

Hydrofluoric Acid-Free Titanium Etching for Rare-Event Searches

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Introduction

- Titanium a common detector material choice for rare event searches; high strength, machinability, low density
- Surface treatment required for direct use or prior to further treatment, e.g. alloying with electroformed copper [1, 2, 3]
- High chemical resistance: hydrofluoric acid generally required
- HF challenging chemical, even more so used underground
- An alternative: sulphuric acid, already used in underground laboratories [4], [5]. H₂SO₄ used for Ti etching in other fields [6]

H₂SO₄ Titanium Etching

- ∅3.15 cm, 0.9 mm thick grade 1 Ti coupons used
- Sheet cast billet, pressed, then hot and cold rolled
- Coupons immersed in 99.8% analytical reagent grade ethanol, cleaning in an ultrasonic bath at 30°C for 30 minutes. Dried at 50°C for 1 h
- Weighed before and after etching
- Three coupons tested for each condition, studying effect of concentration and temperature



Figure 1: Set-up.

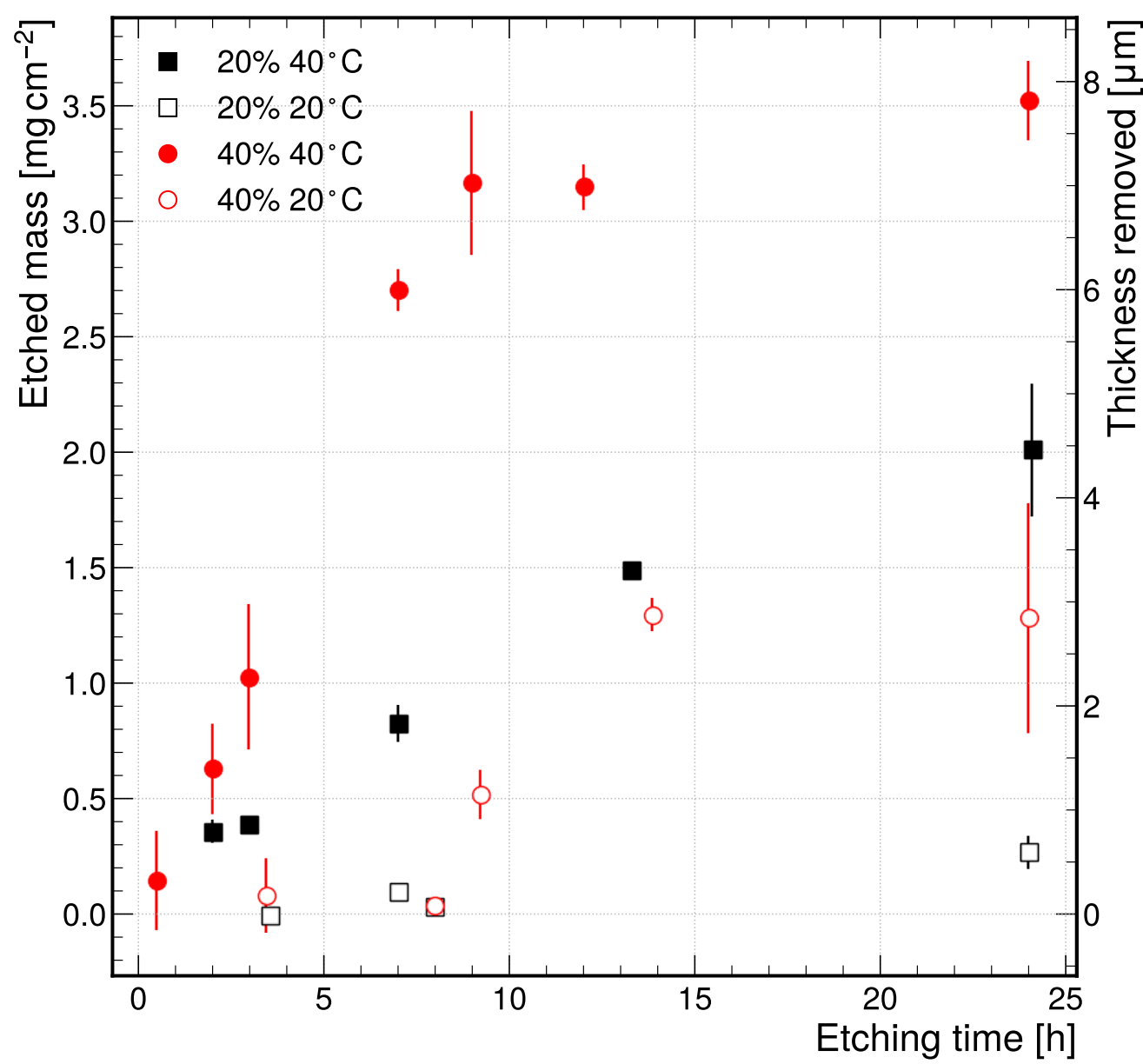


Figure 2: Mass change after etching.

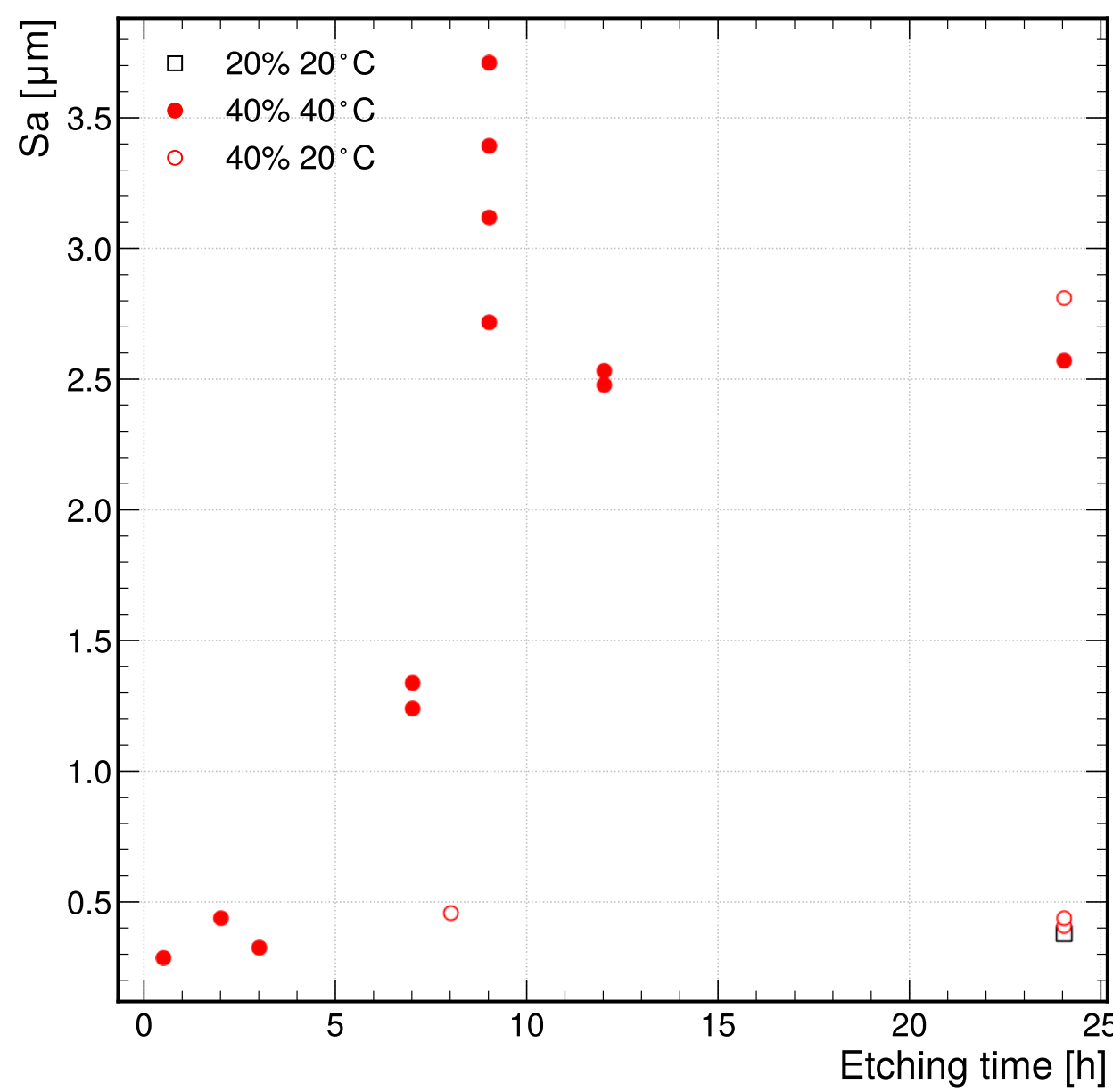


Figure 3: Surface roughness, Sa.

- Higher etching rate observed with higher temperature and concentration
- Similar trend seen in surface roughness

Surface Study: Electron Microscopy

- JEOL 7000F field emission gun-scanning electron microscope (SEM) in Secondary Electron Imaging (SEI) and Backscattered Electron Imaging (BEI) modes, for topology and composition investigations, respectively
- Composition from Energy-Dispersive X-ray (EDX) spectroscopy
- Debris seen from O and Fe prior to etching. Not observed after etching

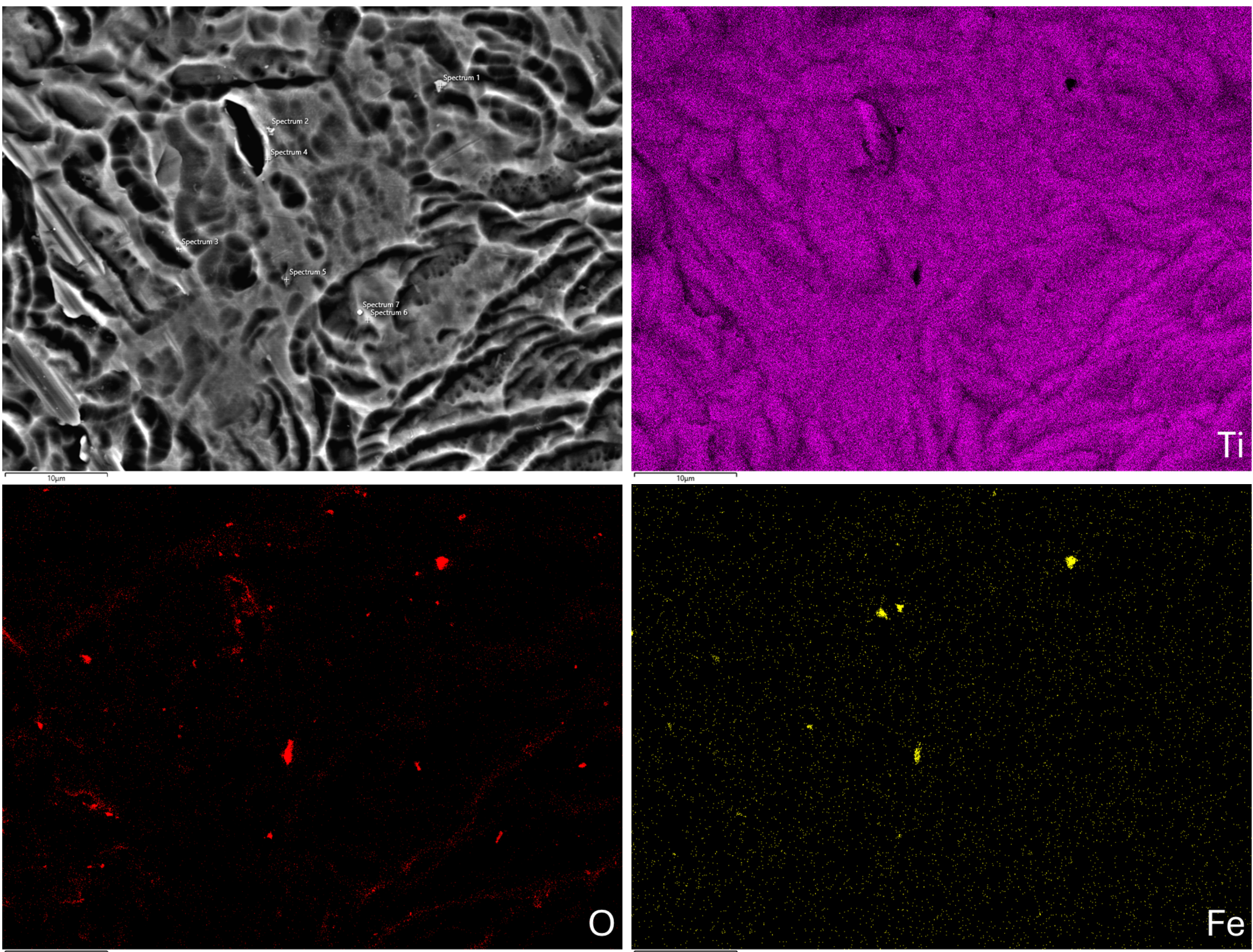


Figure 4: (Top left) BSE image of Ti surface prior to etching and (left to right in rows) the EDX maps of Ti, O, and Fe.

Surface Study: Optical Microscopy

- KEYENCE VHX-7000 digital optical microscope
- Study surface change before and after etching, including roughness

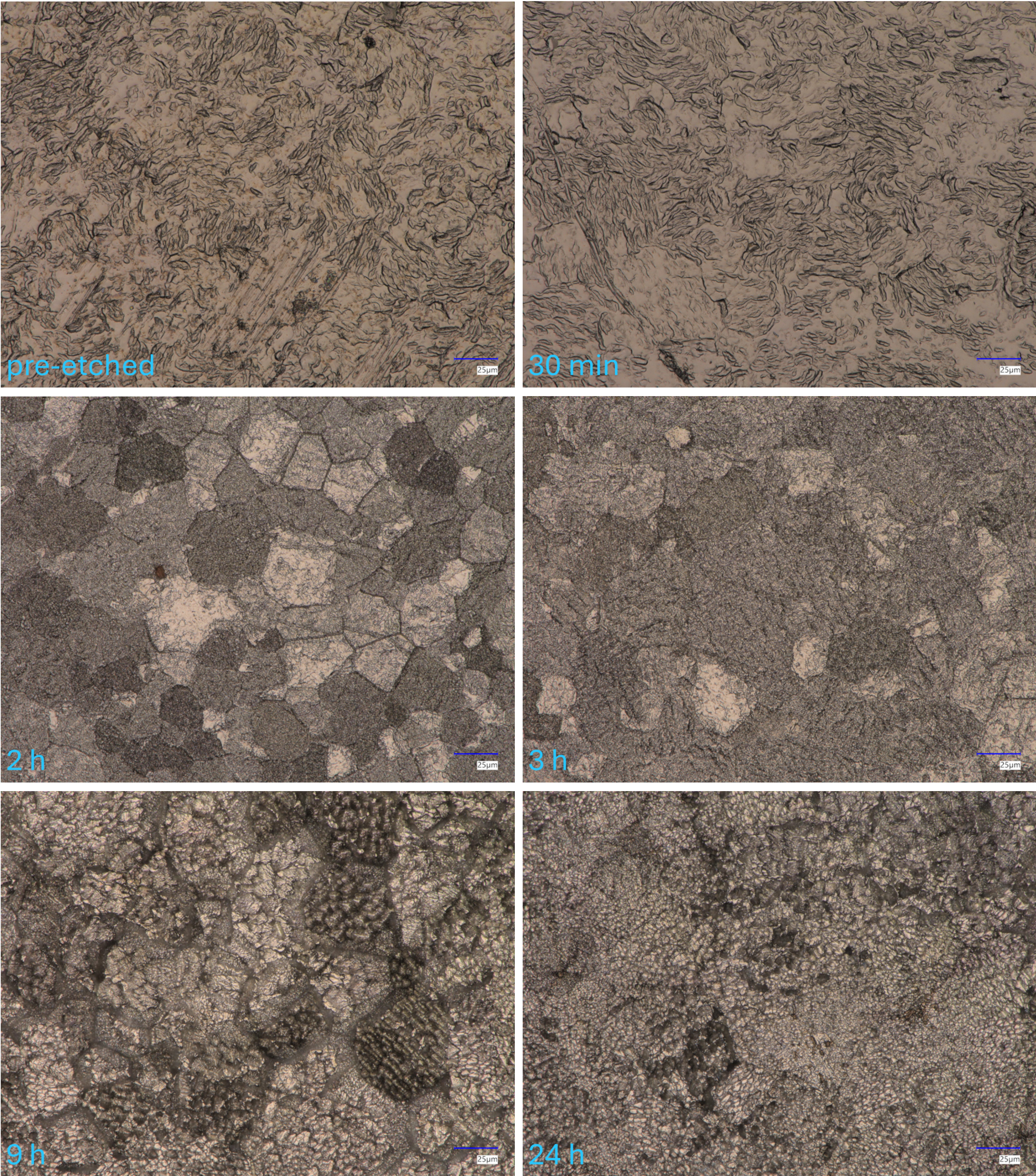


Figure 5: Ti surface x1000 using 40% H₂SO₄ at 40 °C.

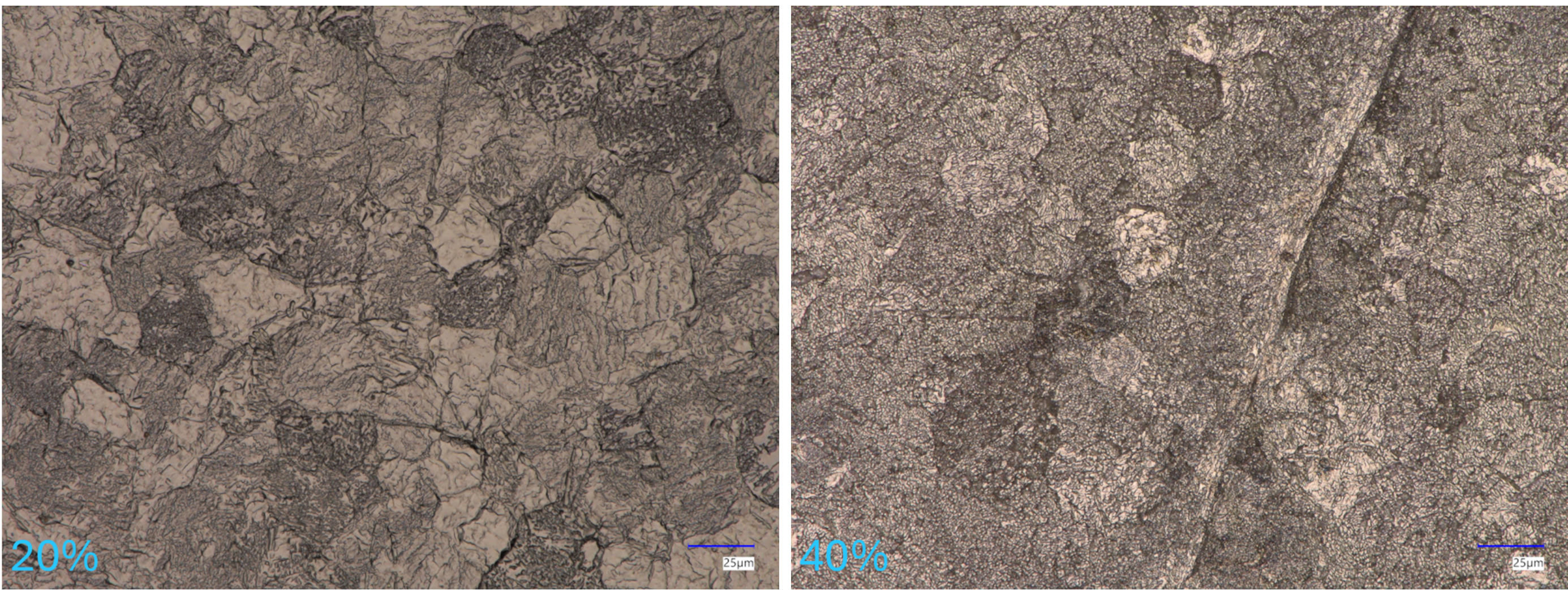


Figure 6: Ti surface x1000 after etching with H₂SO₄ after 24 h.

- Topology of the surface for 20% H₂SO₄ at 20 at 24 h matches that at 2 or 3 h for 40% at 40 °C. Agrees with etched mass measurements

Conclusion

- H₂SO₄ Ti etching for a range of concentrations and temperatures
- Demonstrated etching at room temperature
- Alternative to HF-based etching
- Benefits for safety, ease of use, known constituents, and disposal

Bibliography

- [1] D. Spathara, P. Knights, and K. Nikolopoulos, "Design of high-strength, radiopure, electroformed copper-based alloys for rare-event searches: Impact of layer configuration on heat treatments," Sep. 2025. arXiv: 2509.07406 [physics.ins-det].
- [2] D. Spathara, P. Knights, and K. Nikolopoulos, "Materials design for the synthesis of high strength radiopure copper alloys for rare event detection," *Commun. Phys.*, vol. 8, no. 1, p. 464, 2025. DOI: 10.1038/s42005-025-02363-z arXiv: 2507.00799 [physics.ins-det].
- [3] L. Balogh et al., *Copper electroplating for background suppression in the NEWS-G experiment*, 2021. DOI: https://doi.org/10.1016/j.nima.2020.164844
- [4] P. Knights, I. Manthos, K. Nikolopoulos, G. Rogers, D. Spathara, and P. Walters, "Hydrofluoric acid-free titanium etching for rare-event searches," Jun. 2026. arXiv: 2606.24524 [physics.ins-det].
- [5] P. Knights, K. Nikolopoulos, G. Rogers, D. Spathara, and P. Walters, "Titanium for rare-event searches: Hydrofluoric acid-free etching," Aug. 2025. arXiv: 2508.21029 [physics.ins-det].
- [6] S. Ban, Y. Iwaya, H. Kono, and H. Sato, "Surface modification of titanium by etching in concentrated sulfuric acid," *Dental Materials*, vol. 22, no. 12, pp. 1115–1120, 2006, ISSN: 0109-5641. DOI: https://doi.org/10.1016/j.dental.2005.09.007

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