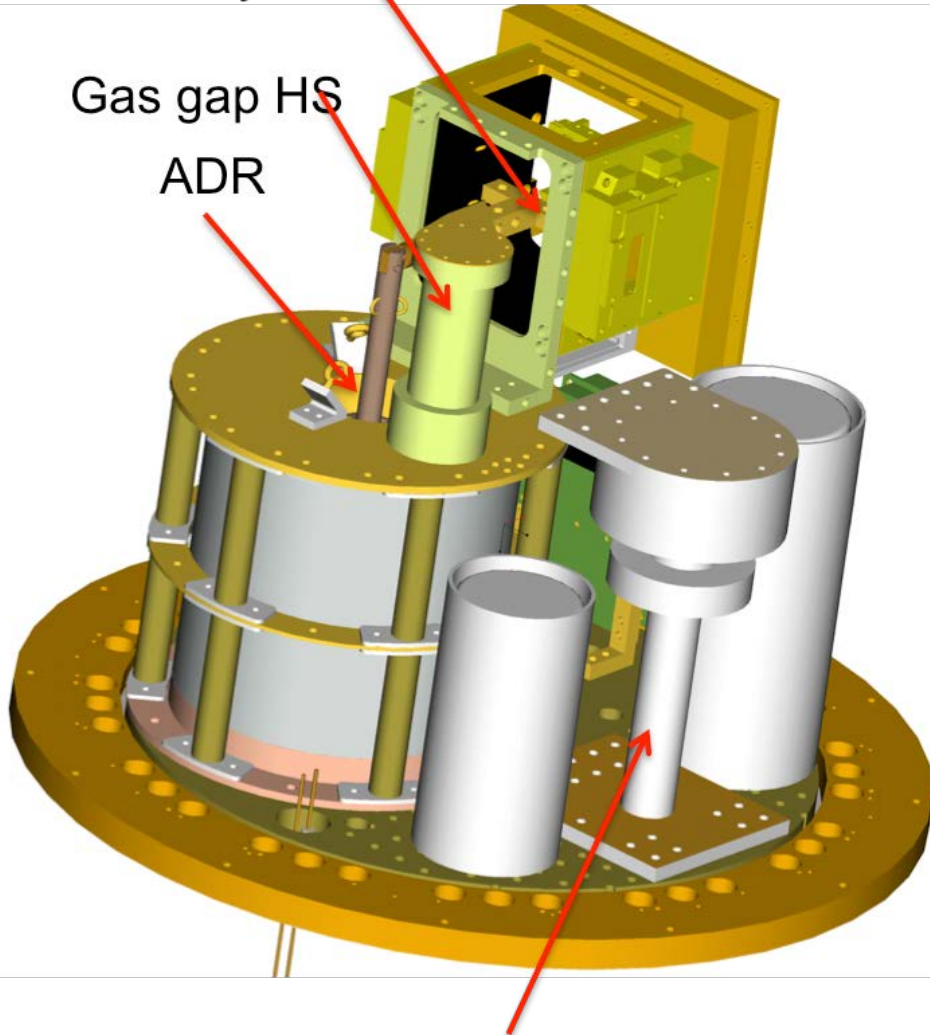


Fully Automatic ADR for Space Missions (50 mK)

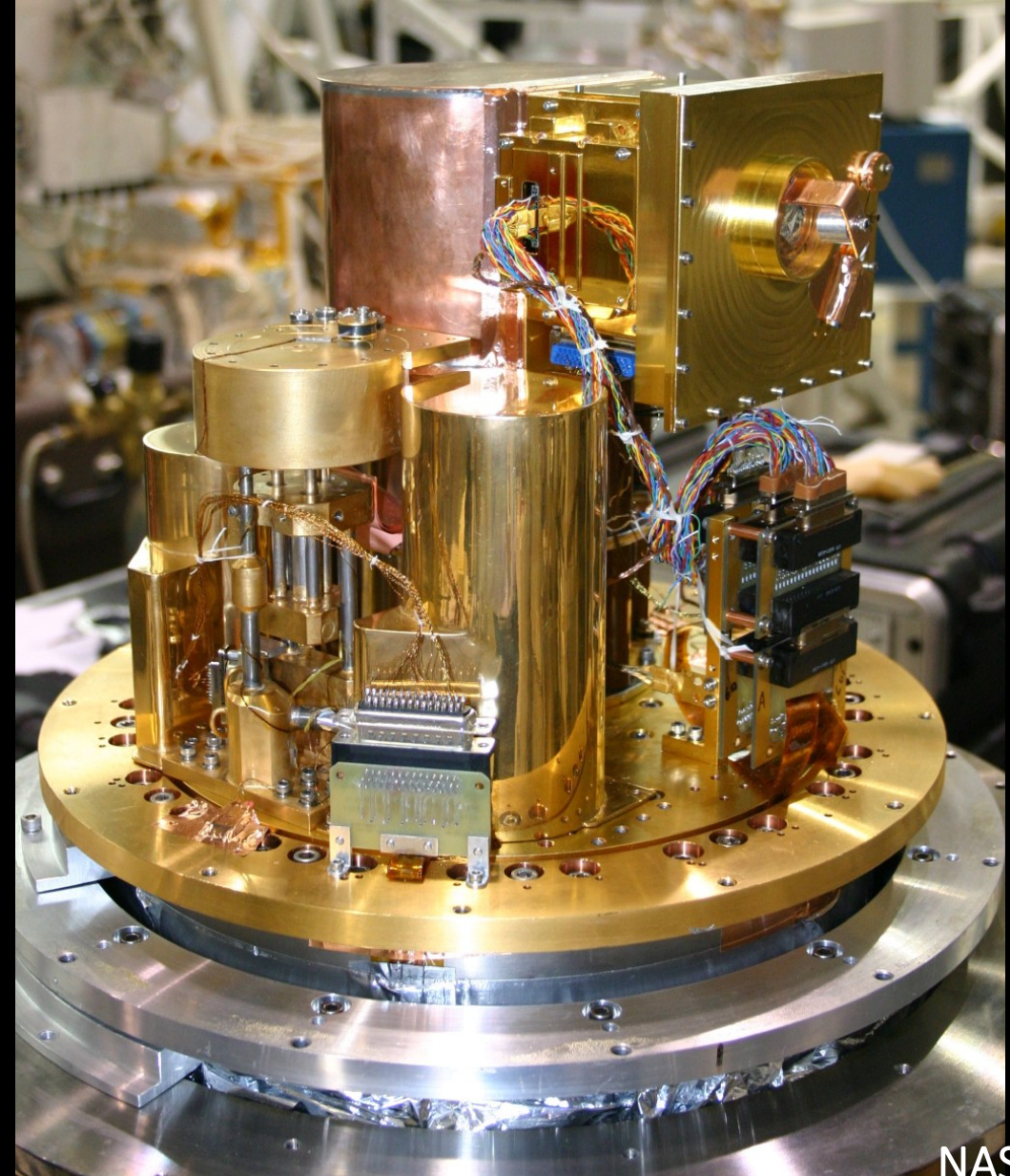
Detector system

Gas gap HS

ADR



He3/He4 closed cycle refrigerator



Temperature Range and Cooling Methods

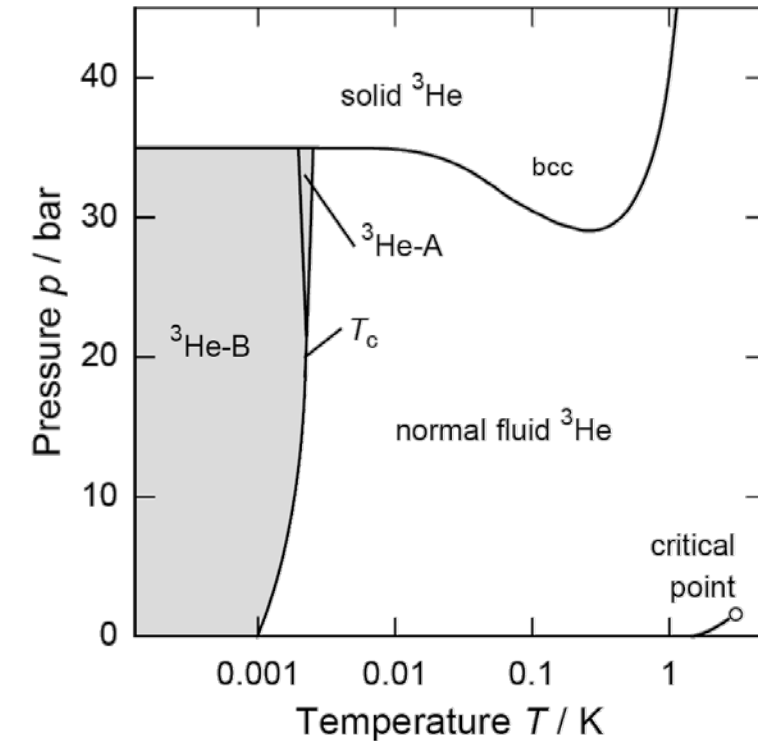
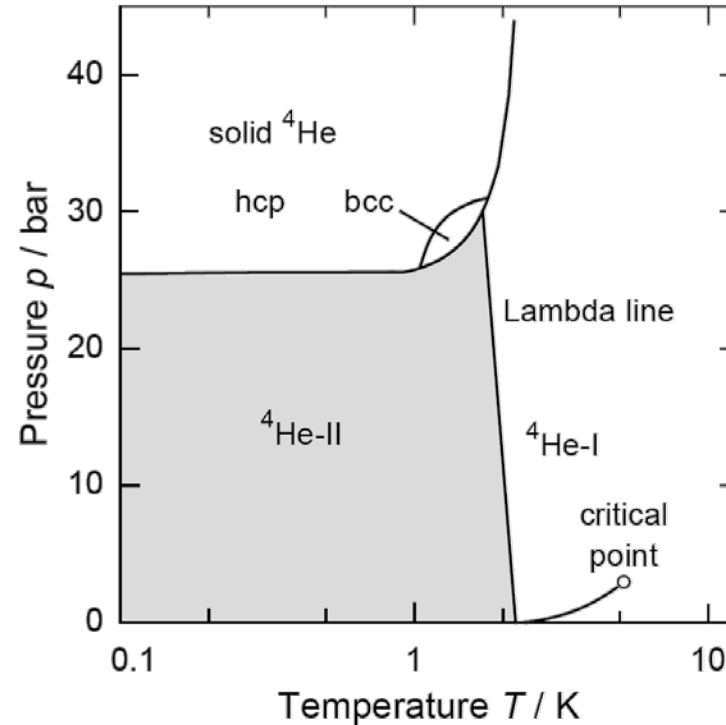
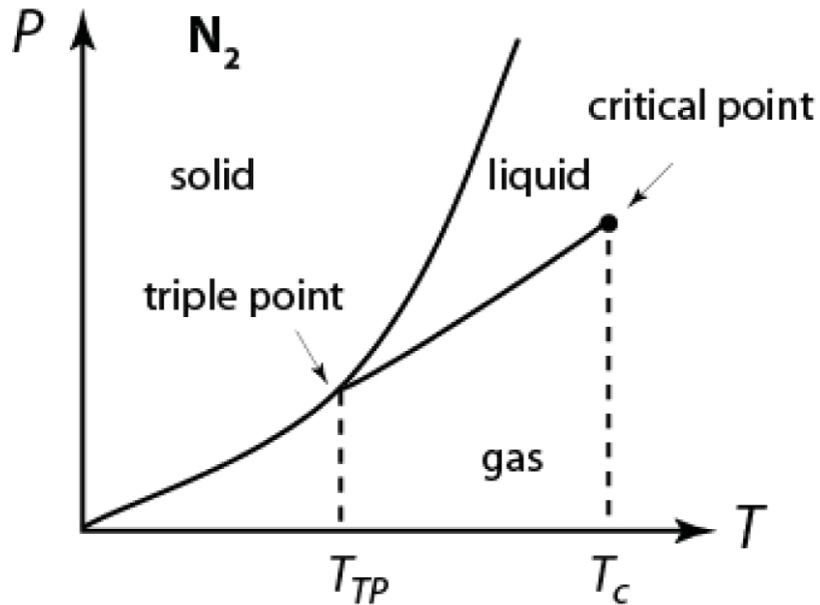
- $T > 0.9 \text{ K}$ ^4He evaporation cryostat (continuous)
- $T > 0.25 \text{ K}$ ^3He evaporation cryostat (continuous)
- $1.0 > T > 0.003 \text{ K}$ ^3He - ^4He dilution refrigerator (continuous)
- $1.0 > T > 0.04 \text{ K}$ Adiabatic electron Demagnetisation Refrigerators (ADR)
- $T < 0.003 \text{ K}$ Adiabatic nuclear demagnetization refrigerators

^3He cryostats, ^3He - ^4He dilution refrigerator & ADR's available commercially:
Bluefors, Cryoconcept, Cryogenic, High Precision Devices, Janis, Leiden Cryogenic,
Oxford Instruments, etc ...

Other methods:

- Pomeranchuk cooling (^3He liquid-solid 1st order transition)
- Electron cooling for small samples

Phase Diagrams: ordinary materials vs ^3He and ^4He



Almost everything behaves like this, apart from ^4He and ^3He

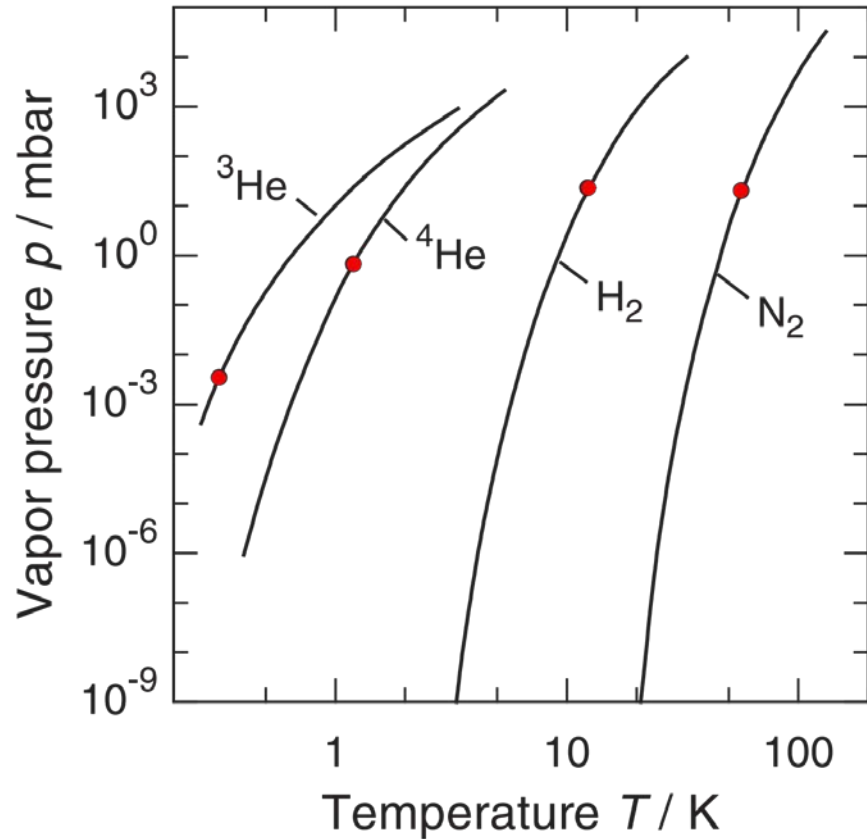
Similarities between ^4He and ^3He :

- Absence of triple point
- Critical point at low temperature
- High pressure to form solid
- Existence of superfluidity

Differences:

- Different statistics (Boson vs Fermi)
- ^3He has magnetic field dependence
- Superfluidity has different origin
- Cost:
 - ^4He £5-£10 per liquid litre
 - ^3He £1.3M per liquid litre!

^4He , ^3He evaporation cryostats



Temperature dependence of ^3He vapour pressure described by Clausius-Clapeyron equation

$$\frac{dP}{dT} = \frac{\Delta S}{\Delta V}$$

Ignoring negligible liquid molar volume, substituting for the (approx constant) latent heat $L = T\Delta S$, find

$$P \propto \exp(-L / RT)$$

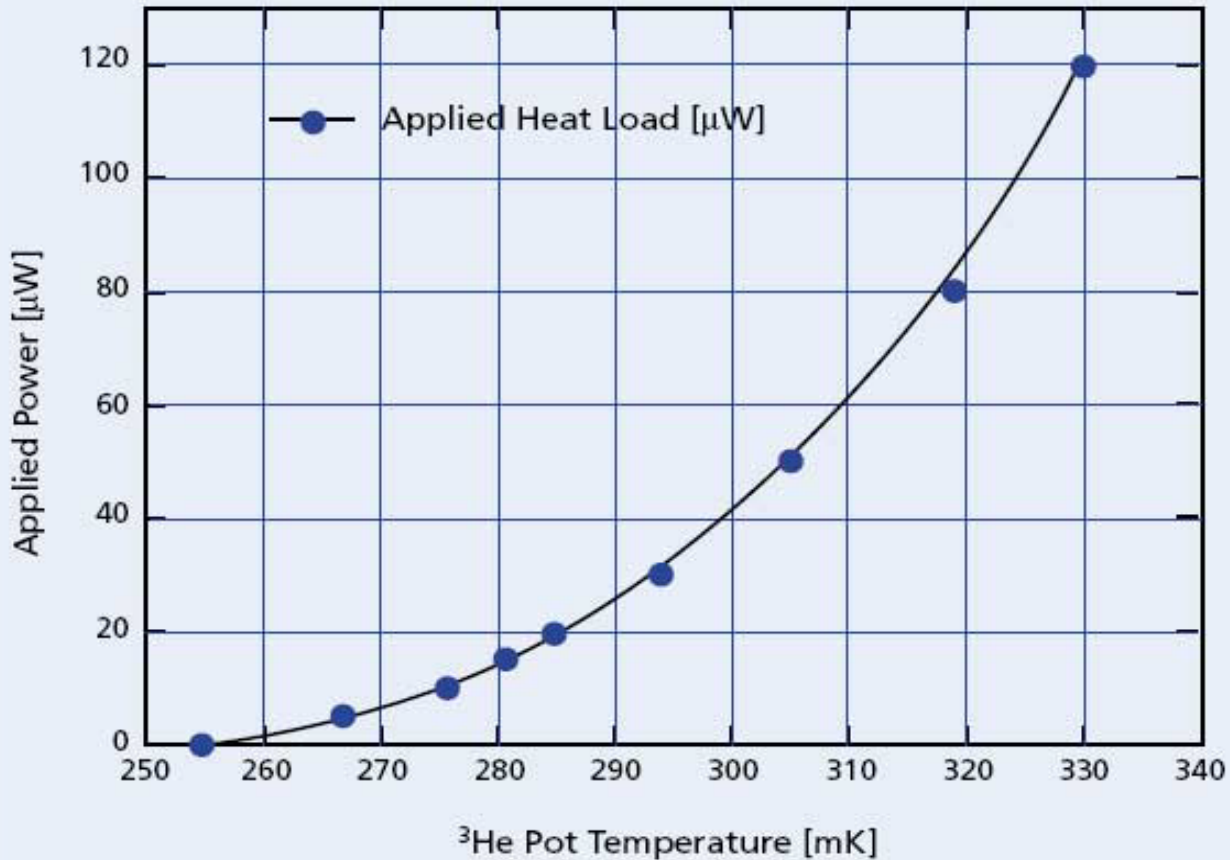
Cooling power \propto mass flow across phase boundary $\propto P$.

\therefore exponentially falling cooling power. $L/R = 2.5\text{K}$ limits T_{base} to 0.2 - 0.3K

How large is change of cooling power when the operating temperature of a ^3He cryostat changes from 0.3K to 0.2K?

How about from 0.2K to 0.1K?

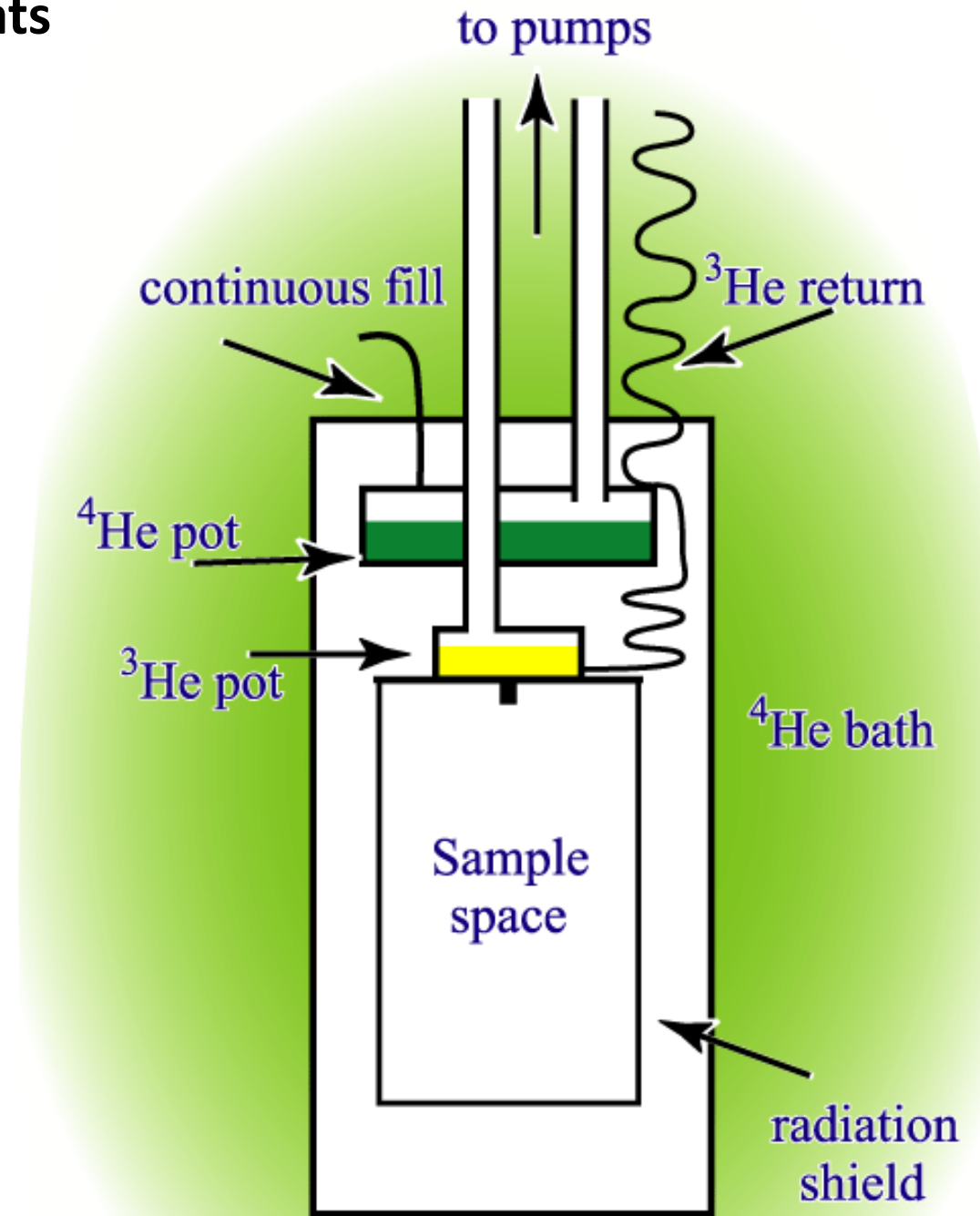
^3He evaporation cryostats



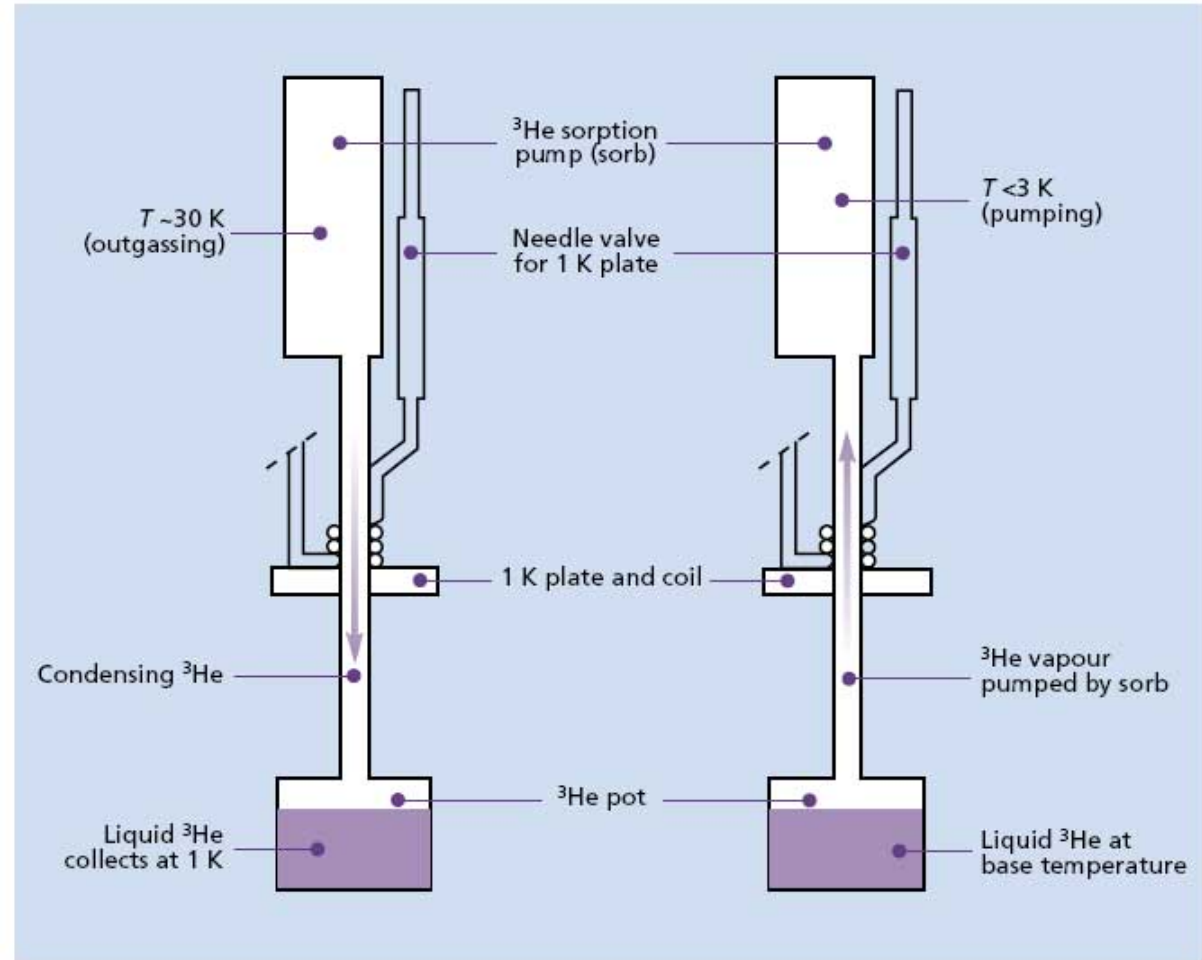
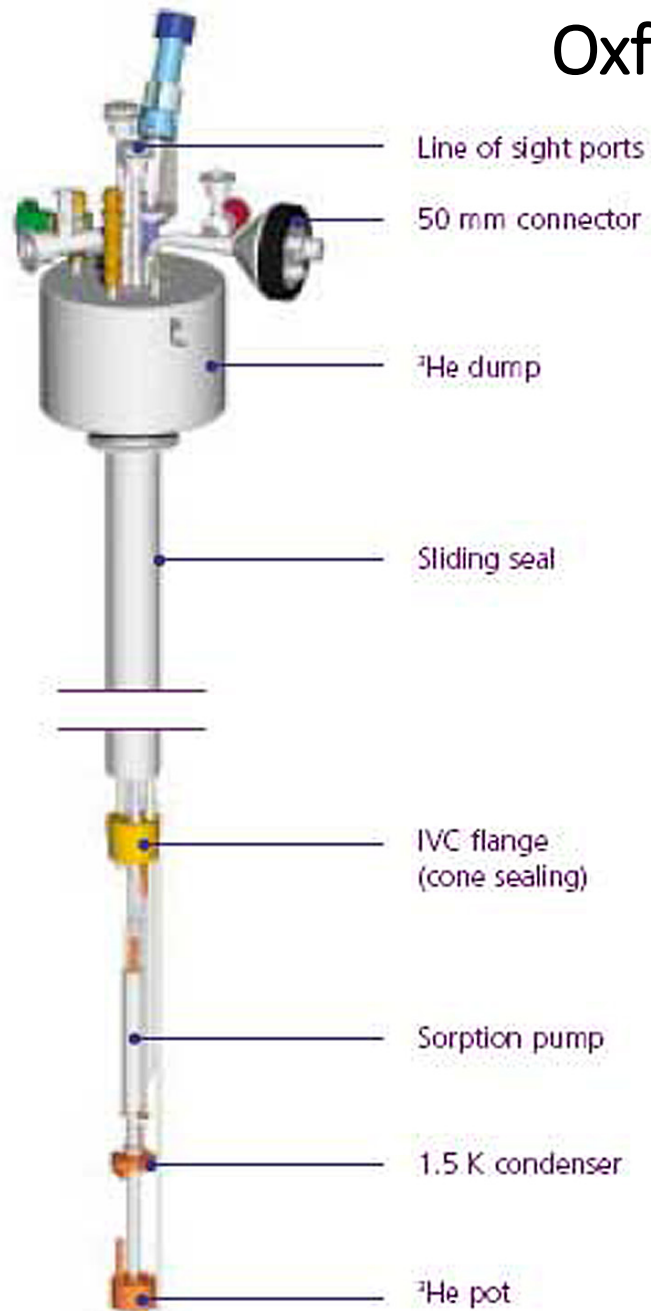
Typical cooling power of the Oxford Instruments HelioxVL

Base temperature: ≤ 245 mK for ≥ 90 hrs with no applied heat load

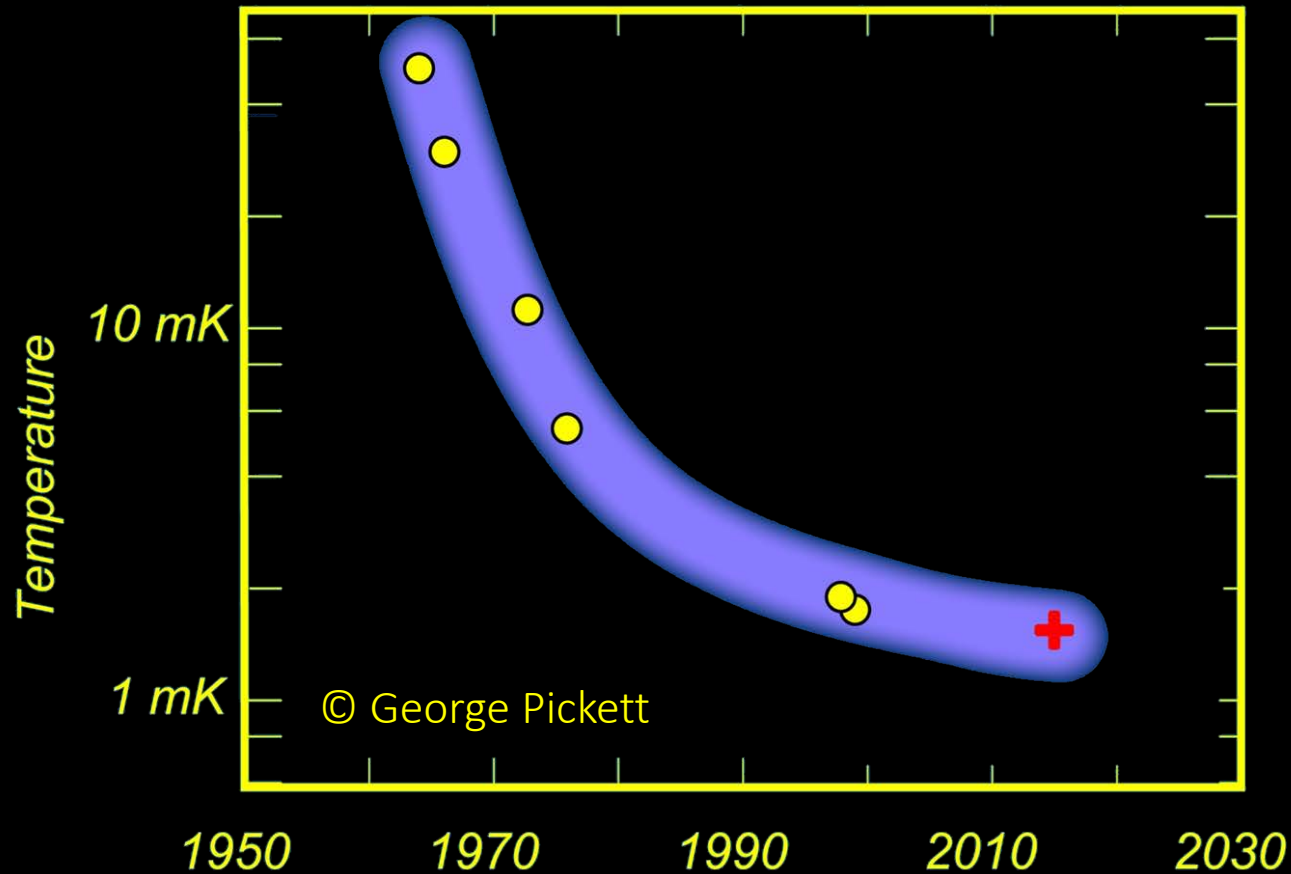
Cooling power: $40 \mu\text{W}$ at ≤ 290 mK, with a hold time > 10 hrs



Oxford Instruments Heliox system

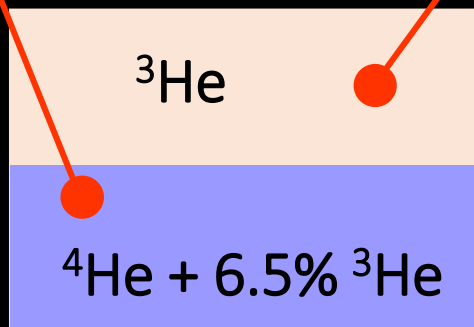
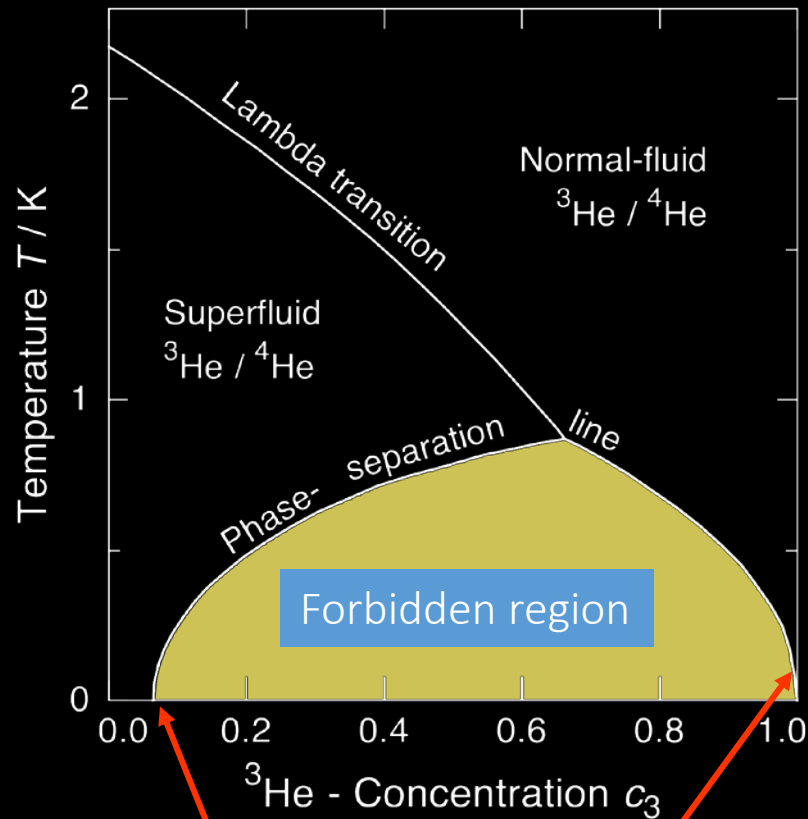


^3He - ^4He dilution refrigerator: History and the Minimal Temperature



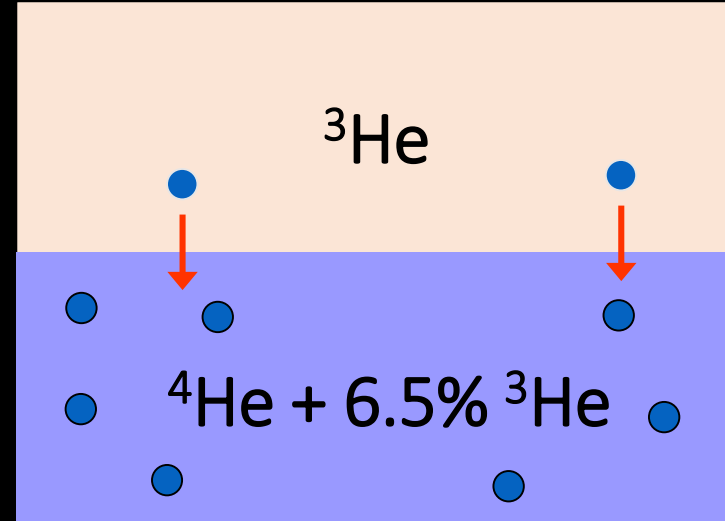
- 1951 Basic idea suggested by Heinz London
- 1962 Detailed Concept worked out by London, Clark, Mendoza
- 1965 First realisation Das, De Bruyn Ouboter, Taconis $T_{\min} = 220$ mK
- 1972 Single shot, V. Edelman, $T \sim 60$ mK
- 1999 Lowest Temperature obtained, J.C. Cousins *et al.*, $T_{\min} = 1.75$ mK

^3He - ^4He mixture:



Principle of cooling is mixing $^3\text{He}/^4\text{He}$

Transition of ^3He into the ^4He rich phase



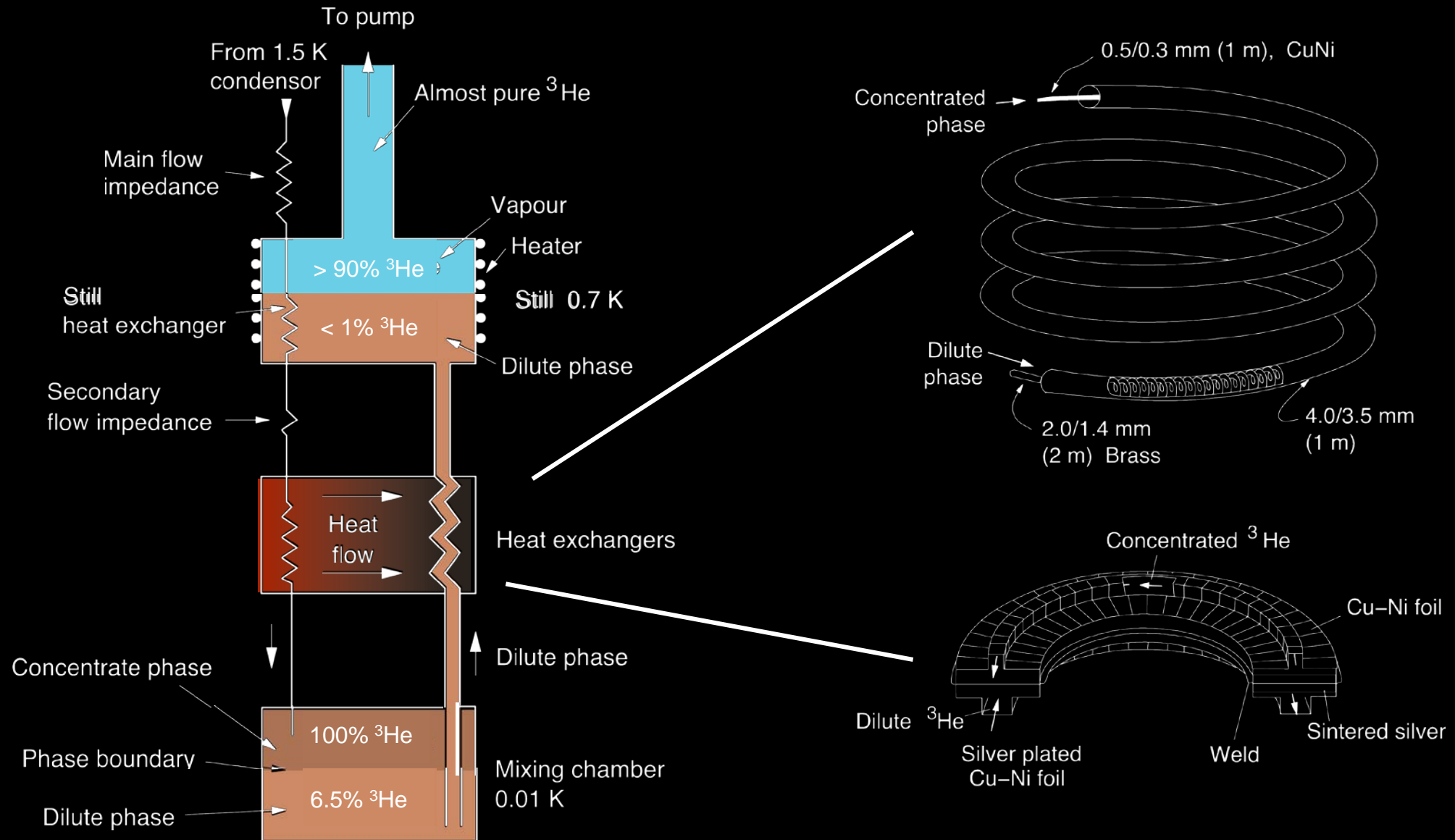
Cooling by evaporation of ^3He into ^4He quasi-vacuum



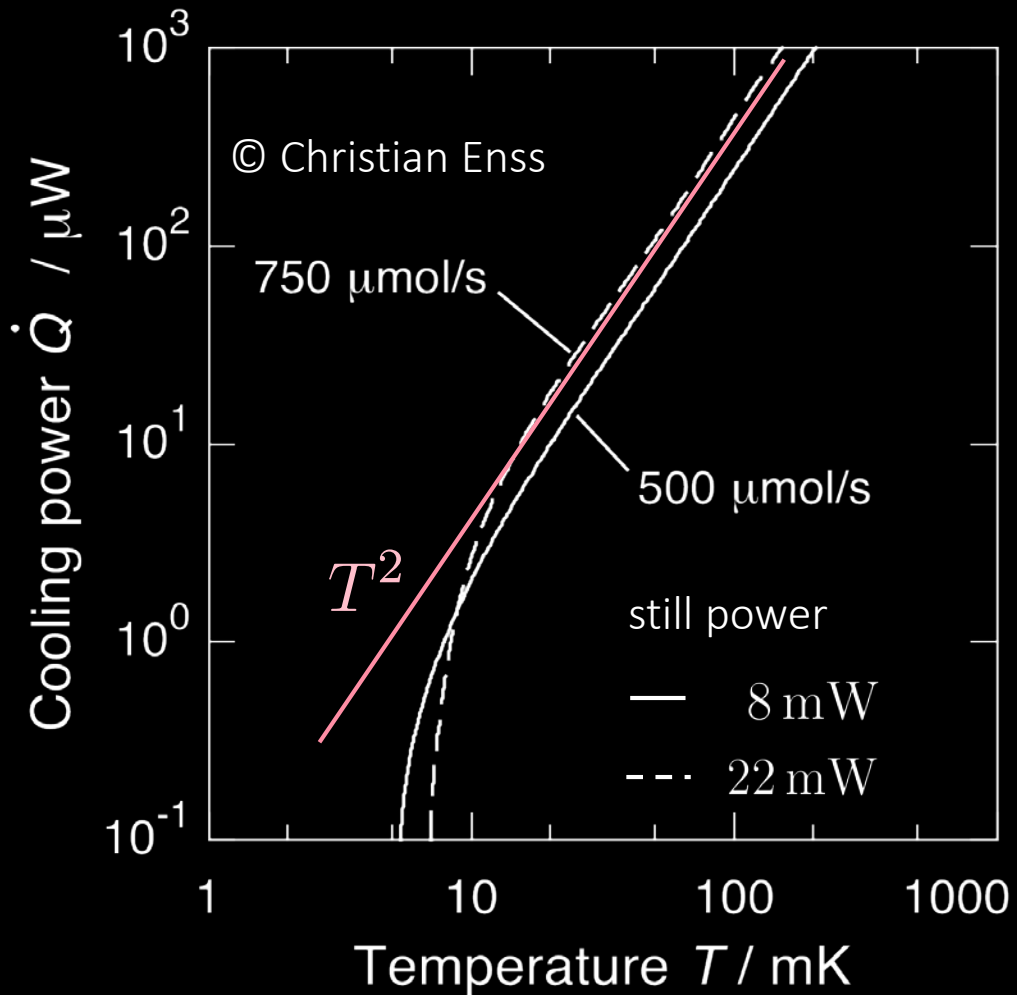
Cooling power \propto ^3He circulation rate \dot{n}_3 can be $\sim 1000 \mu\text{moles s}^{-1}$ with big pumps

$$\dot{Q} \sim 80 \dot{n}_3 (T^2 - T_{base}^2)$$

Realisation of $^3\text{He}/^4\text{He}$ Cooling Cycle



Minimum Temperature and Flow Dependence of Cooling Power



Minimum Temperature

There is no principle limit ... it is determined by the heat leak!

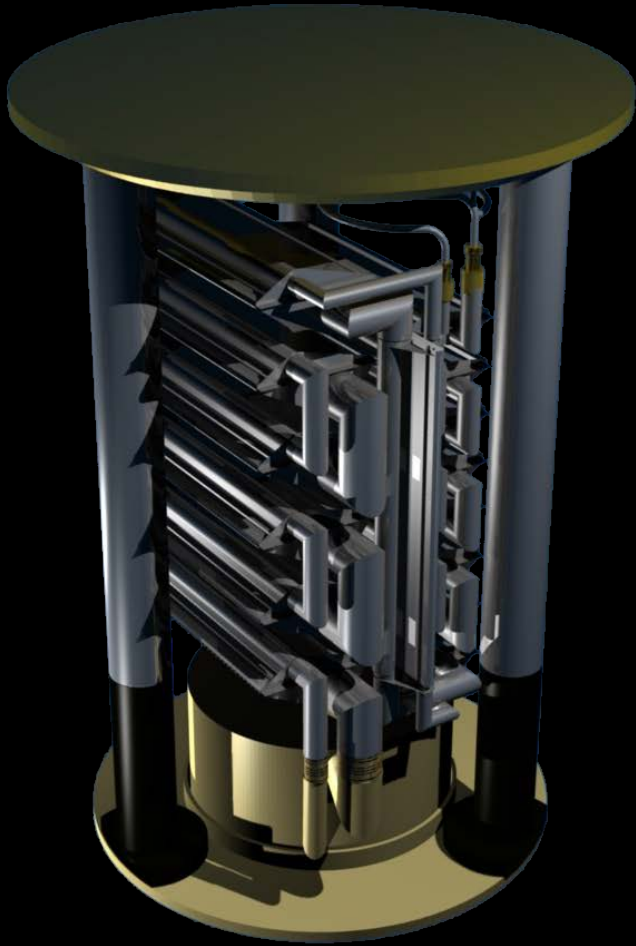
Limiting case of vanishing cooling power: $\dot{Q}_{\text{mc}} = 0$

$$\frac{T_{\text{ex}}}{T_{\text{mc}}} = 2.8$$

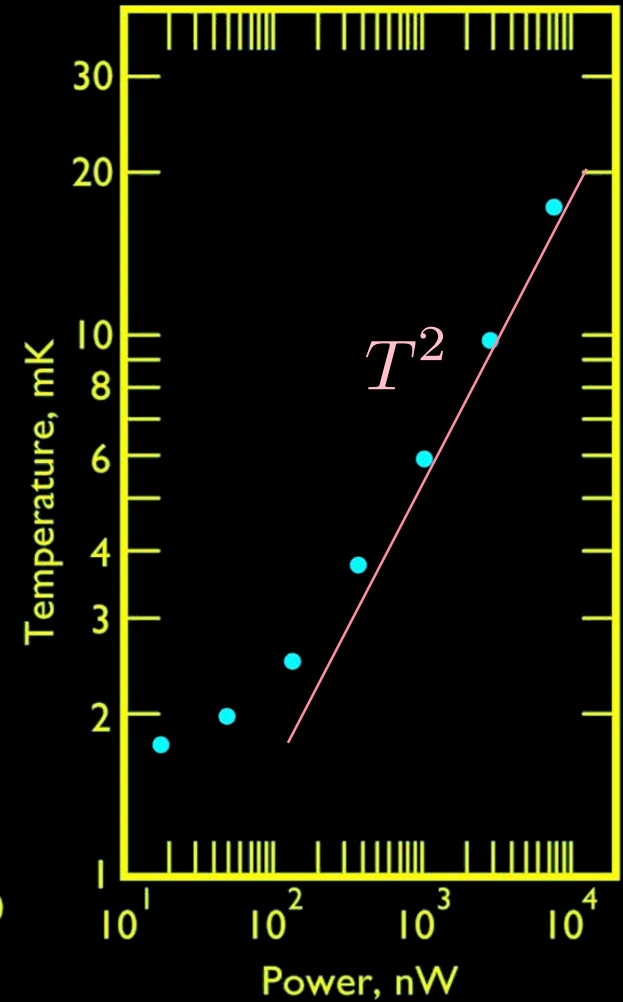
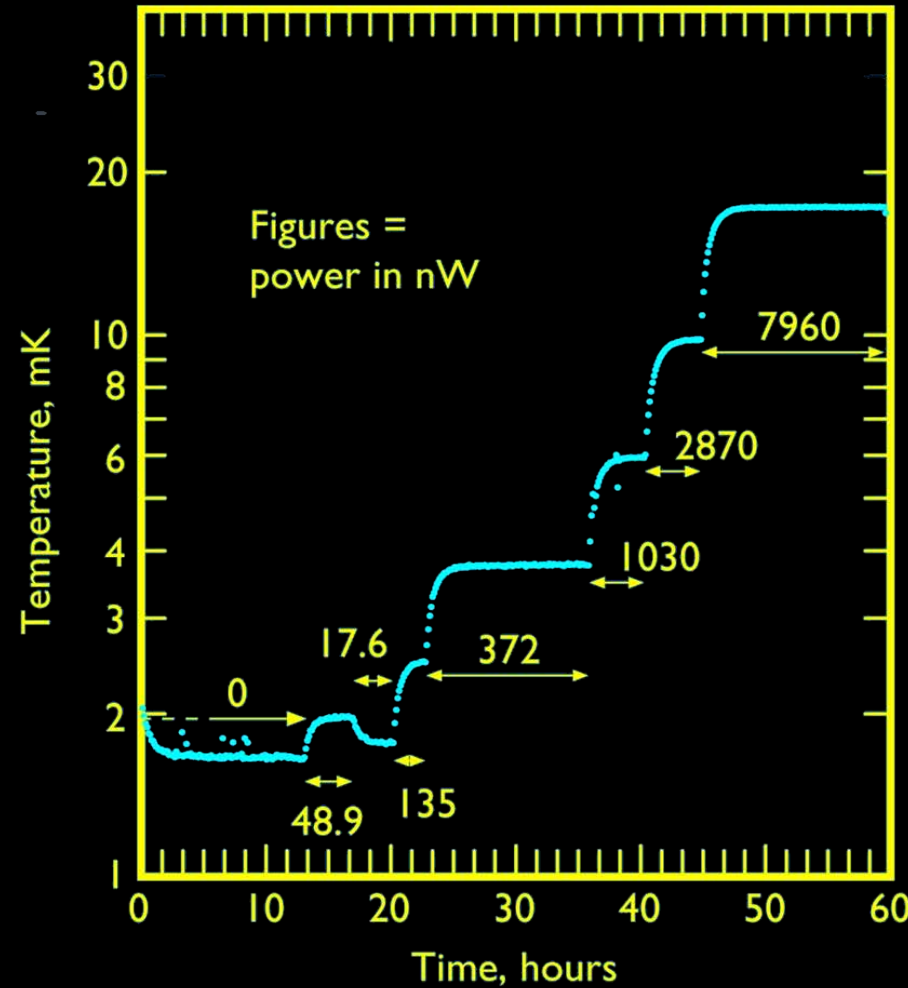
This underlines the importance of the heat exchanger quality

Unavoidable heat leak: viscous friction of ^3He
Another constrain is the thermal conductivity of ^3He

Lancaster Dilution Refrigerator

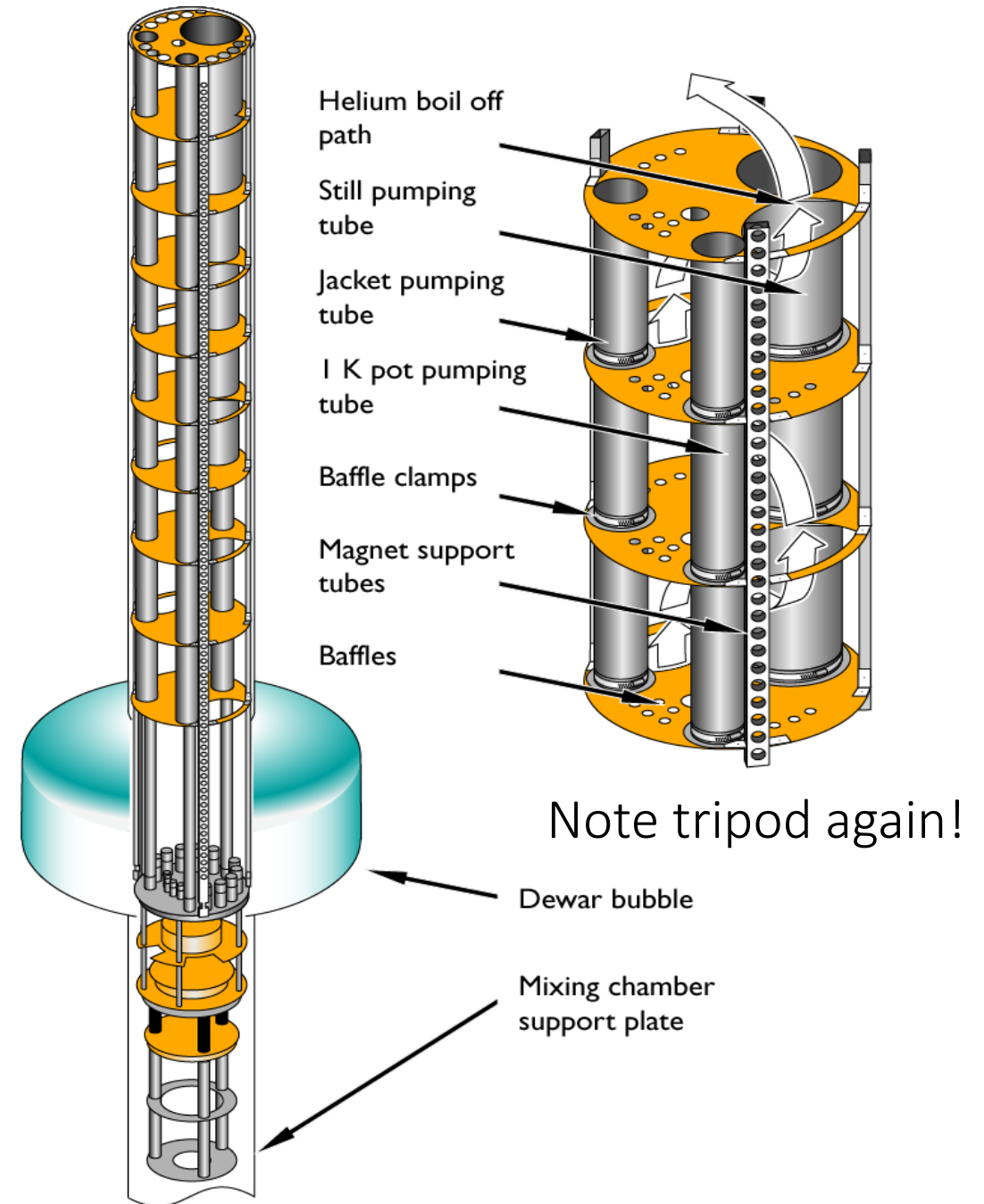
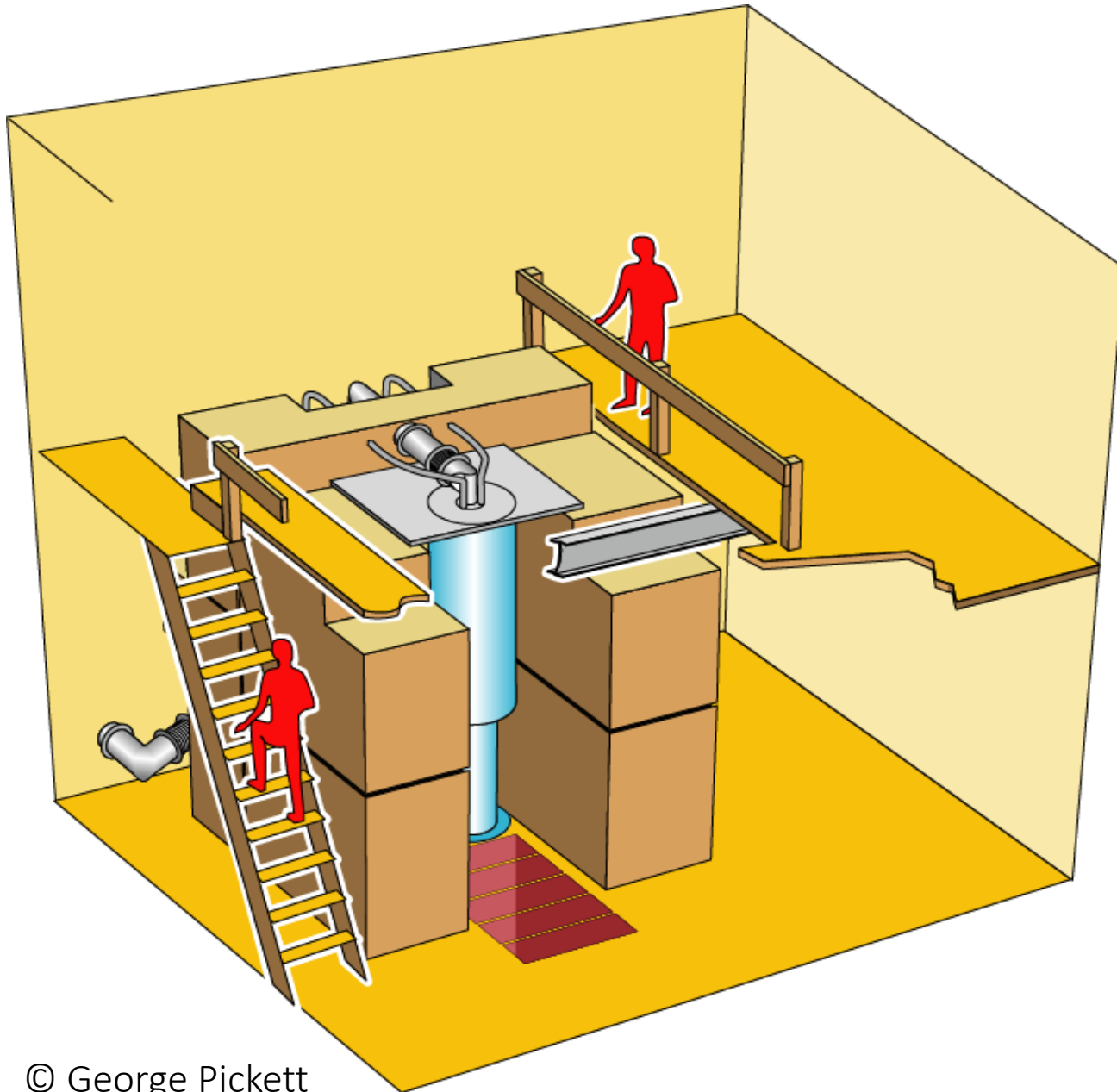


Note tripod!

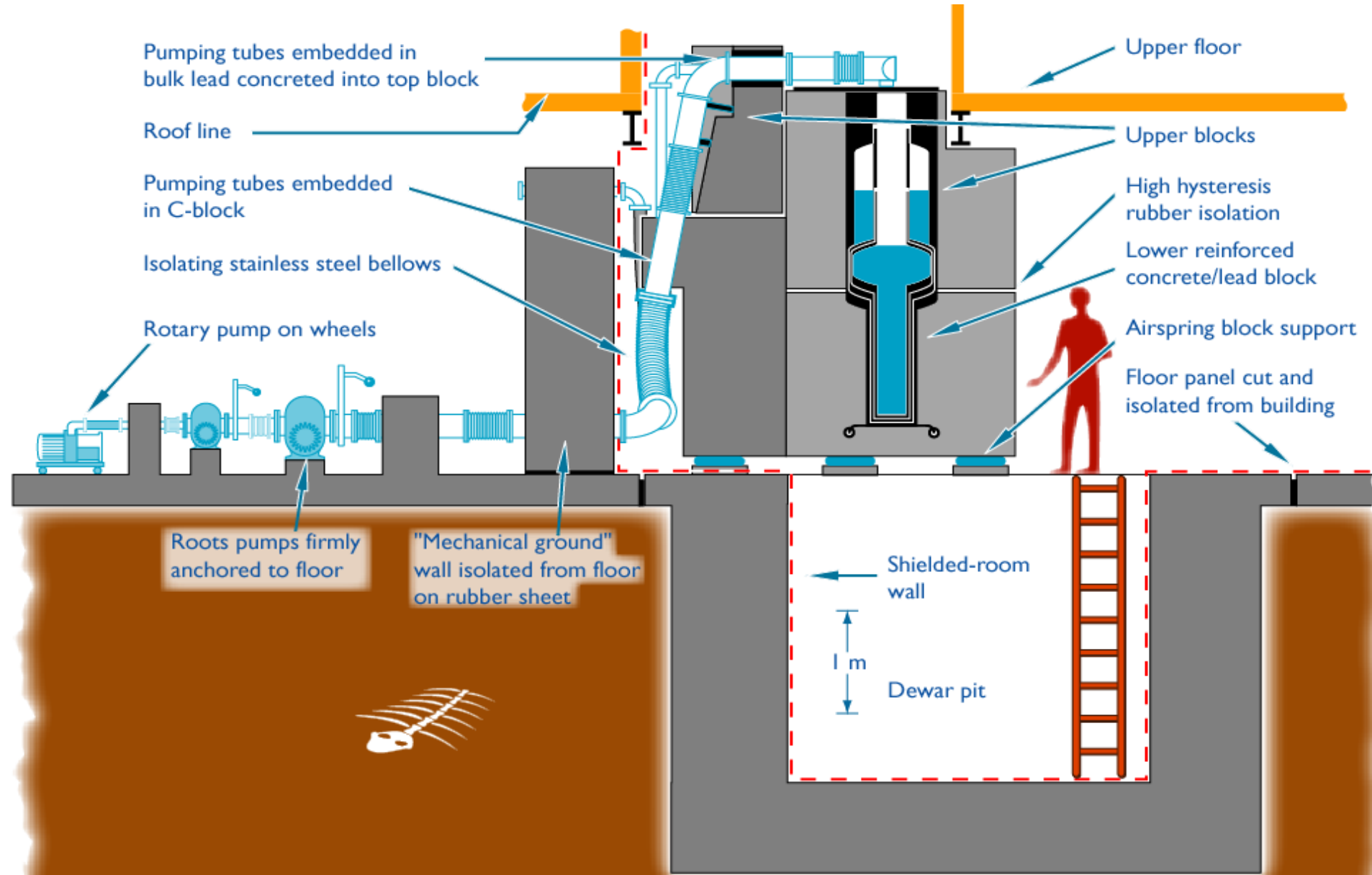


J.C. Cousins et al., J. Low Temp. Phys. **114**, 547 (1999).

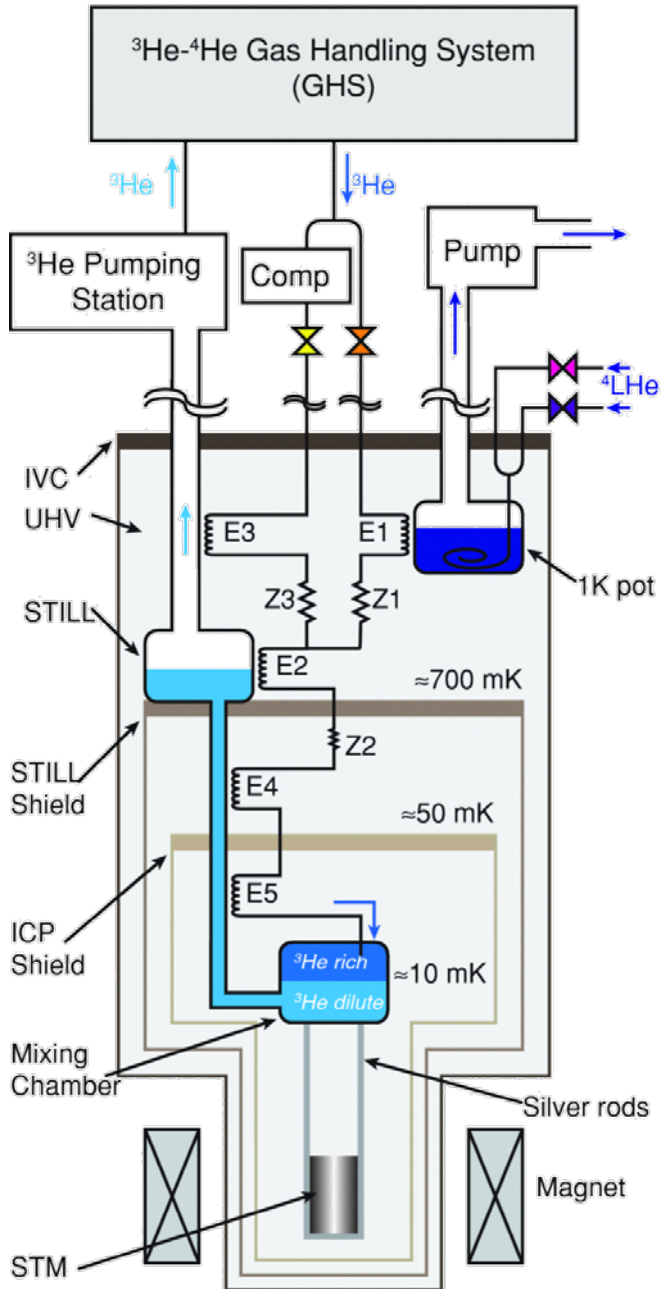
Lancaster Advanced Cryostat



Lancaster Advanced Cryostat



Wet



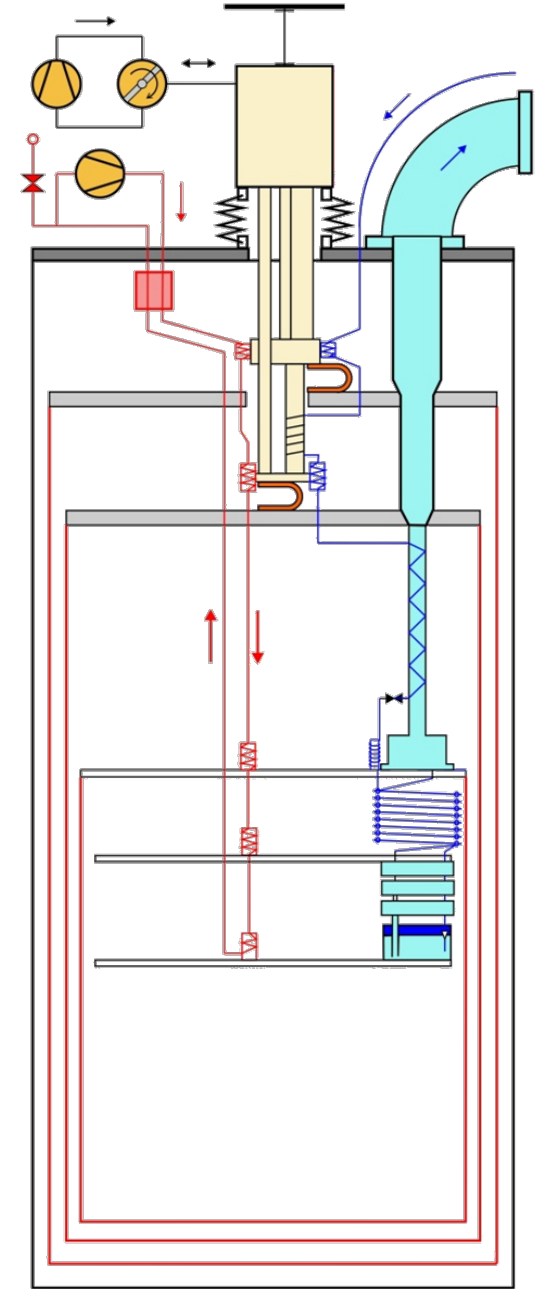
ROTA, Helsinki

Realisations



**Oxford Instruments
Triton 400**

Dry



Wet

vs

Dry

refilling cycle

low vibrations (without 1 K Pot)

4 K power 0.7 Watt x liters He evaporated

vacuum cooled seals

needs liquid helium handling

higher energy efficiency

limited amount of space/flange size

no refilling cycle

vibrations from pulse tube

4 K power 1-2 Watt x number of pulse tubes

no vacuum cooled seals

fast turn around times

less energy efficient

large space and flange size comes easy

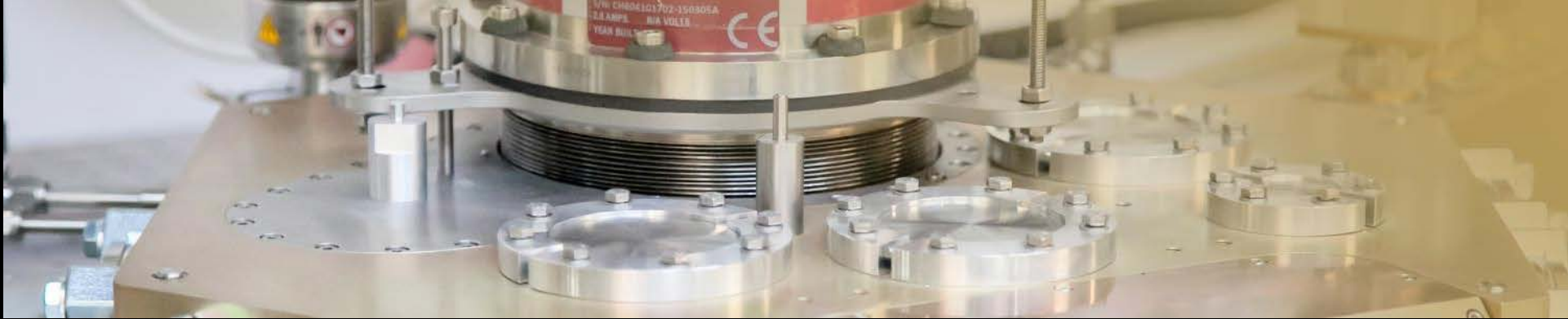
needs compressor

needs precool system



Things are **strongly** moving towards **dry** systems

Ultra Low Vibrational Dry-Cryostat



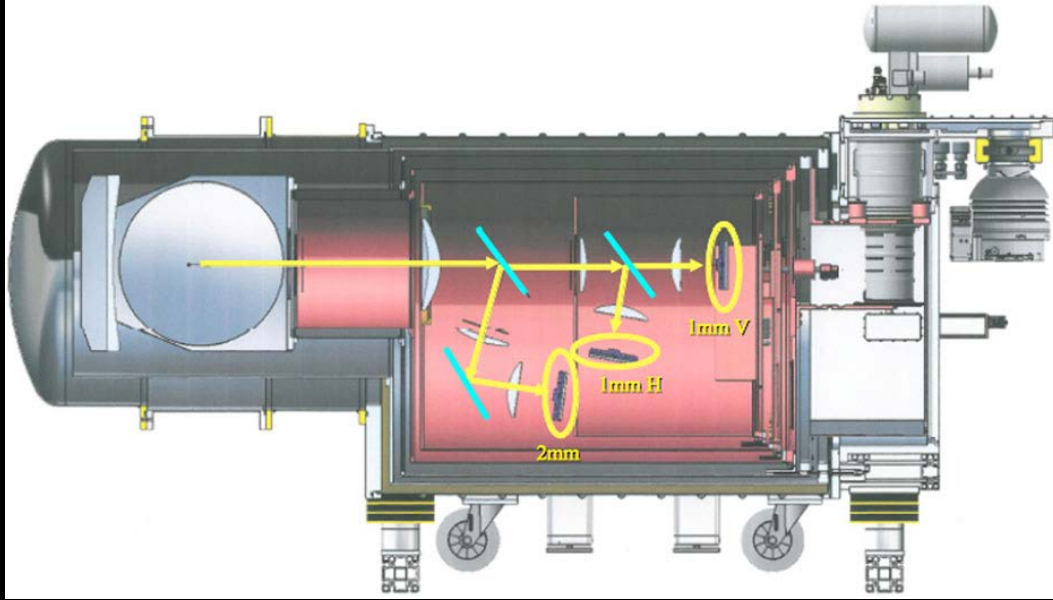
below 40 nm in the 5 Hz – 1000 Hz region

no mechanical contact of pulse tube head and dilution refrigerator

cooling by exchange gas



NIKA-2 (New Iram Kid Array) Cryostat



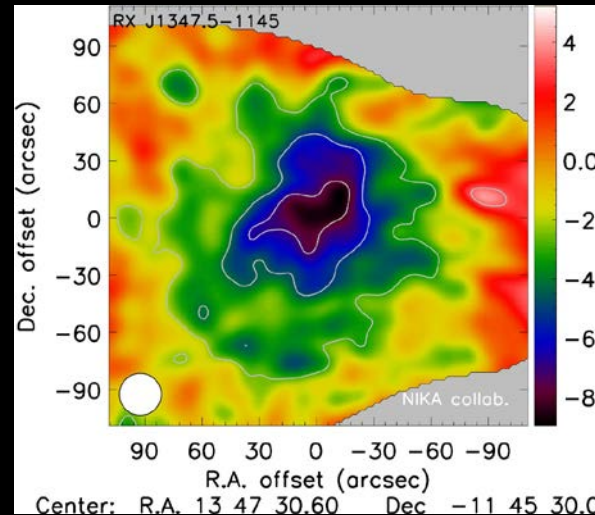
Two pulse tubes with 2 W cooling power

Dilution unit $\sim 100\text{mK}$

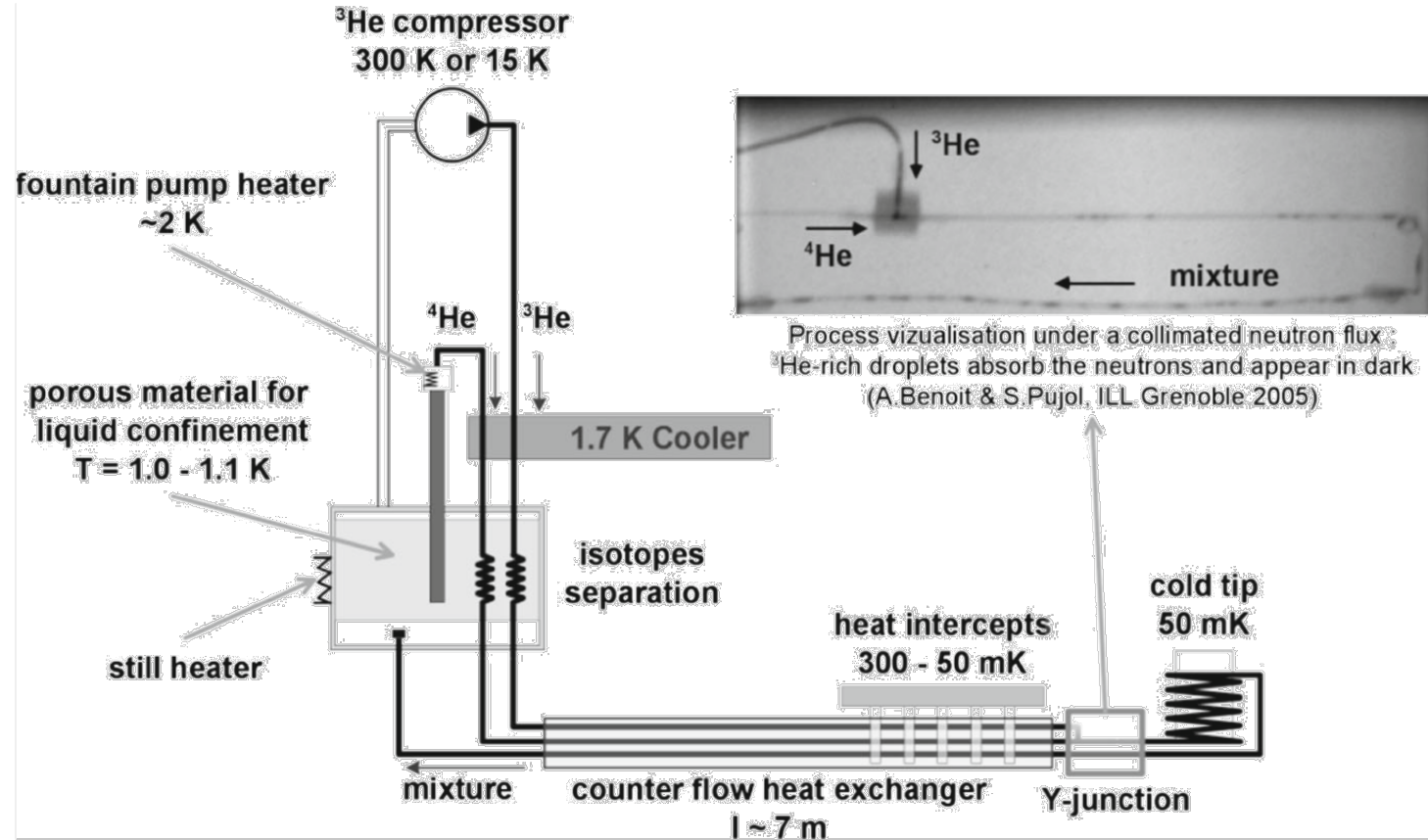
Compact integration for telescope

1000 2mm wavelength detectors, 2×2000 1.2 mm

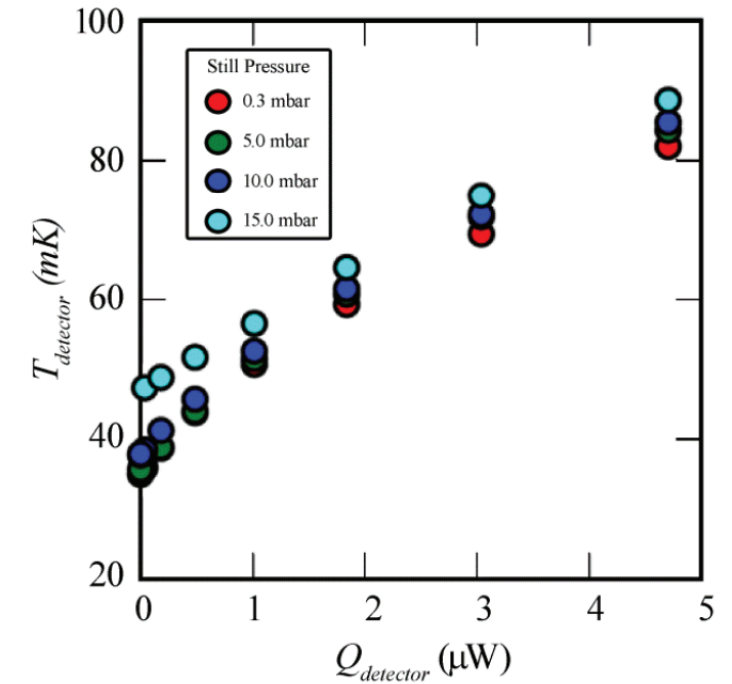
Cluster: RX J1347.5-1145



Closed-Cycle Dilution Refrigerator for Space



1 μ W at 51 mK was demonstrated in 2012



Adiabatic Demagnetisation Refrigerators

© Christian Enss
CBT, Berlin

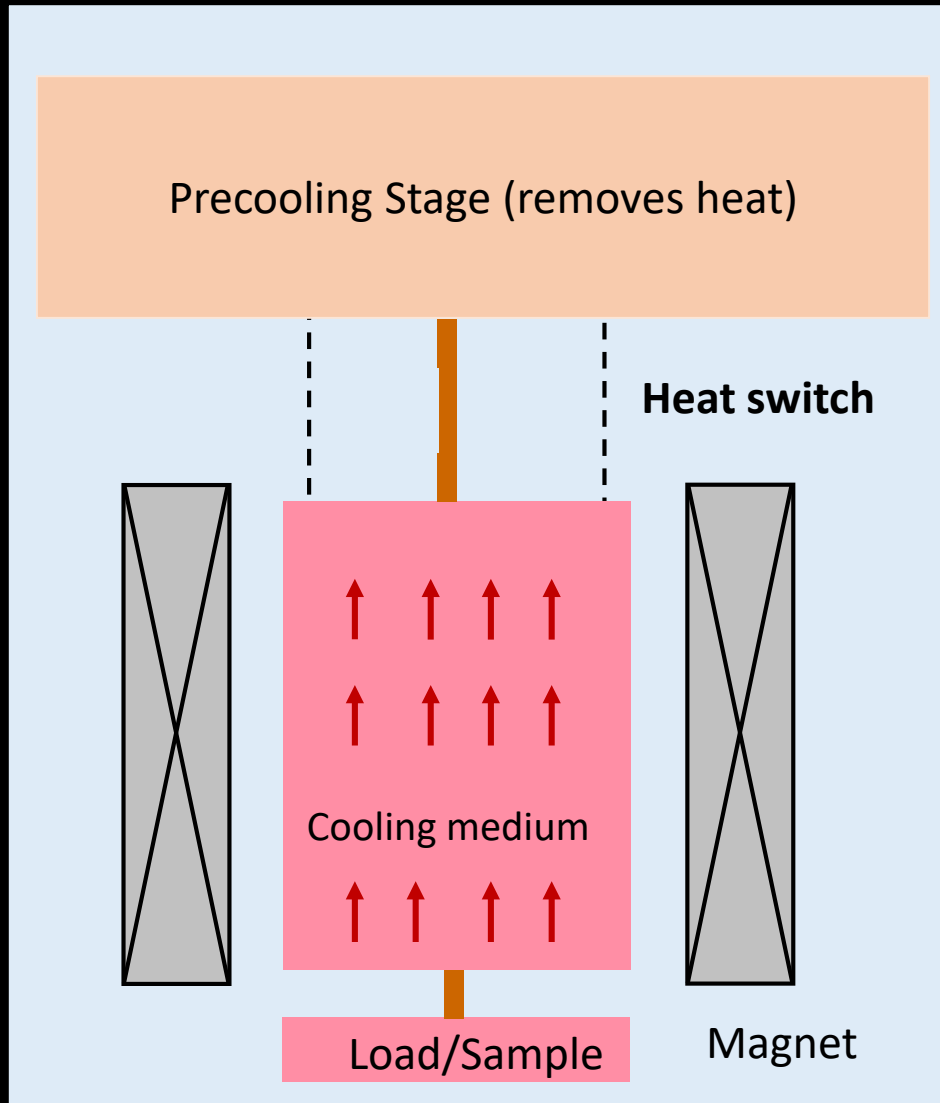


© Dan McCammon, Wisconsin University



© George Pickett

Adiabatic Demagnetisation: principle of operation



1. Precooling

Isothermal Magnetisation

$$\Delta Q_{\text{mag}} = -T_i [S(B_i, T_i) - S(0, T_i)]$$

2. Thermal Isolation

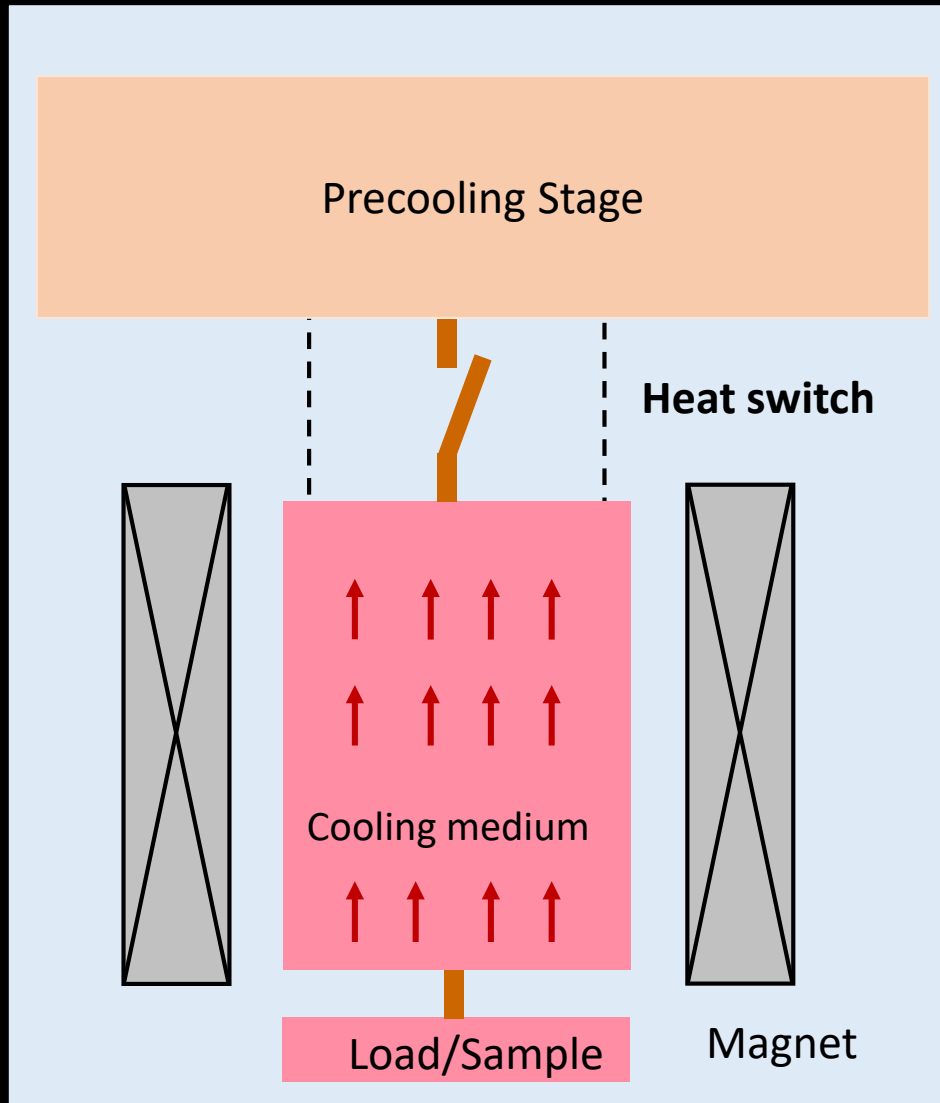
Heat switch **closed**

3. Adiabatic Demagnetisation

$$S = S\left(\frac{B}{T}\right) = \text{const.}$$

4. Warming up

Adiabatic Demagnetisation: principle of operation



1. Precooling

Isothermal Magnetisation

$$\Delta Q_{\text{mag}} = -T_i [S(B_i, T_i) - S(0, T_i)]$$

2. Thermal Isolation

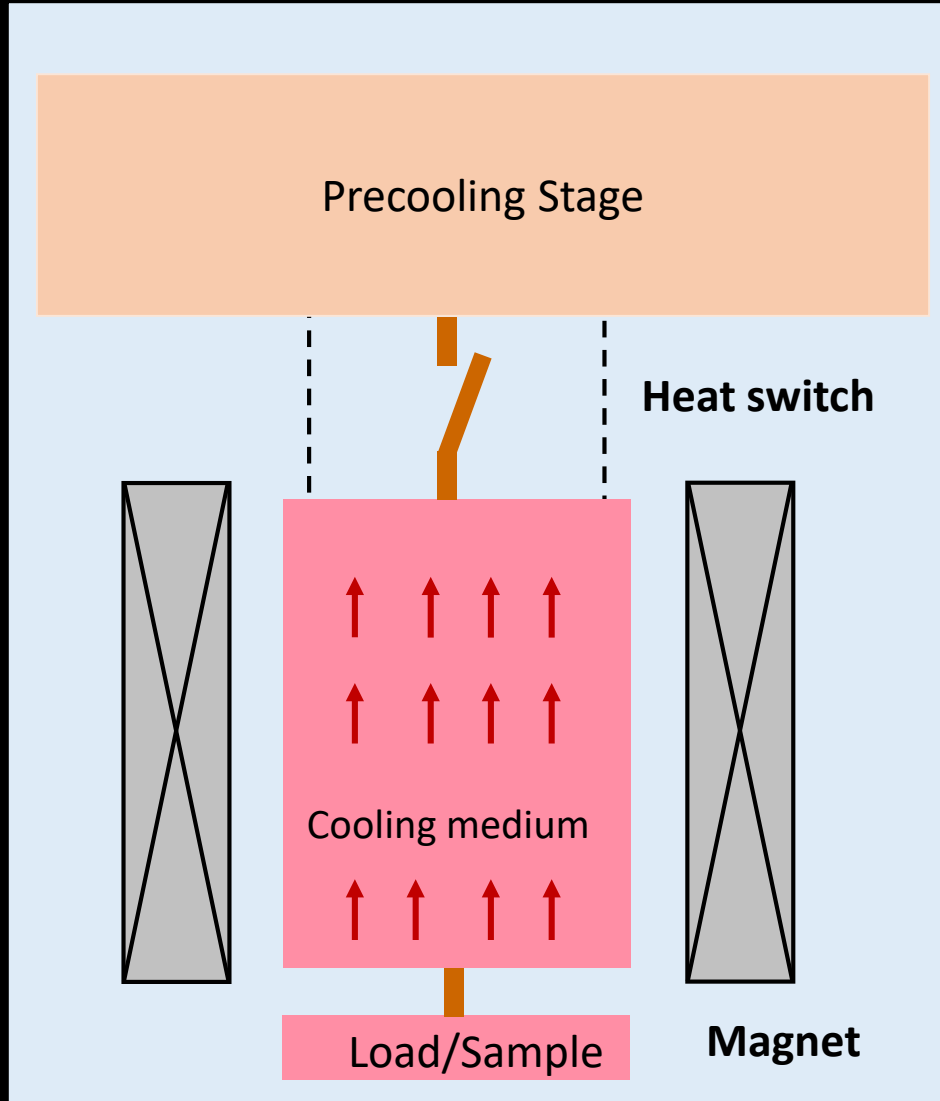
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Adiabatic Demagnetisation: principle of operation



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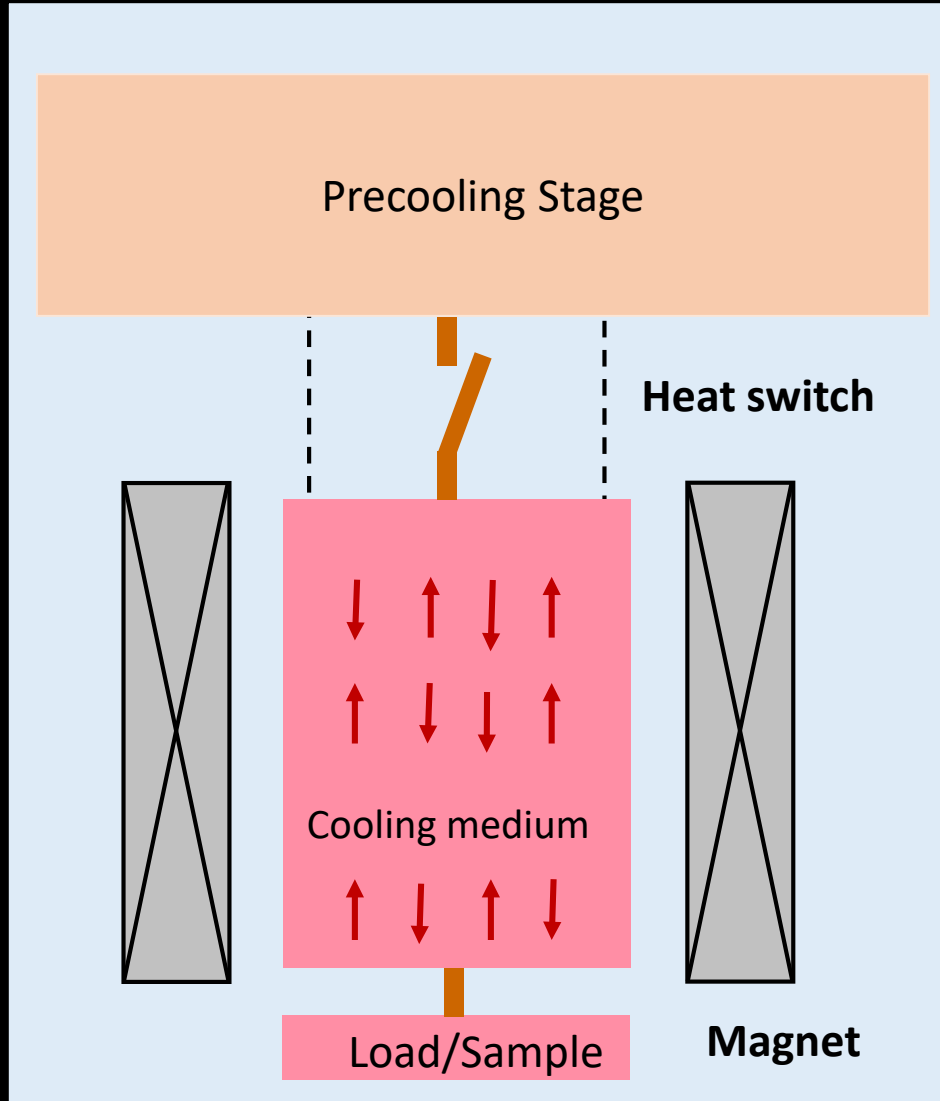
Heat switch **opened**

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Adiabatic Demagnetisation: principle of operation



1. Precooling

Isothermal Magnetisation

$$\Delta Q_{\text{mag}} = -T_i [S(B_i, T_i) - S(0, T_i)]$$

2. Thermal Isolation

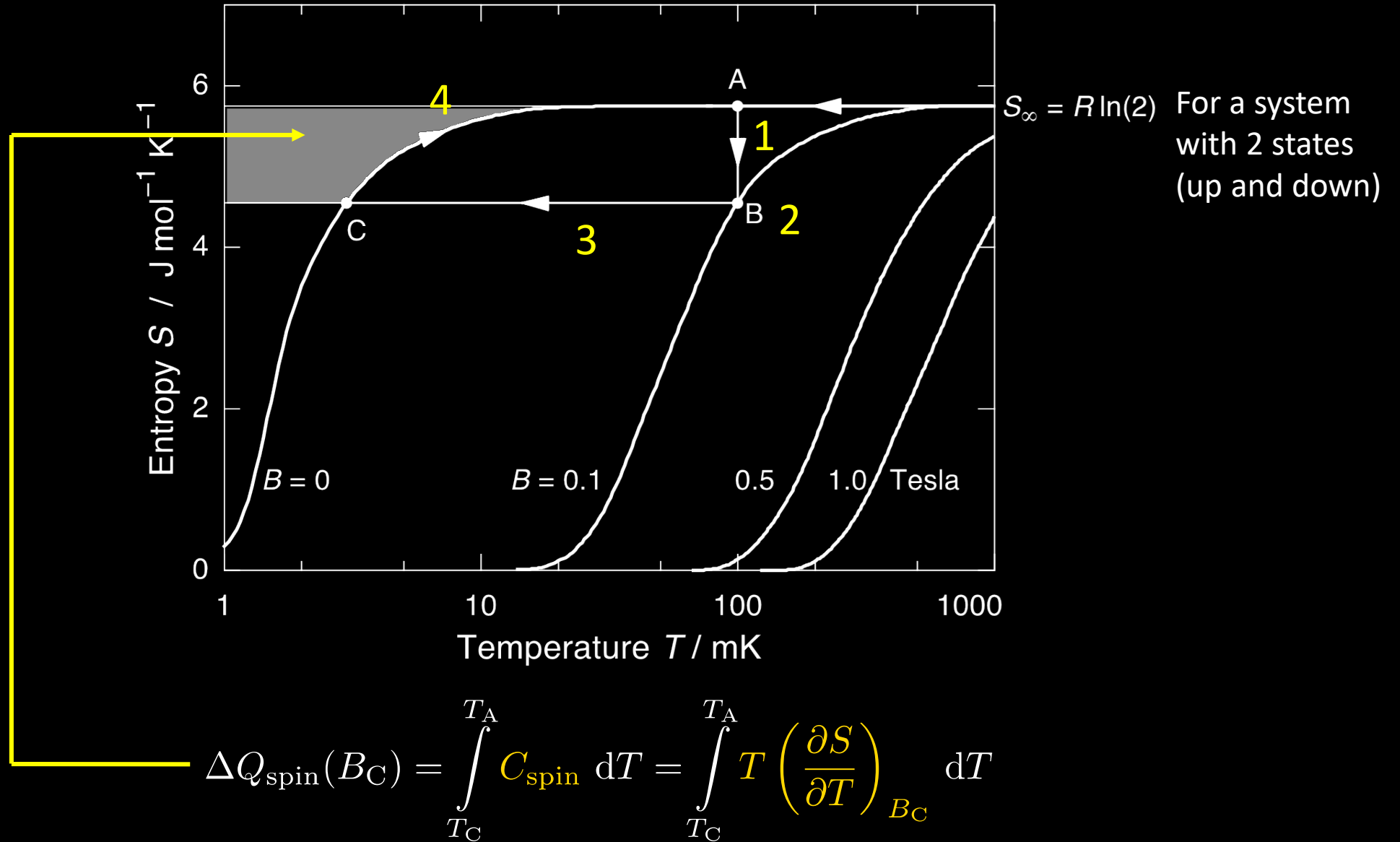
Heat switch **opened**

3. Adiabatic Demagnetisation (reduce B and hence T)

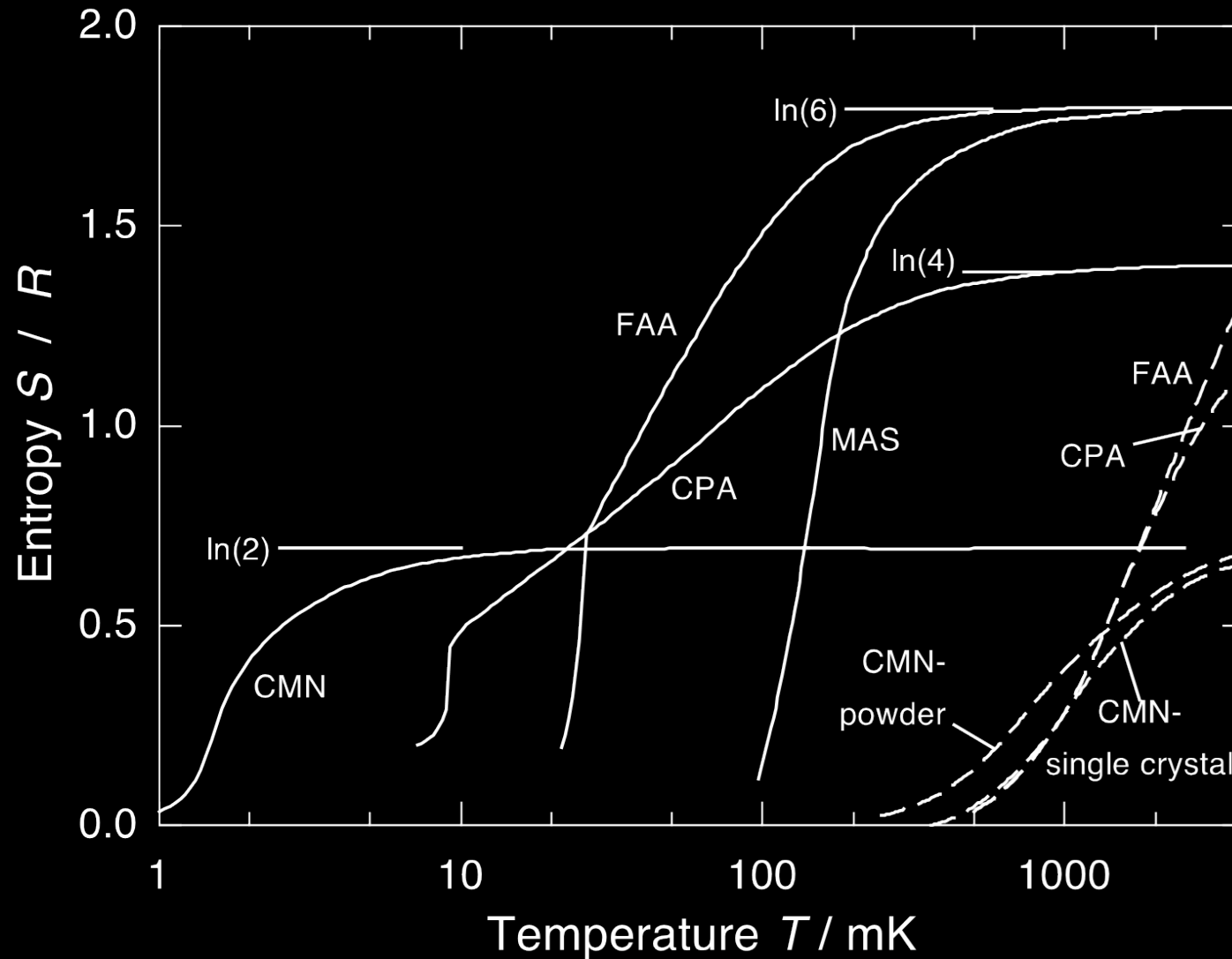
$$S = S\left(\frac{B}{T}\right) = \text{const.}$$

4. Warming up

Entropy and Cooling Capacity



Entropy of Paramagnetic Salts



FAA: Ferric ammonium alum
MAS: Manganous ammonium sulphate
CPA: Chrome potassium alum
CMN: Cerium magnesium nitrate

Problem: Thermal Conductivity of Paramagnetic Salts



© Dan McCammon, Wisconsin University

Salt pill for space application

15.000 gold wires

Salt pill grown around the wires

Possible solution: Super-heavy electron material as metallic refrigerant for adiabatic demagnetization cooling.

Science Advances 09 Sep 2016: Vol. 2, no. 9, e1600835
DOI: 10.1126/sciadv.1600835

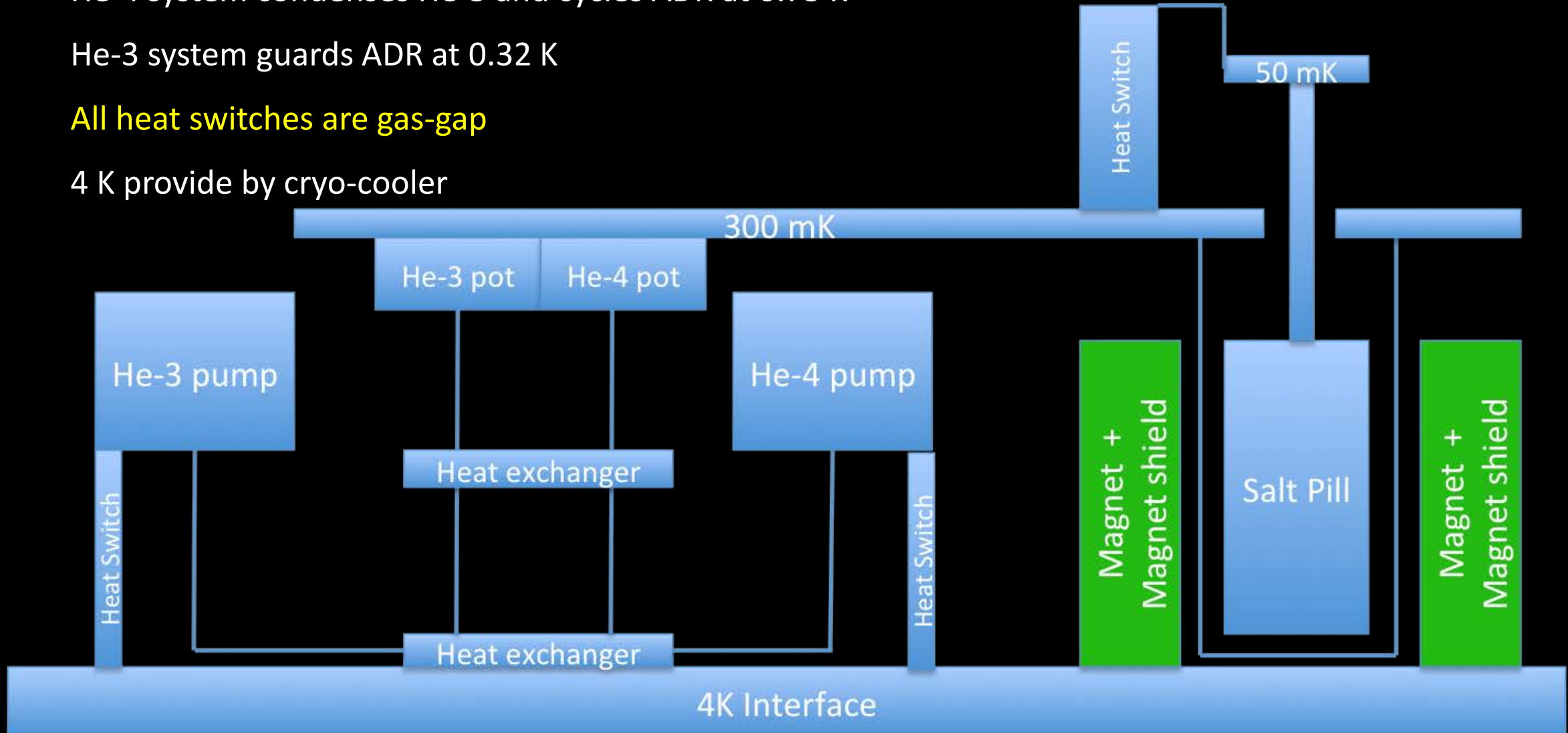
Fully Automatic ADR for Space Missions

He-4 system condenses He-3 and cycles ADR at 0.75 K

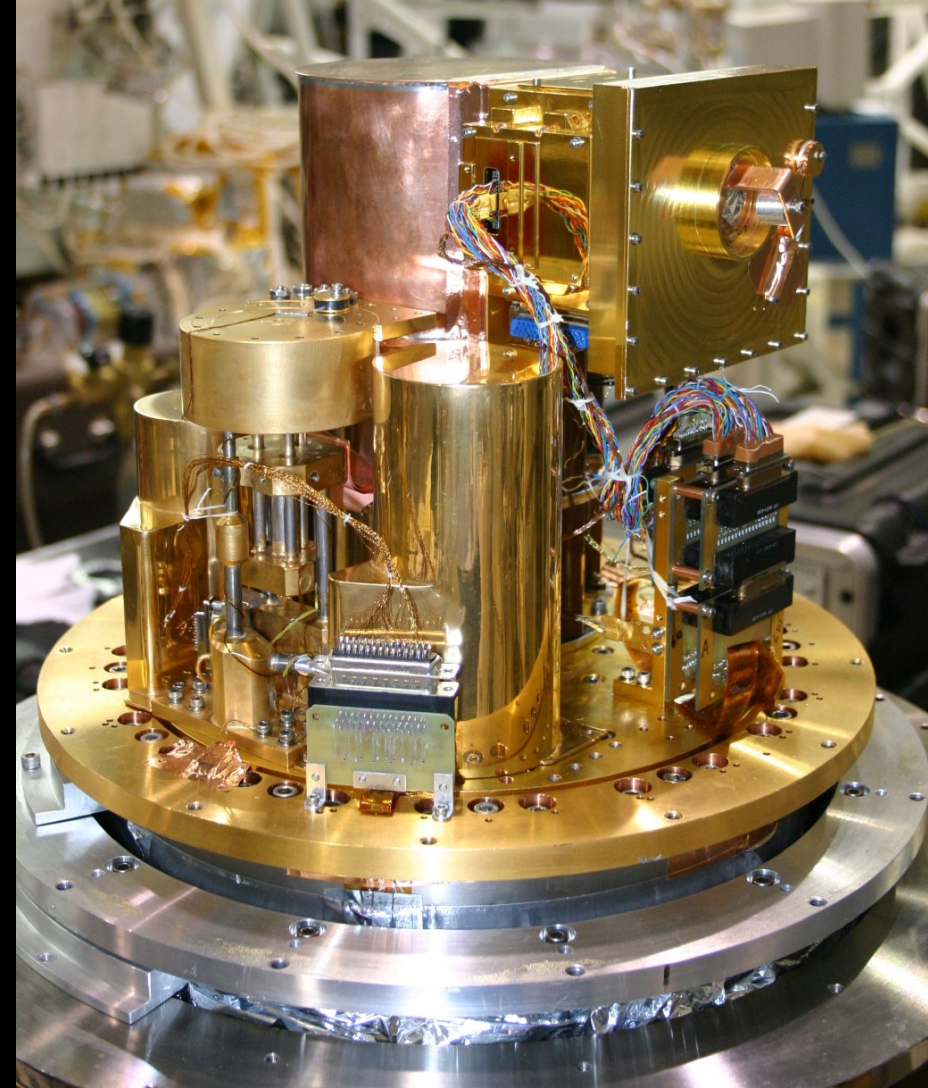
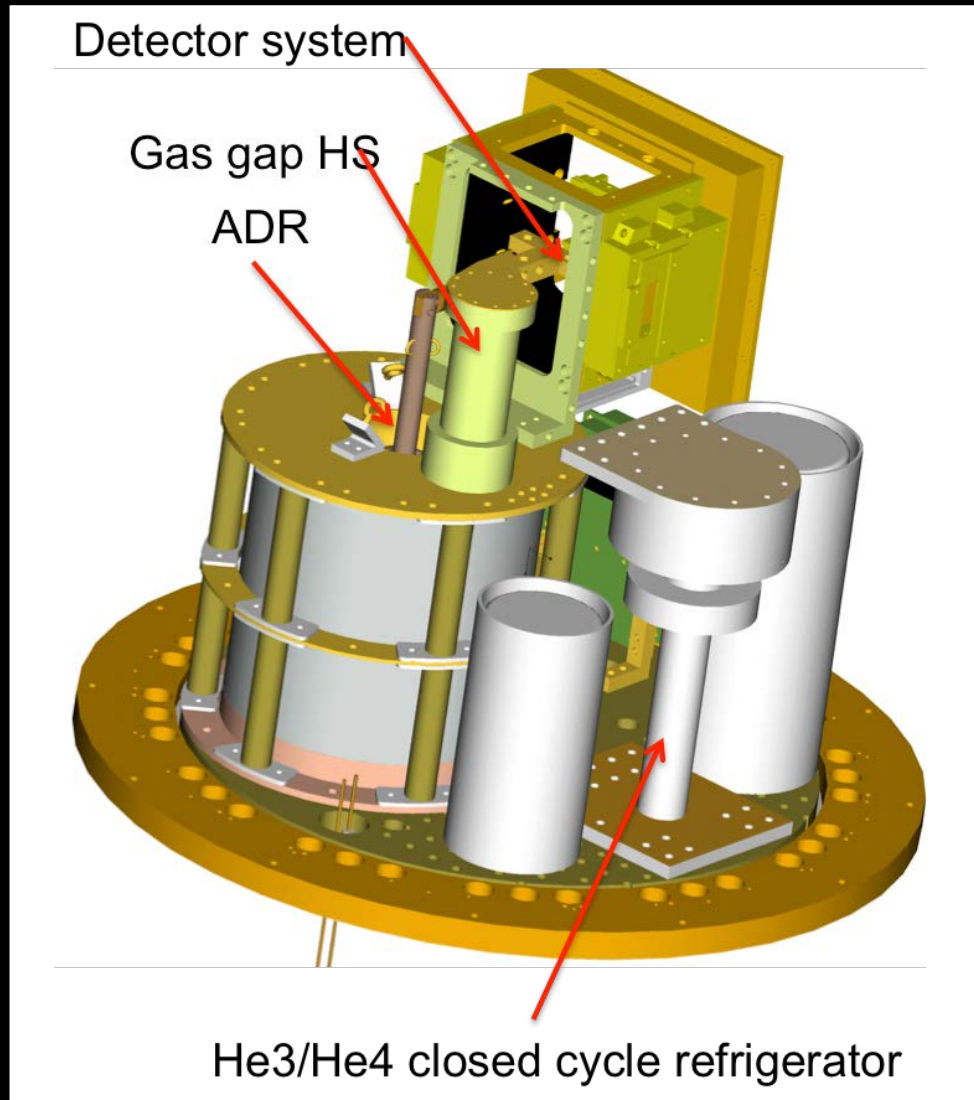
He-3 system guards ADR at 0.32 K

All heat switches are gas-gap

4 K provide by cryo-cooler



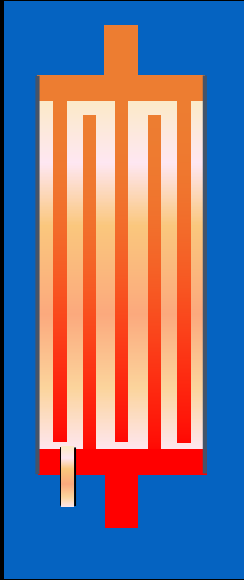
Fully Automatic ADR for Space Missions (50 mK)



For latest developments check NASA Technical Reports Server <https://ntrs.nasa.gov/search.jsp>
NASA GSFC

Heat Switches

Gas gap heat switches



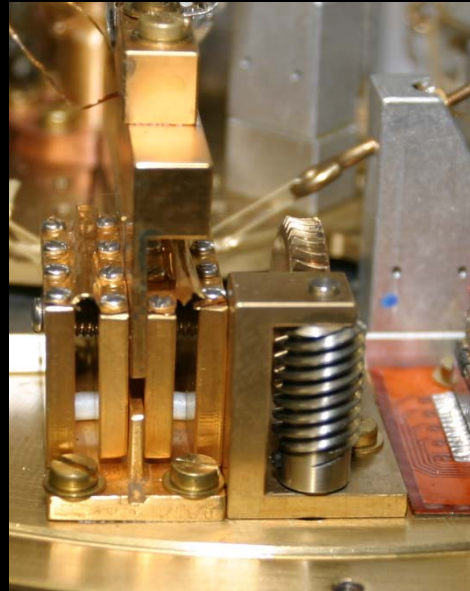
exchange gas → pumping

^4He superfluid layer
→ creep

H_2 ortho-para
conversion

^3He no exothermic reaction
no creep
high vapour pressure

Mechanical heat switch

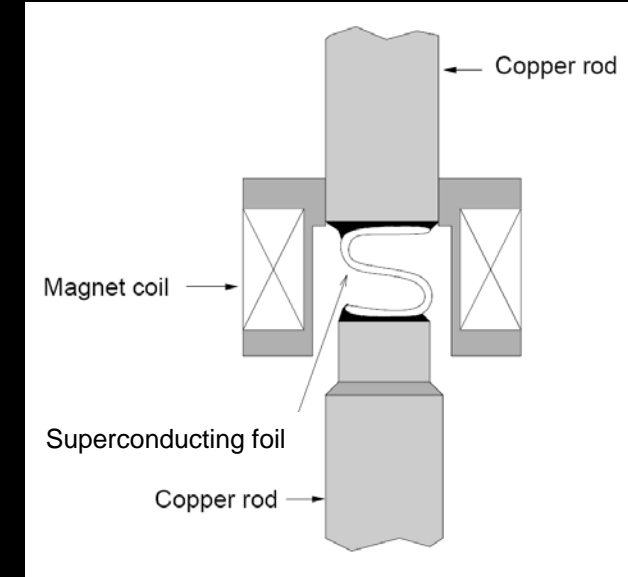


large force needed
typically 100 N

closed few mW/K
... 1 W/K @ 15K

heating on opening

Superconducting heat switch



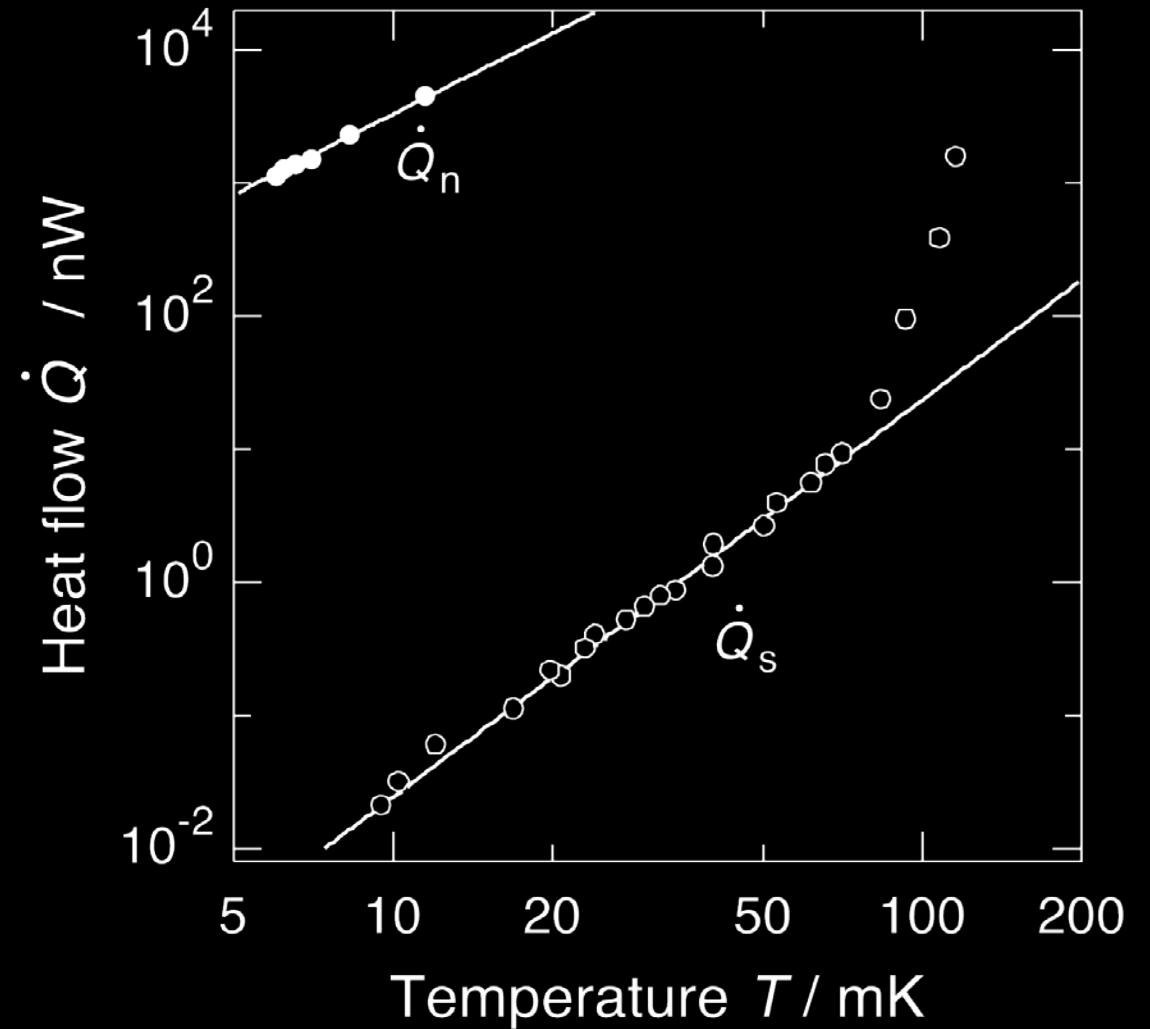
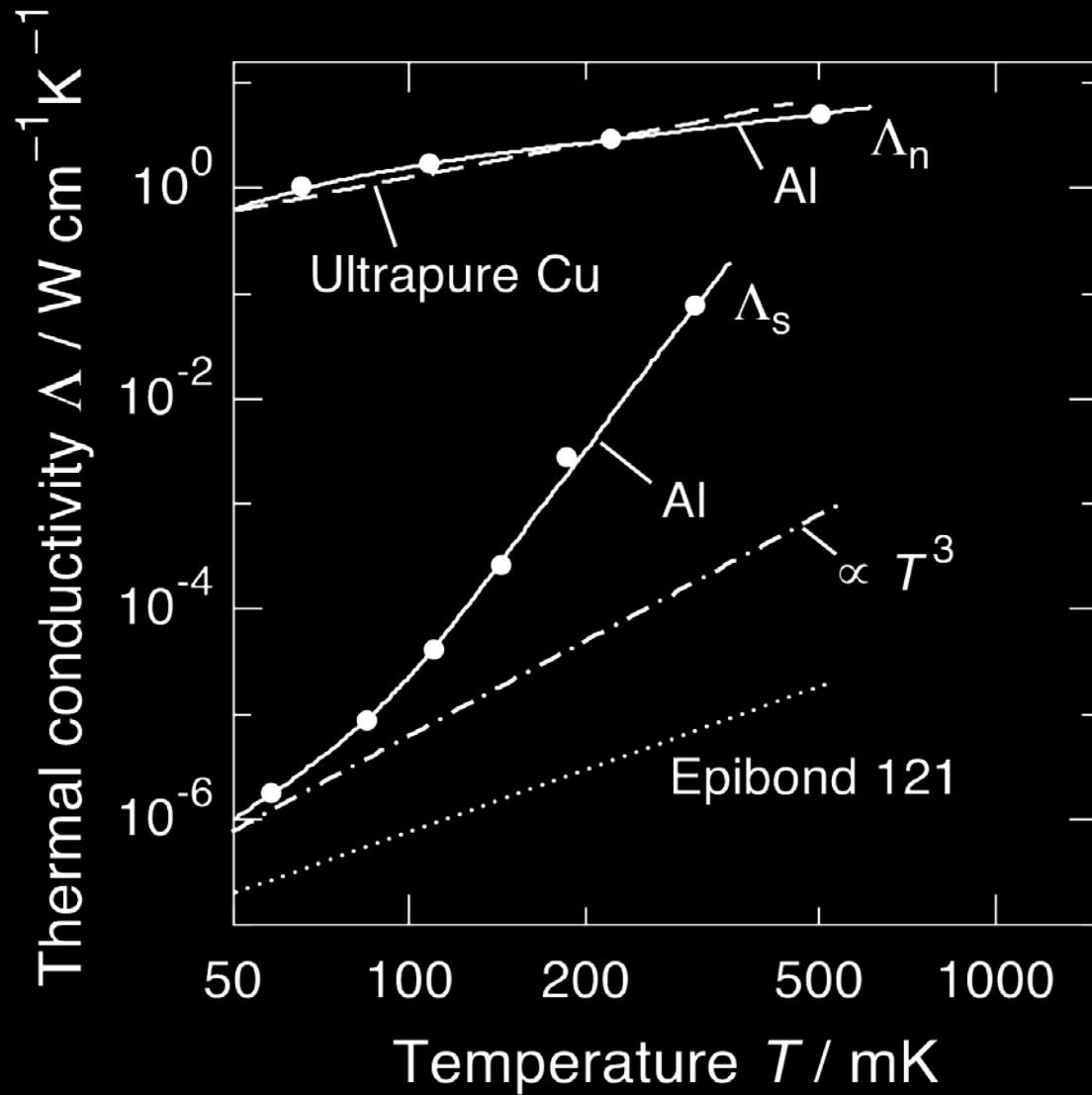
only well below T_c

open means low conductivity

eddy currents

flux trapping

Superconducting Heat Switch

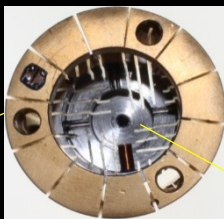


PTB Nuclear Demagnetisation Fridge

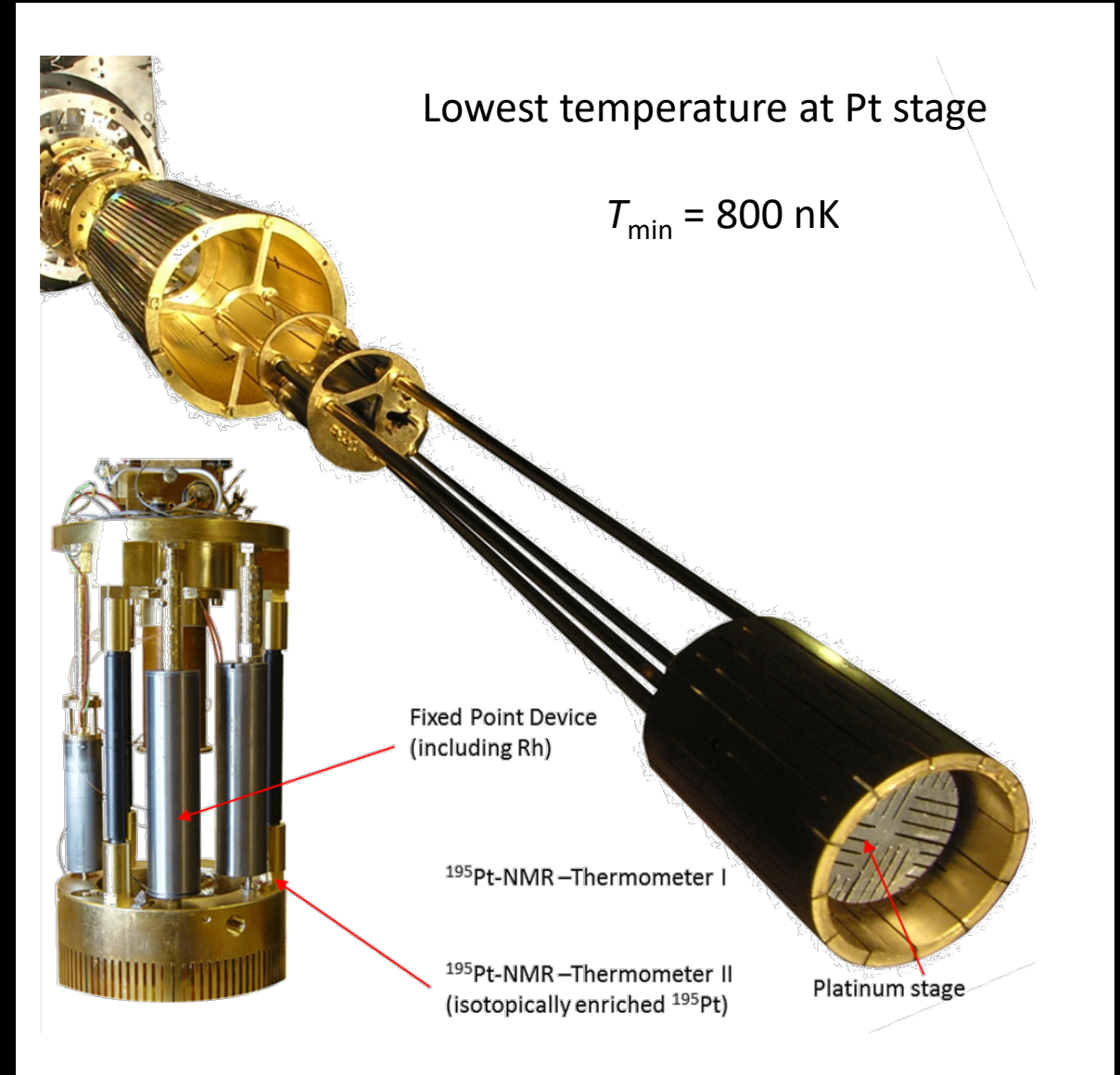
heat switch

Cu stage

heat switch



Pt stage



Cryogenic-Free Demagnetisation Fridge

Bluefors LD400 dry dilution fridge

9 T magnet

Superconducting Al heat switch

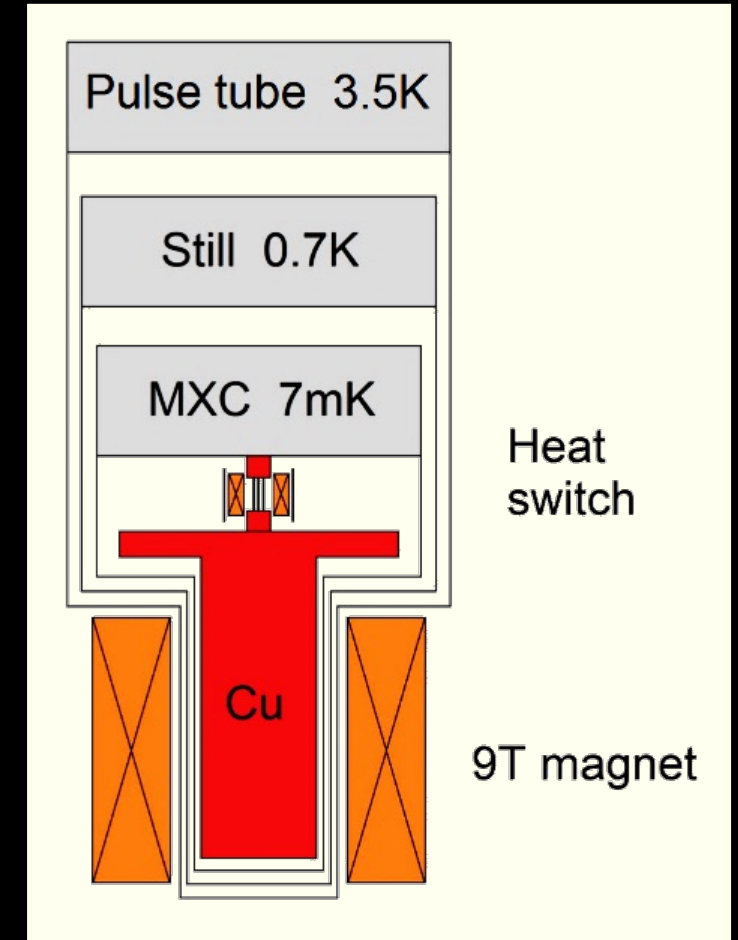
Copper nuclear stage

Problem:

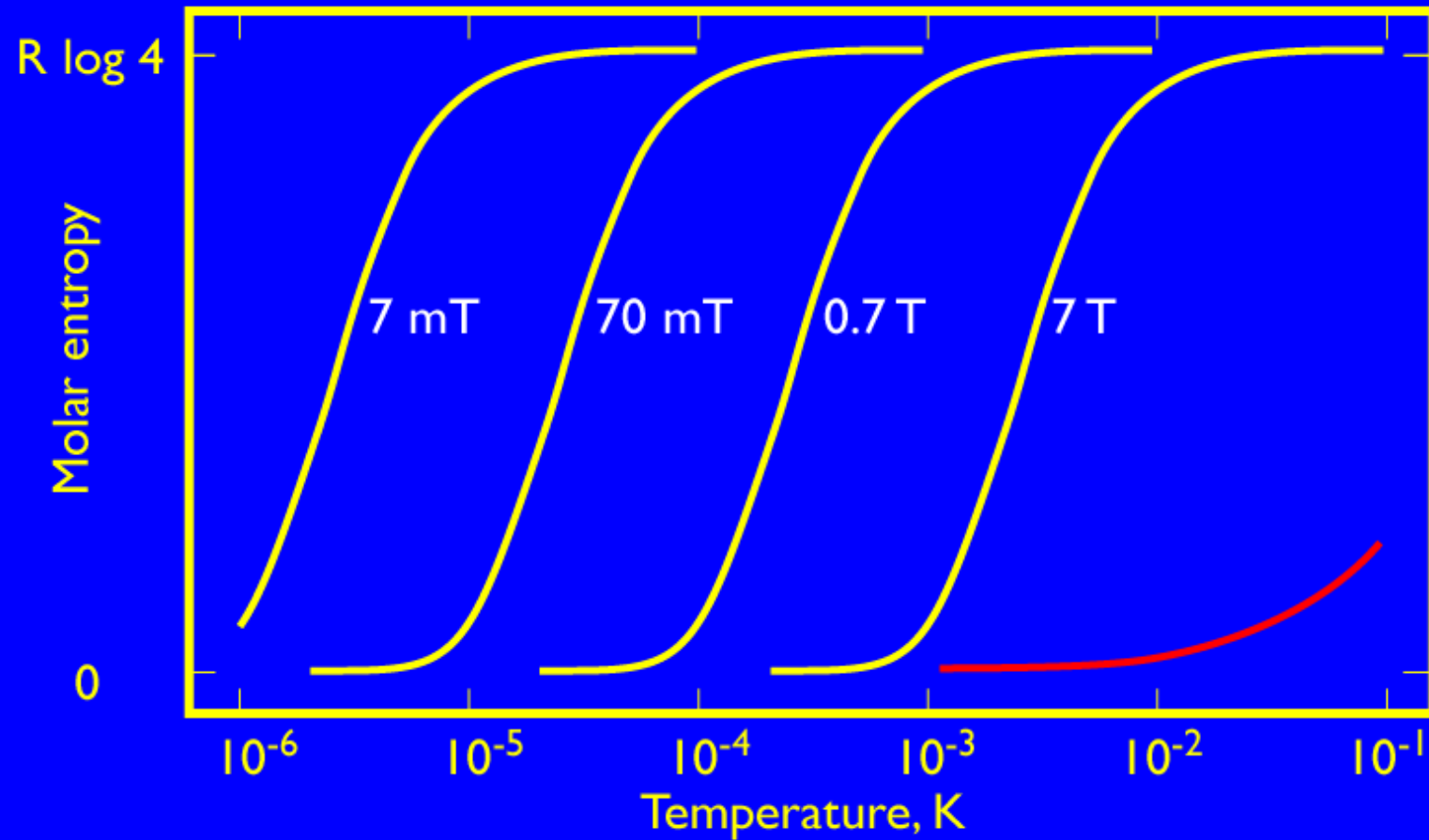
Vibrations \longrightarrow eddy current heating

$$\dot{Q}_{\text{eddy}} \propto \dot{B}^2$$

$$T_{\text{min}} \sim 0.16 \text{ mK}$$

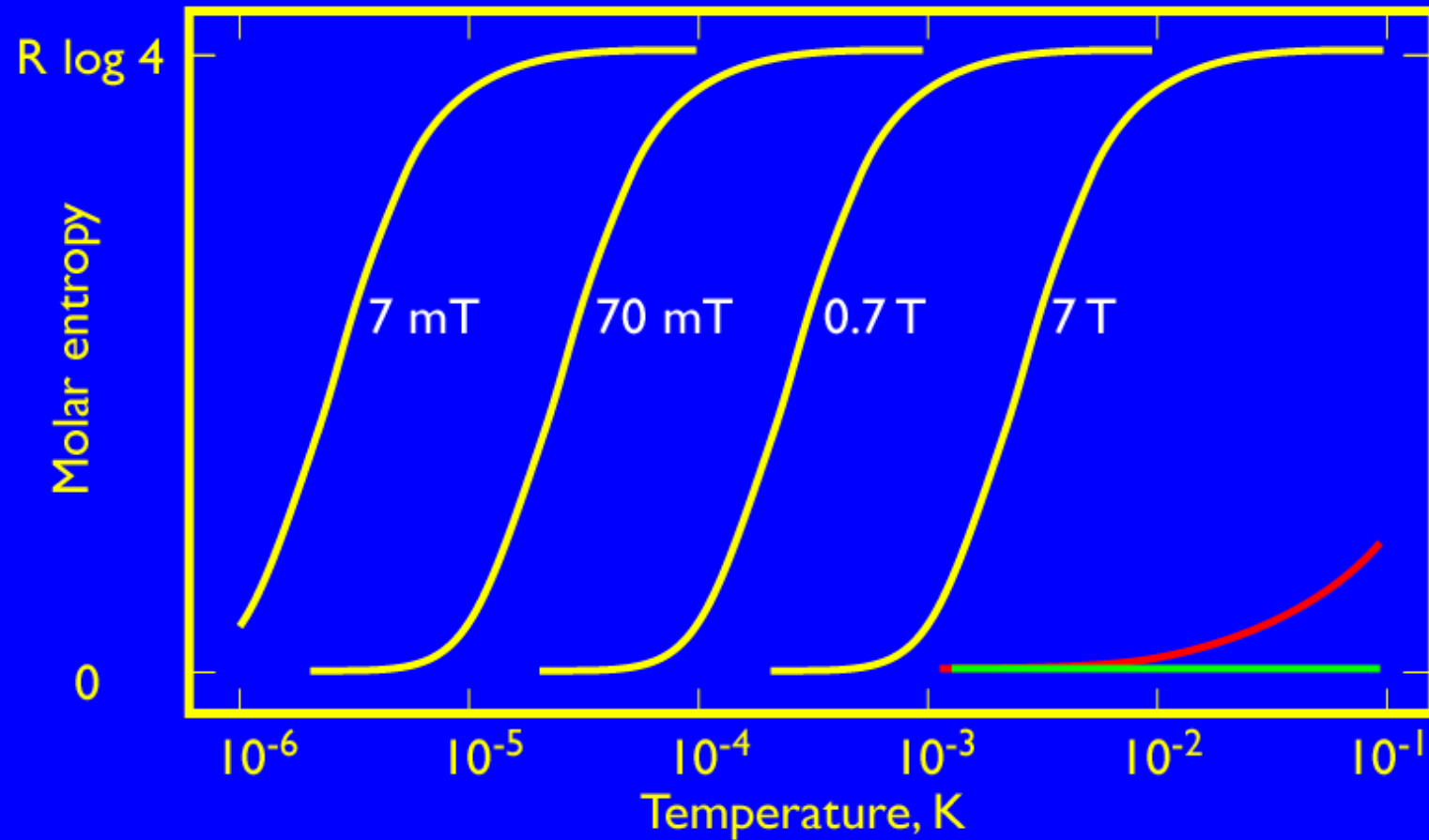


Lancaster Nuclear Demagnetization Cryostat



Molar entropy of liquid ^3He ~2 orders lower at 10 mK.

Lancaster Nuclear Demagnetization Cryostat

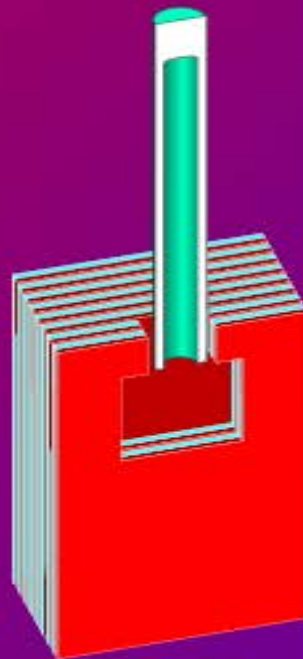


Molar entropy of Cu electron gas 8 orders lower at 10 mK.

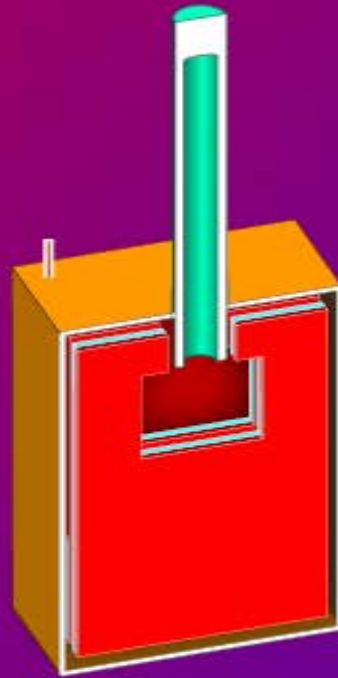
Since we only need a small volume of copper, let's get it as close to the specimen as possible, that is immerse it in the liquid.

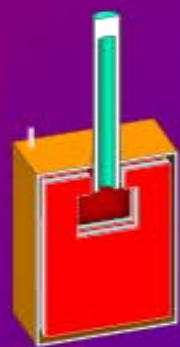
Since we only need a small volume of copper, let's get it as close to the specimen as possible, that is immerse it in the liquid.

So here is our inner stage, a stack of Cu plates, immersed in the liquid, each with a layer of sintered silver on the surface for thermal contact.

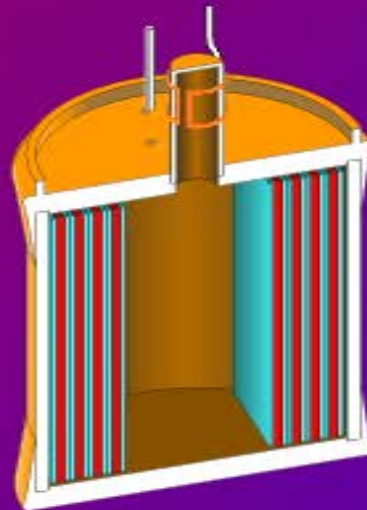


We wrap this in a thin-walled paper/epoxy box, and add a silver sinter pad to make contact for precooling and a filling tube.





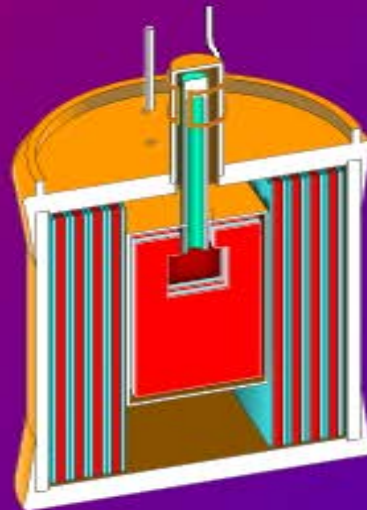
To cut down the heat leak we add a second stage, also furnished with a precooling link, and filling tube.



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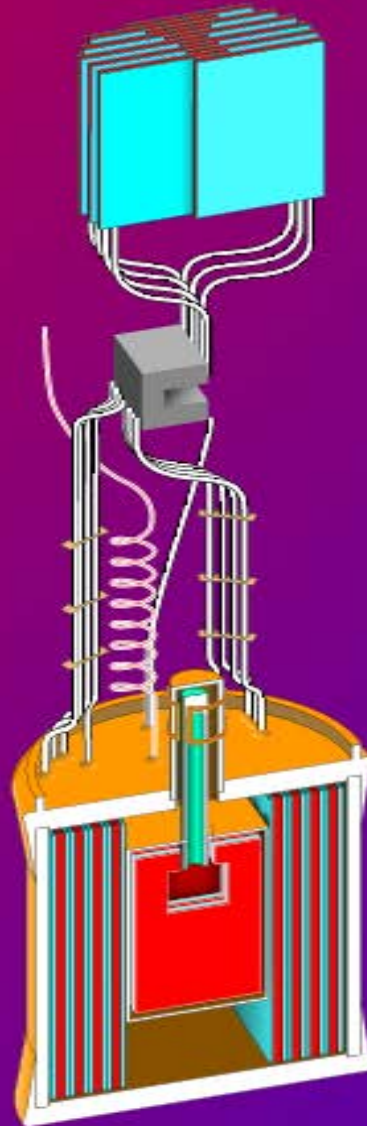
And we put the inner cell inside.

This allows the inner cell to have a very thin wall (and thus low slow-release heat leak) because the pressure is supported by the outer cell wall.

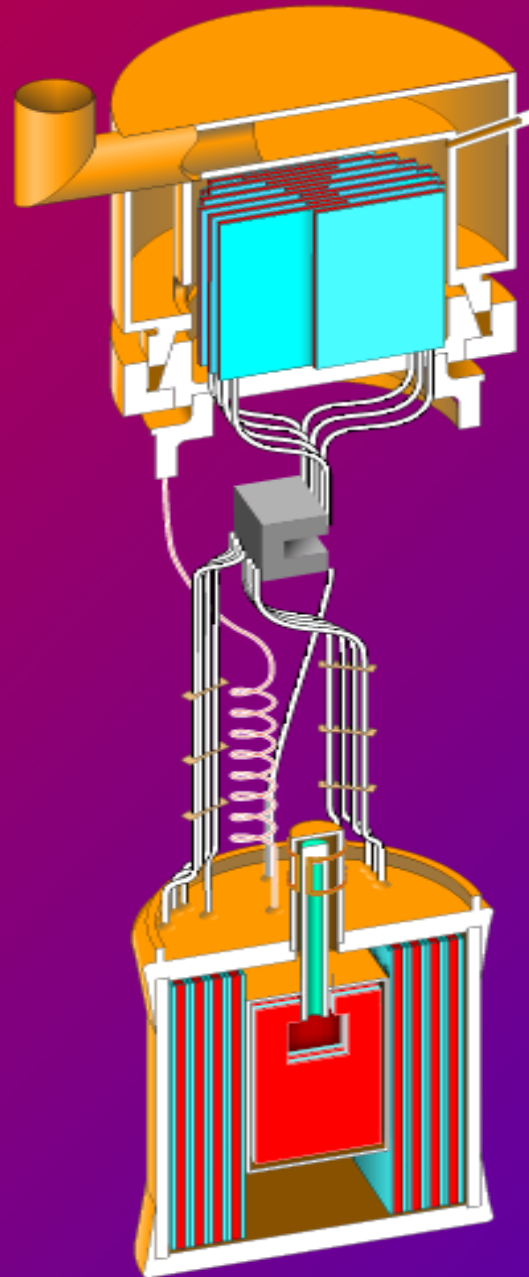


The thermal contact to the mixing chamber is also made by silver sinter plates and connected to the specimen via high purity Ag wires ($RRR \sim 10^3$):

Plus a single crystal Al heat switch.

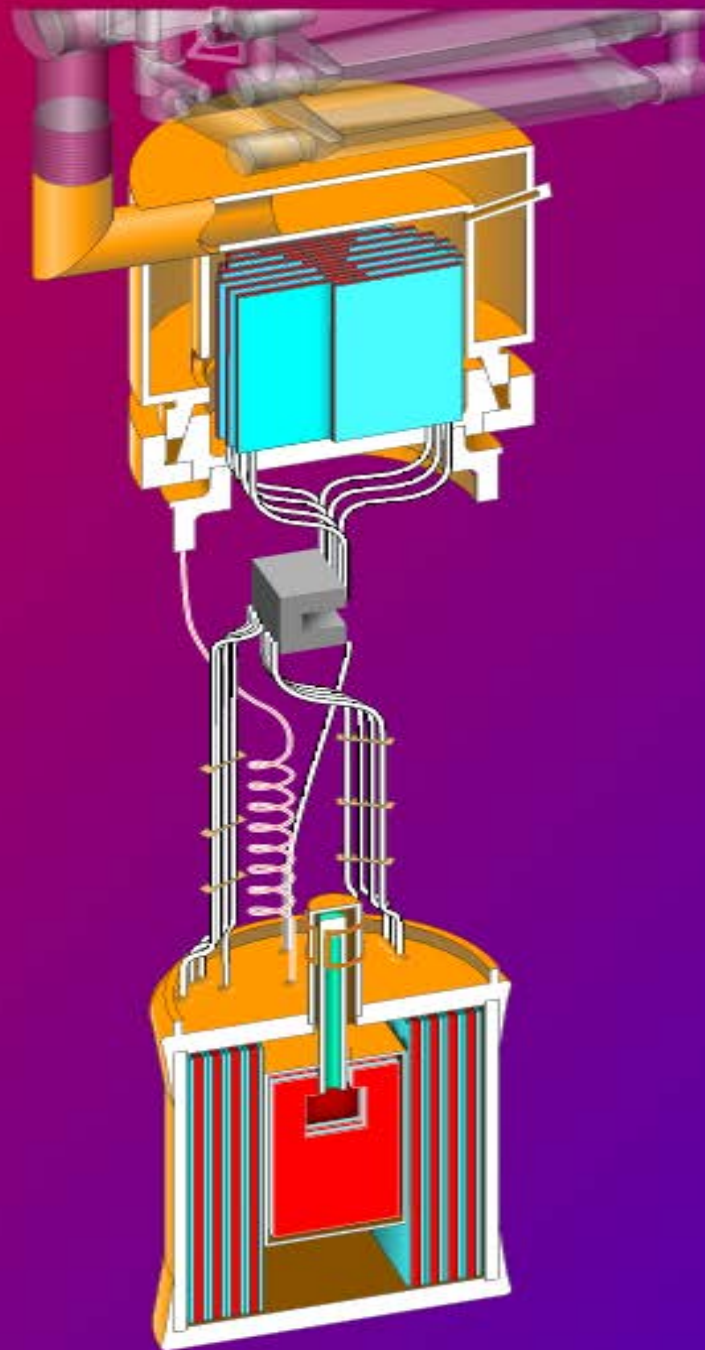


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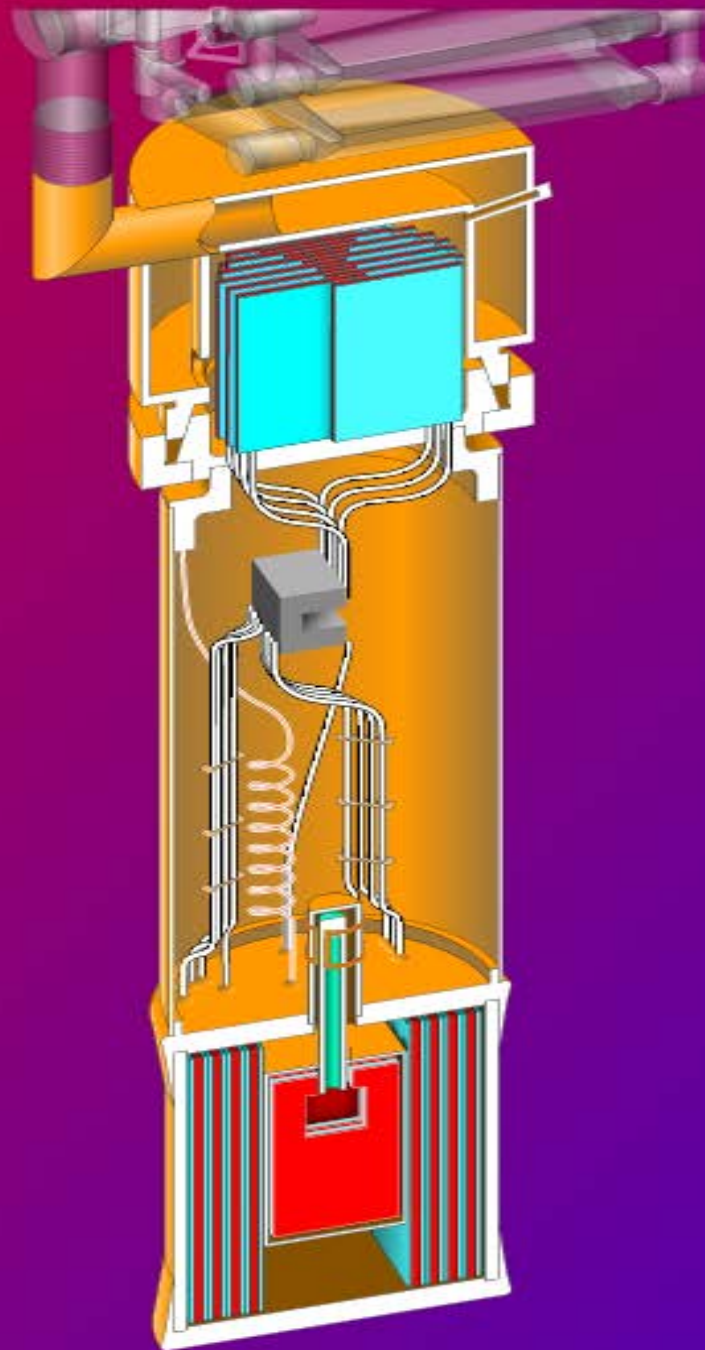
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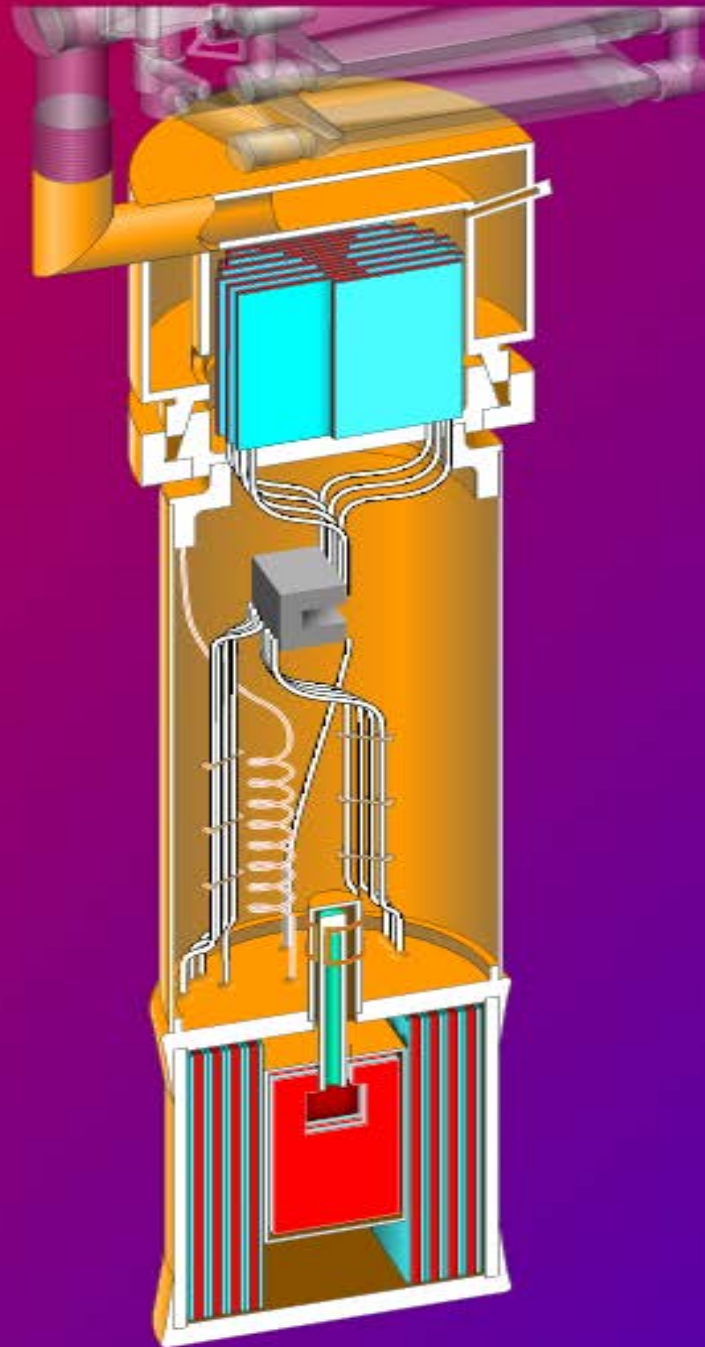
And finally the rigid thin-walled plastic support tube for the nuclear stage.



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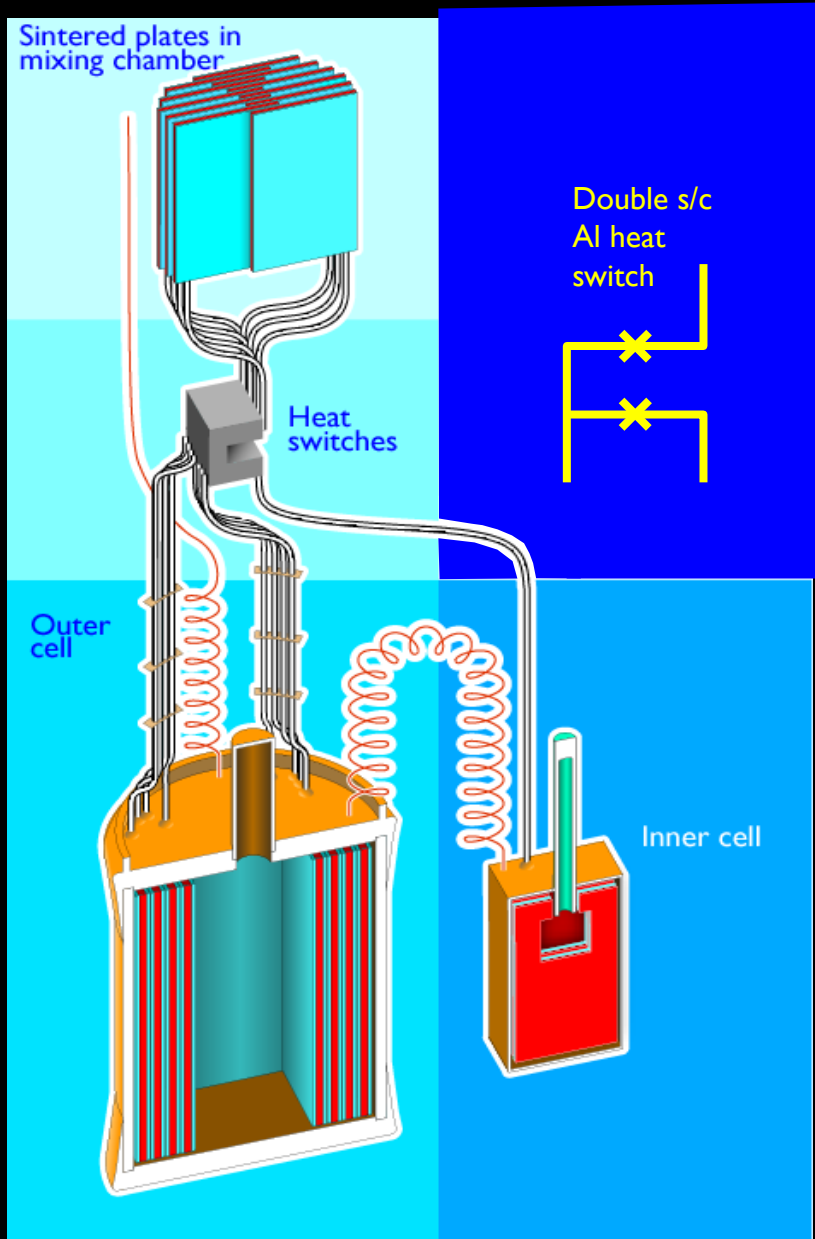
And finally the rigid thin-walled plastic support tube for the nuclear stage.



First, the “nuclear” stage is part of the experiment and dispensable.

Secondly, the whole thing is built with very low level technology, with glued plastic pieces more like schoolboy model aeroplane methods. (This temperature regime is marginal enough already.)

Lancaster Nuclear Demagnetization Cryostat



Temperature of superfluid He3
~80 micro Kelvin

Stay cold for up to a week



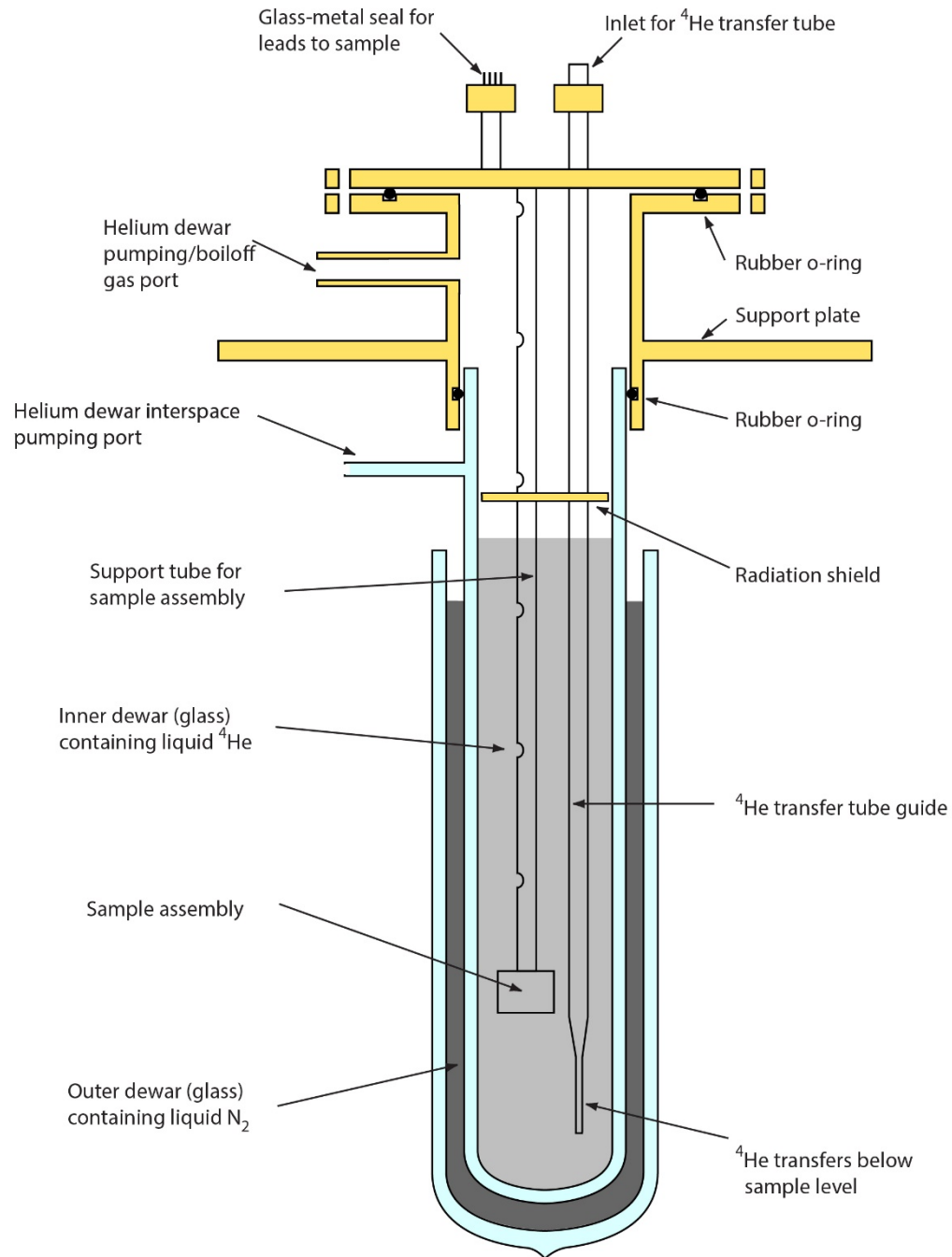
Cryostat design considerations

- Conduction along glass walls
- Black body radiation from the silvered glass walls:

$$Q = \varepsilon \sigma T^4 S,$$

where ε is the emissivity, $\sigma = 5.67 \times 10^{-8} \text{ W K}^{-4} \text{ m}^{-2}$ is Stefan-Boltzmann constant

- Conduction through residual gas in the vacuum space - catastrophic if helium!
- Conduction through the support tube
- Conduction along experimental wires.



Difficulty of Thermal Contact

The lowest sample temperature is often determined **not** by the refrigerator cooling power/base temperature but by the thermal link to the sample

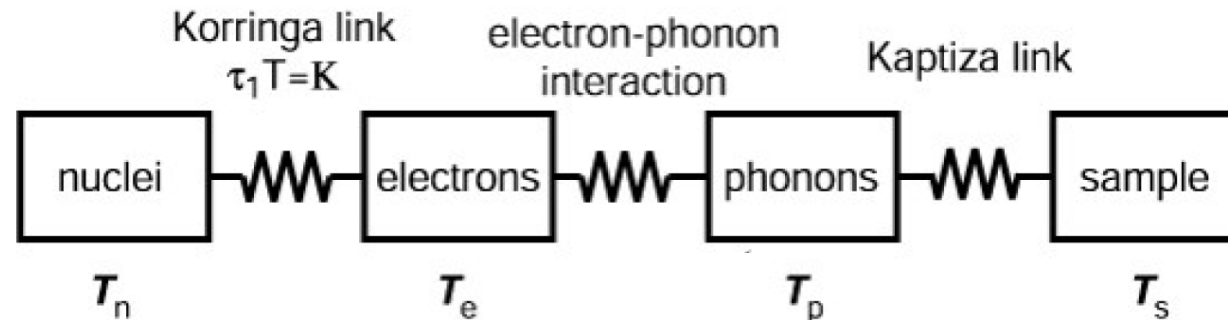
Heat flow \dot{Q} through finite thermal resistance R causes ΔT

$$\dot{Q} = \frac{\Delta T}{R}$$

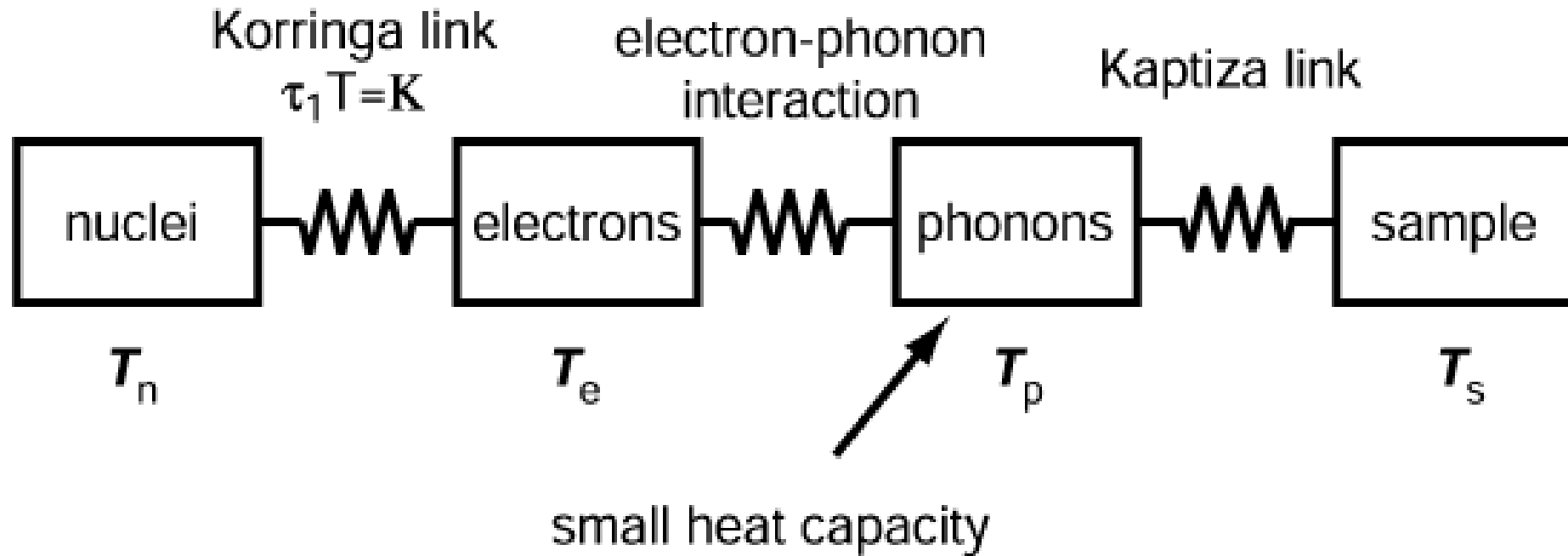
Thermal time constant is

$$\tau = RC$$

where C is the sample heat capacity



Difficulty of thermal contact



Colling Limits

- Nuclear spins $T_{\text{nuc}} \sim 100 \text{ pK}$
- Electrons $T_e \sim 1 \text{ } \mu\text{K}$
- ^3He $T_3 \sim 80 \text{ } \mu\text{K}$
- ^3He - ^4He mixtures $T_3 \sim 100 \text{ } \mu\text{K}$??

Metallic contact

Heat flow \dot{Q} through finite thermal resistance R causes ΔT : $\dot{Q} = \Delta T / R$

Weidemann-Franz law relates thermal and electrical conductivities K , σ

$$L_0 = \frac{K}{\sigma T} = 2.45 \times 10^{-8} \text{ W } \Omega \text{ K}^{-2}$$

Residual resistance ratio $RRR = \rho_{293\text{K}} / \rho_{4\text{K}}$ typically:
50 – 100 for copper wire
1 for alloys - Stainless Steel, CuNi
but for pure material, heat treating can raise this to 1000 – 10000

Thermal resistance $R = \ell / K A$, length ℓ , cross sectional area A

$$K = L_0 \frac{RRR}{\rho} T, \quad \rho = \sigma^{-1}$$

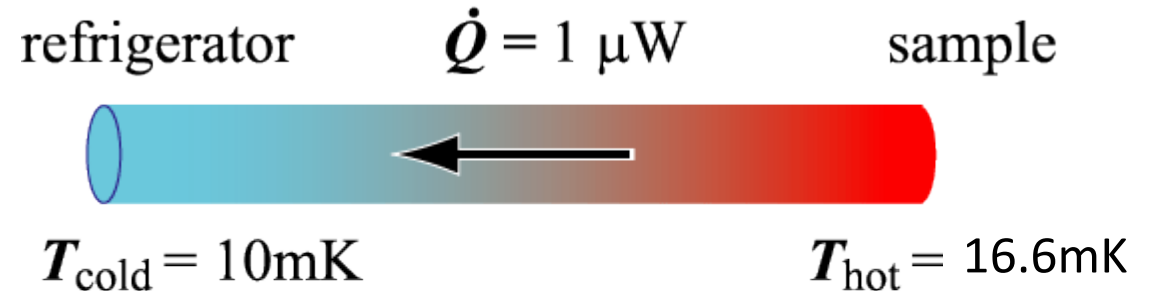
so
$$\dot{Q} = \frac{KA}{\ell} \Delta T$$

If ΔT is big, must integrate

$$\dot{Q} = \int_{T_{cold}}^{T_{hot}} \frac{KA}{\ell} dT$$

$$\dot{Q} = \frac{A}{\ell} L_0 \frac{RRR}{\rho} \frac{(T_{hot}^2 - T_{cold}^2)}{2}$$

Example: copper rod, one end at fridge base temp with a $1\mu\text{W}$ load...



$$\rho = 1.7 \times 10^{-8} \Omega \text{ m} \quad RRR = 100$$

$$\text{diameter} = 1 \text{ mm} \quad \text{length} = 1 \text{ cm}$$

Metallic sinters

Boundary resistance between fine metal particles (sinters) and the He liquids inversely proportional to the sinter area.

T dependence complicated (treat with care):

$${}^3\text{He} - {}^4\text{He} \quad R \propto 1 / T^2$$

$${}^3\text{He } T > 10 \text{ mK} \quad R \propto 1 / T^3$$

$${}^3\text{He } T < 10 \text{ mK} \quad R \propto 1 / T$$

Screw Joints

screw joints can have contact resistance $< 0.1 \mu\Omega$ with care.

$$R \sim 4 / T \text{ K}^2/\text{W}$$

Insulators

Best material

Vespel SP22

✓ strong, machinable

✓ low thermal conductivity

$$K = 17 \times 10^{-4} T^2 \text{ W m}^{-1} \text{ K}^{-1}$$

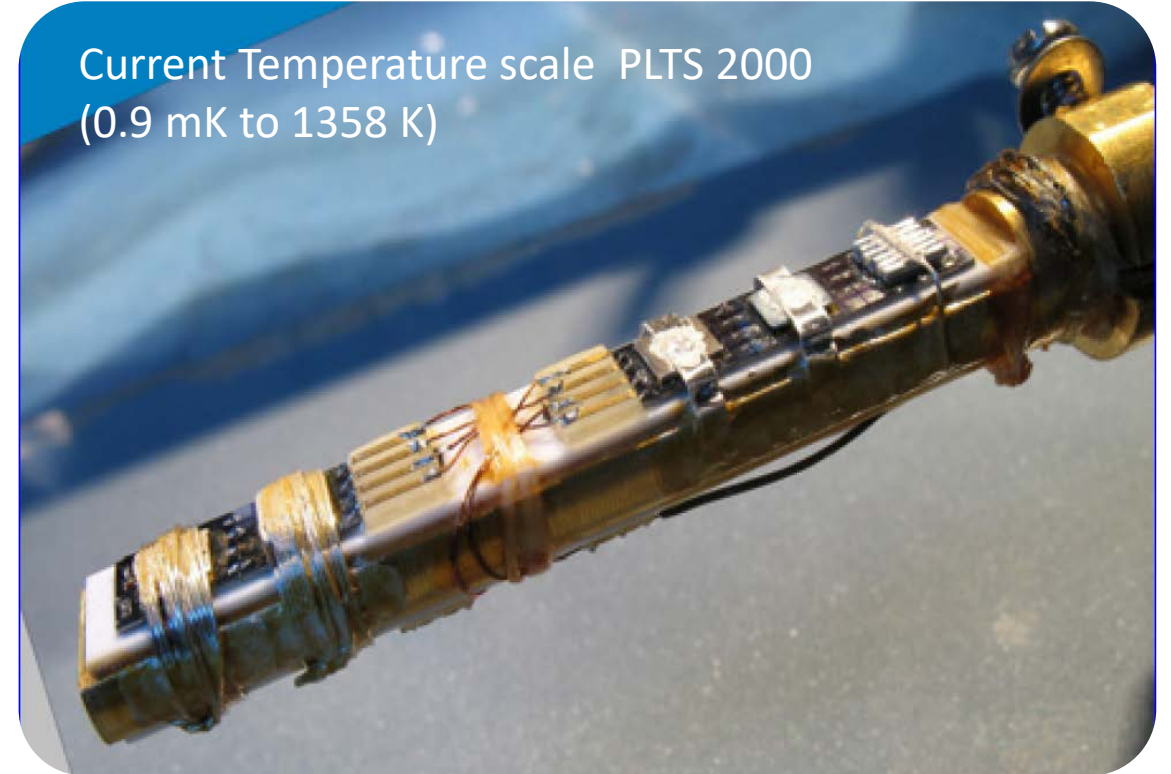
Locatelli, Cryogenics **16**, 374, 1976

Thermometry at Low Temperatures

- determination of temperature is often as difficult as the experiment itself
- anything can be used as a thermometer as long it has a temperature dependence
- temperature is by far the most uncertain scale ...

Primary thermometers: can be used without any prior calibration

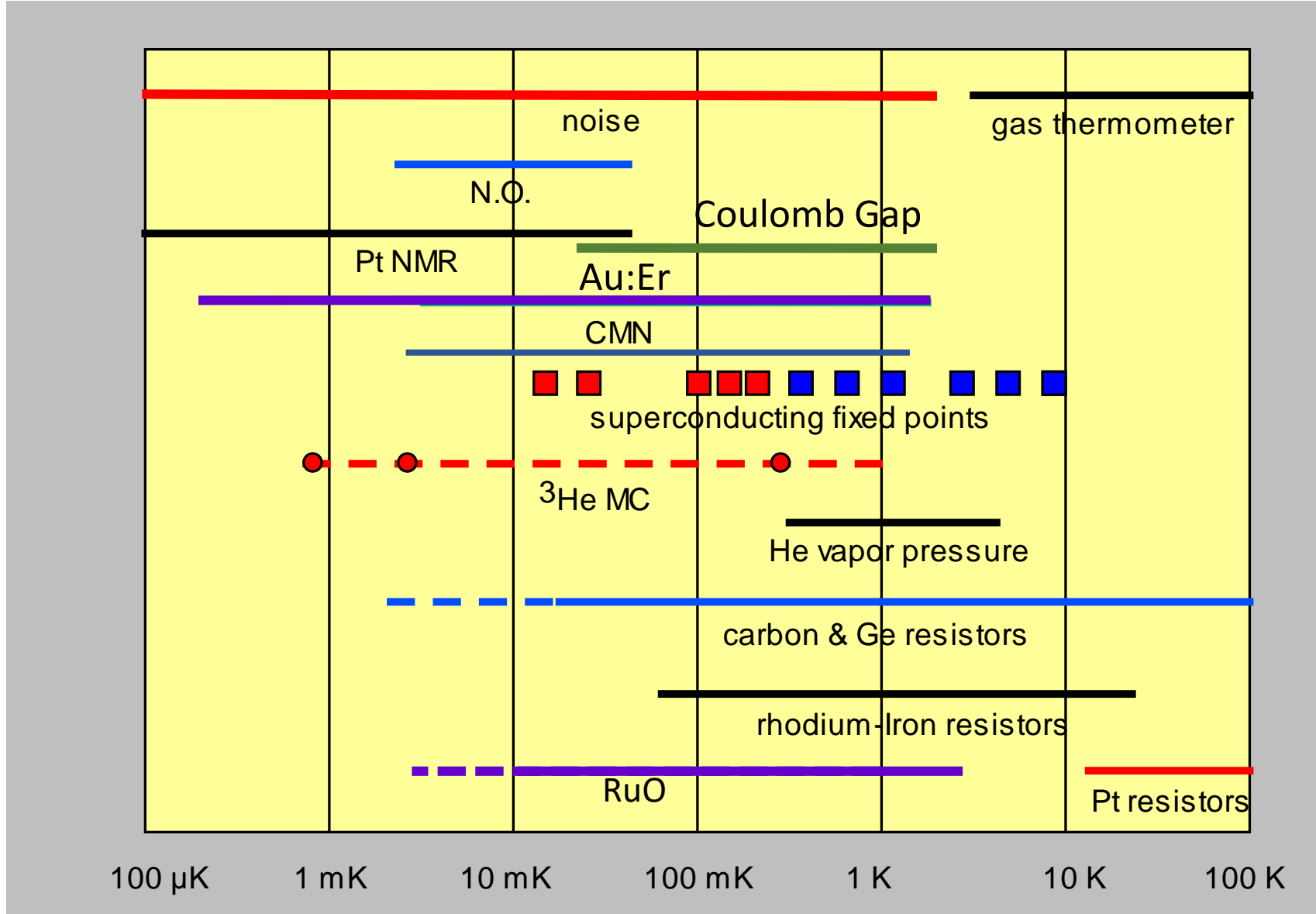
Secondary thermometers: must be calibrated against an other thermometer



Primary thermometers
Superconducting fix points
Noise

Secondary thermometers
Resistance
Capacitance
Magnetic susceptibility

Working Range of Thermometers



Books

- O. V. Lounasmaa: “Experimental Principles & Methods below 1K”, Academic Press 1974
- D. S. Betts “Refrigeration and Thermometry below 1K”, Sussex Univ Press 1976
- R. C. Richardson & E. N. Smith “Experimental Techniques in Condensed Matter Physics at Low Temperatures”, Addison Wesley 1988
- F. Pobell, “Matter & Method at Low Temperatures”, Springer Verlag 2007
- G. White & P. Meeson, “Experimental Techniques in Low-Temperature Physics”, Oxford University Press 2002
- C. Enss & S. Hunklinger, “Low-Temperature Physics”, Springer Verlag 2005
- J. W. Ekin, “Experimental Techniques for Low Temperature Measurements”, Oxford University Press 2011