



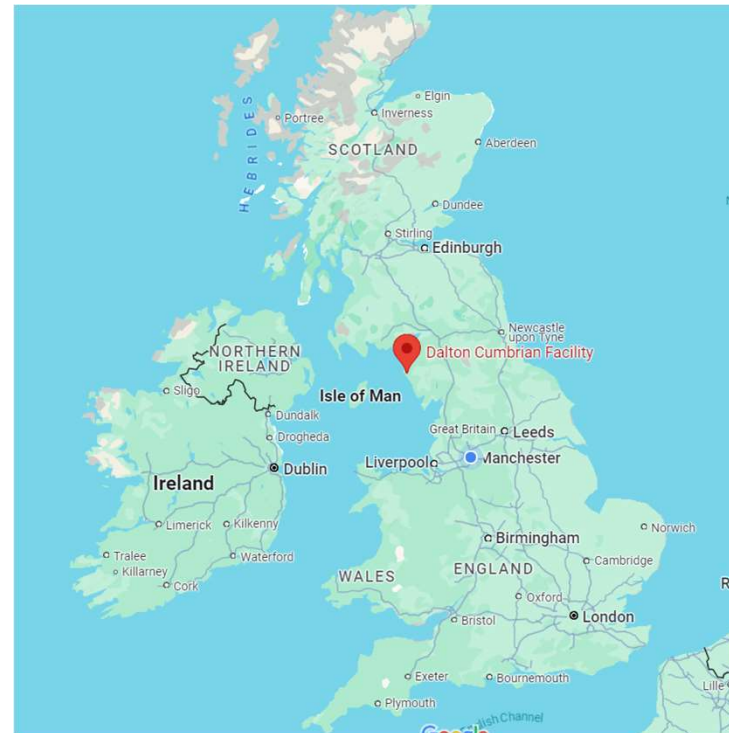
Radiation Chemistry Experiments at Upgraded CLARA Facility

Alex Baidak, The University of Manchester

4th CLARA User Meeting, 03-Dec-2024



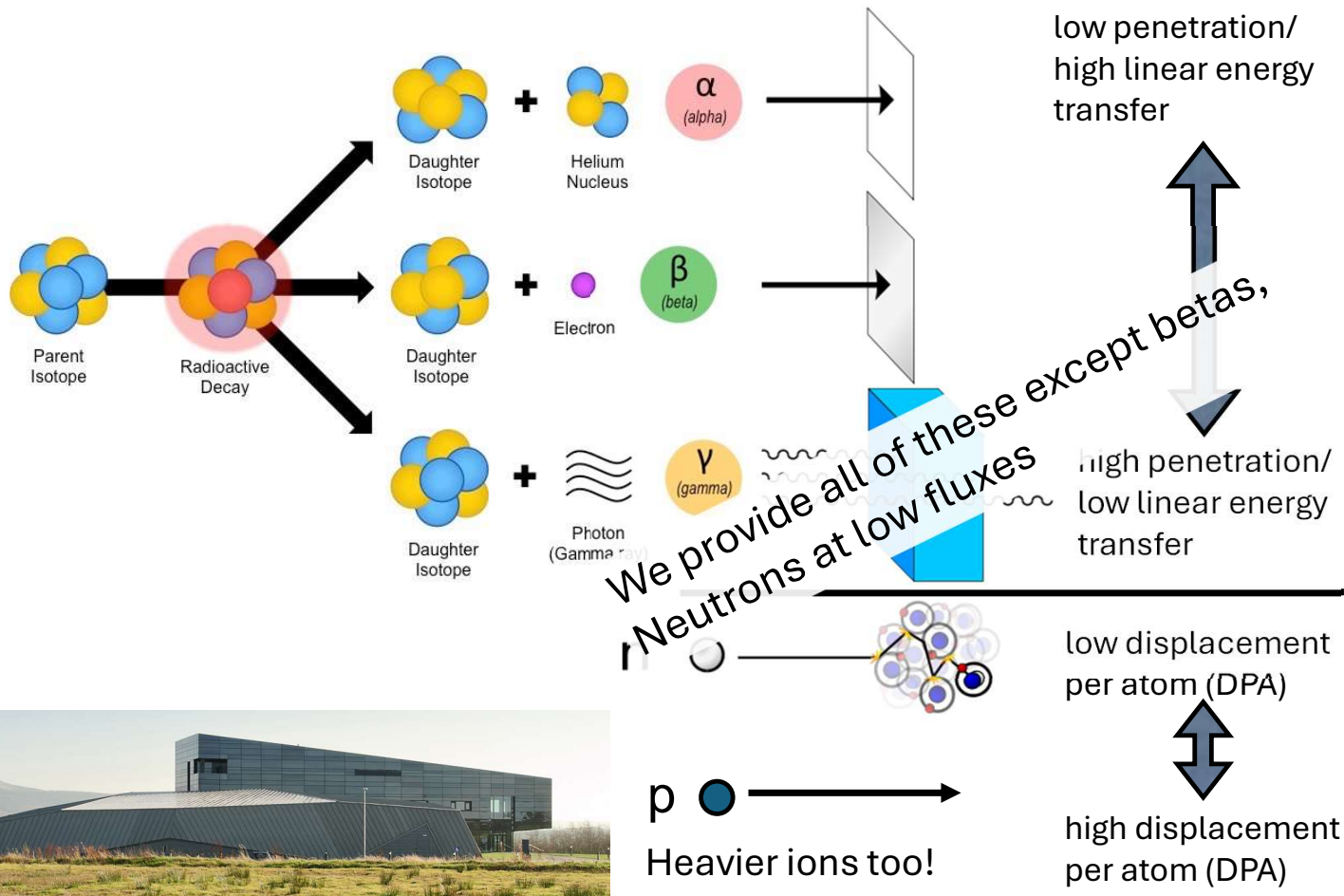
The Dalton Cumbrian Facility (DCF)



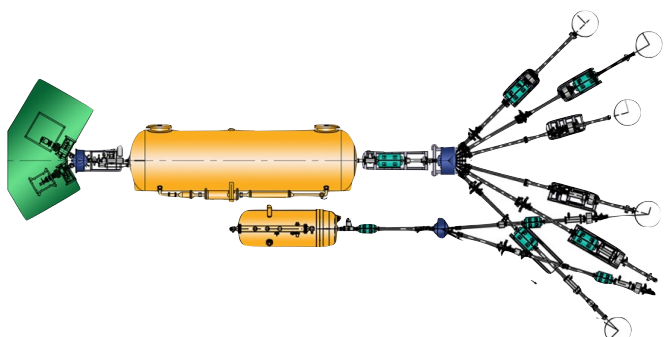
Purpose: foster and support the understanding of radiation-driven processes and the responsible use of radiation.

Vision: support innovation in applications of radiation science and engineering.

DCF's Radiation Provision



DCF's Radiation Sources



Two ion accelerators

5MV tandem ion accelerator

-10 MeV $^1\text{H}^+$ up to 100 μA ,

-15 MeV $^4\text{He}^{2+}$ up to 15 μA

-35 MeV $^{12}\text{C}^{6+}$ up to 150 nA

2.5MV single ended ion accelerator

e.g. $^1\text{H}^+$ and $^4\text{He}^{2+}$

Feeding 6 beamlines, including: Hot cell (active samples) Dual Beam, Extensive Characterisation (SIMS, EELS, Tensile Rig, ...)

Foss Therapy Services 812 ^{60}Co irradiator



~ 1.2 MeV gamma rays
~ 330 Gy/min – 30 Gy/hr

X-ray cabinet source Precision MultiRad350

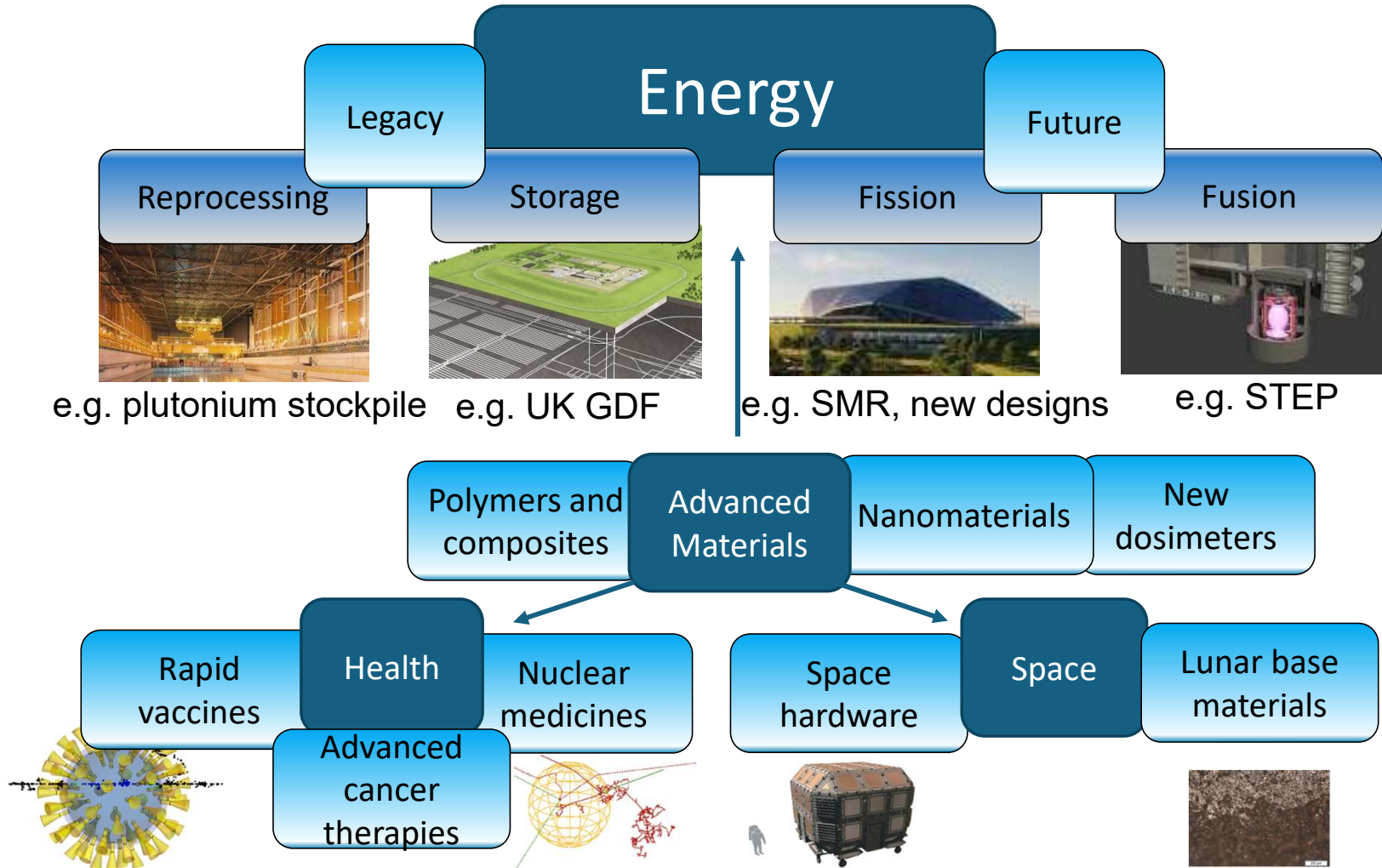


30 – 350 kVp X-rays
up to 140 Gy/min

All supported by:

- A team of 12 technical specialists
- Extensive on-site analytical capabilities
- Robotic/automated sample handling
- Cutting-edge radiation transport and radiation chemistry modelling

Why The Dalton Cumbrian Facility? Because Radiation Changes Matter



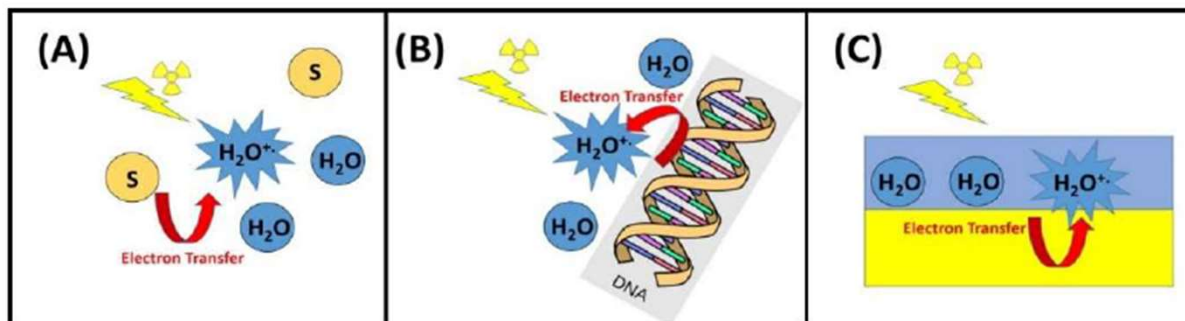
Unresolved Questions in Ultrafast Radiation Chemistry

Primary ionization step: $\text{H}_2\text{O} \longrightarrow \text{H}_2\text{O}^+ + \text{e}^-$

Elementary proton transfer: $\text{H}_2\text{O}^+ + \text{H}_2\text{O} \longrightarrow \text{OH} + \text{H}_3\text{O}^+$

- Lifetime of the radical cation as a function of T and pH
- Absorption spectra of the H_2O^+ and $\cdot\text{OH}$ under different conditions
- The extent of delocalisation of the hole in H_2O^+
- Dynamics of the solvent rearrangement upon ionisation
- How the electron dynamics couples to the ion dynamics
- Pre-solvated electron capture
- Dissociative electron capture

Ultrafast Radiation Chemistry: Possible Scenarios



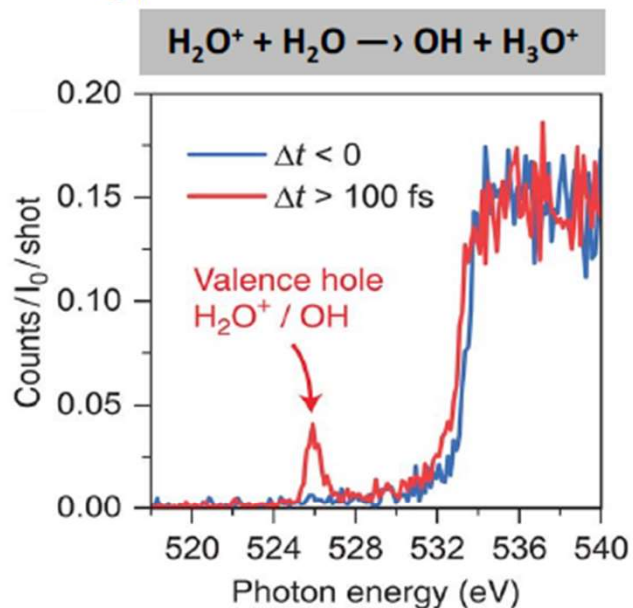
Radiolysed water can form radical cation that engages in an electron transfer in competition with the proton transfer:

- (A) highly concentrated solutions in which the water radical cation can oxidise solute molecules directly – e.g. spent nuclear fuel processing (PUREX)
- (B) highly-structured water layers formed in contact with biomolecules, such as DNA – cancer therapy
- (C) water/solid interface – safe nuclear waste storage, Pu stewardship, corrosion in reactors and storage ponds.

Potential Experiment 1: Water Radiolysis under Extreme pH

Pure water:

Observing proton transfer via OH radical signature

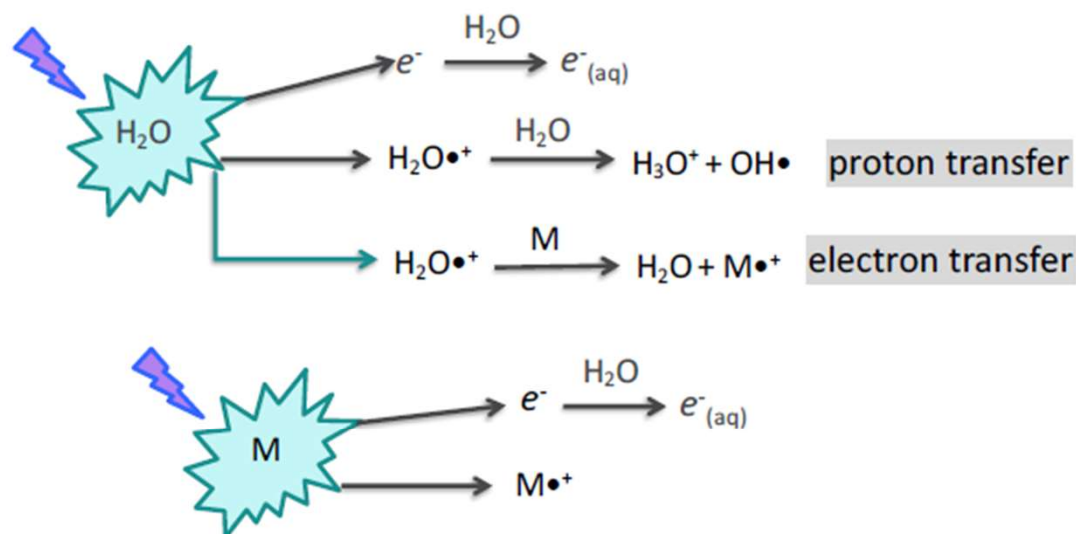


- Clean hydroxyl radical signature in water window
- Used as a tracer for reaction rates

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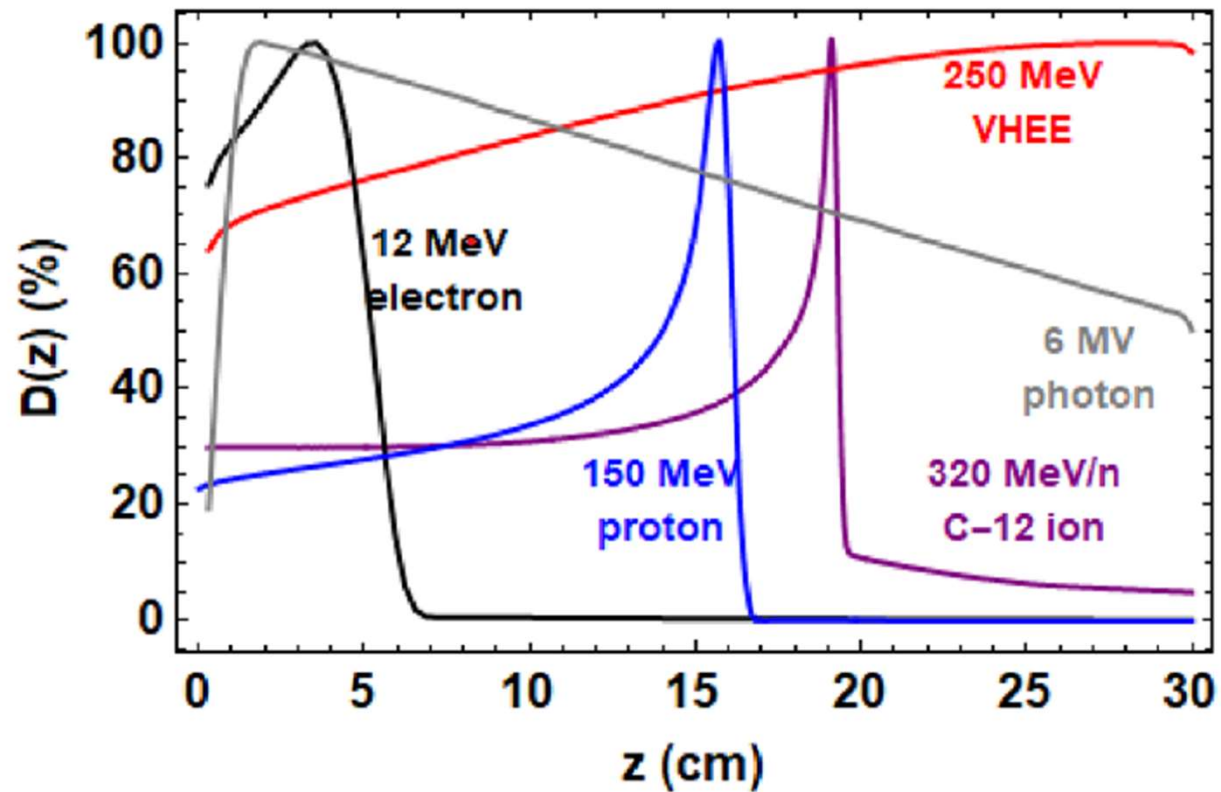
Highly alkaline system:

Competition between proton and electron transfer

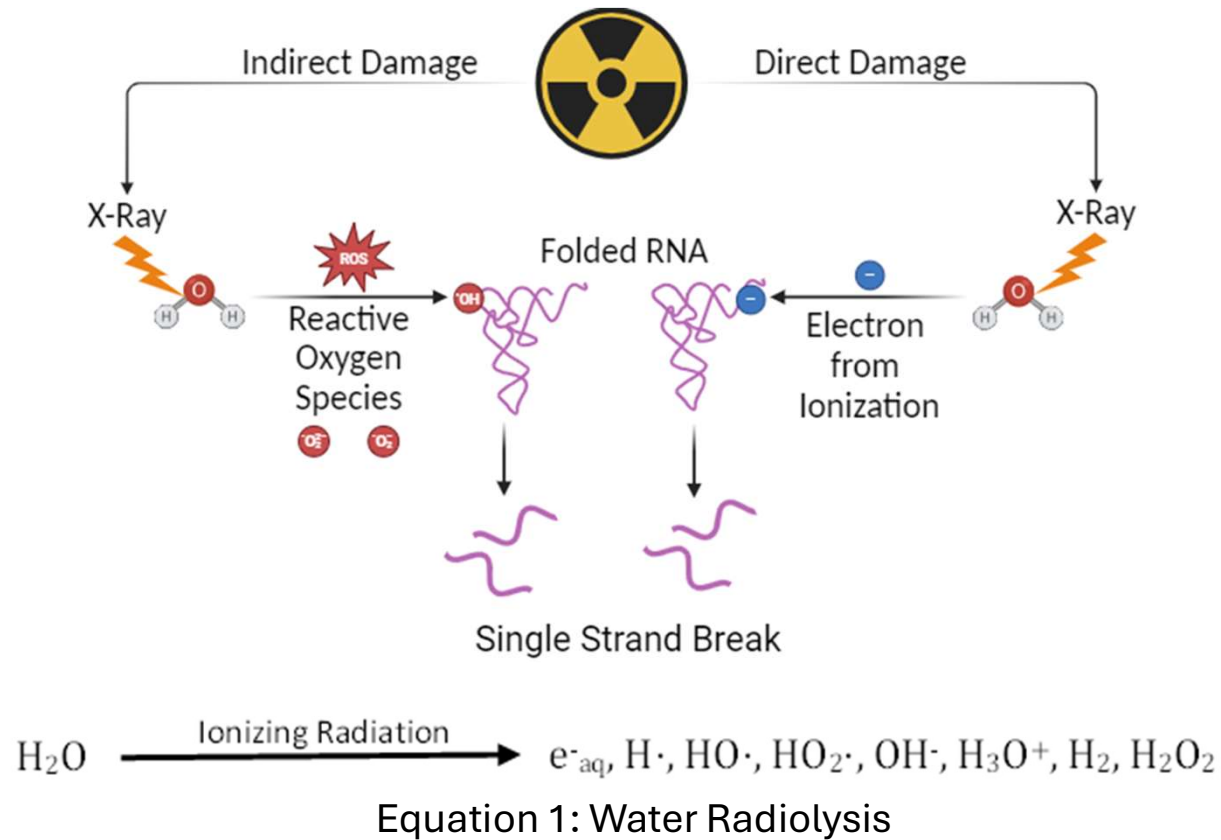


- New reactants formed in spur if M is a nearest neighbor

Potential Experiment 2: Radiation Damage to DNA/RNA

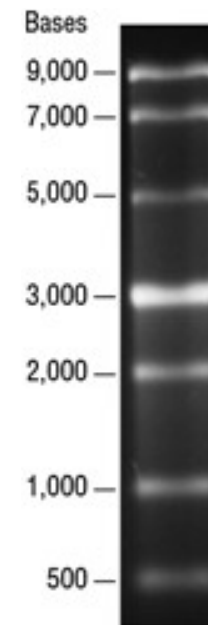


Ionising radiation has two main damage pathways



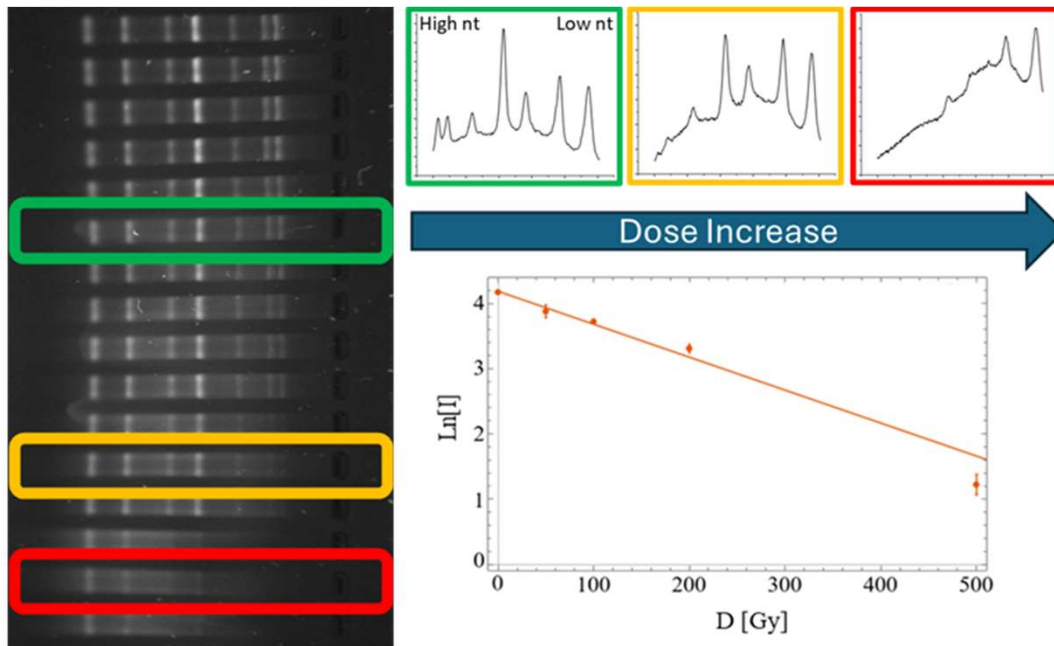
RNA Sample

- Size range- 500 bases to 9,000 bases.
 - 9000, 7000, 5000, 3000, 2000, 1000 & 500.
- 3000 base fragment is twice as intense to serve as reference.
- Storage buffer:
 - 20 mM sodium citrate, 1 mM EDTA, pH 6 at 25°C.



Single stranded RNA Ladder

Method Outline



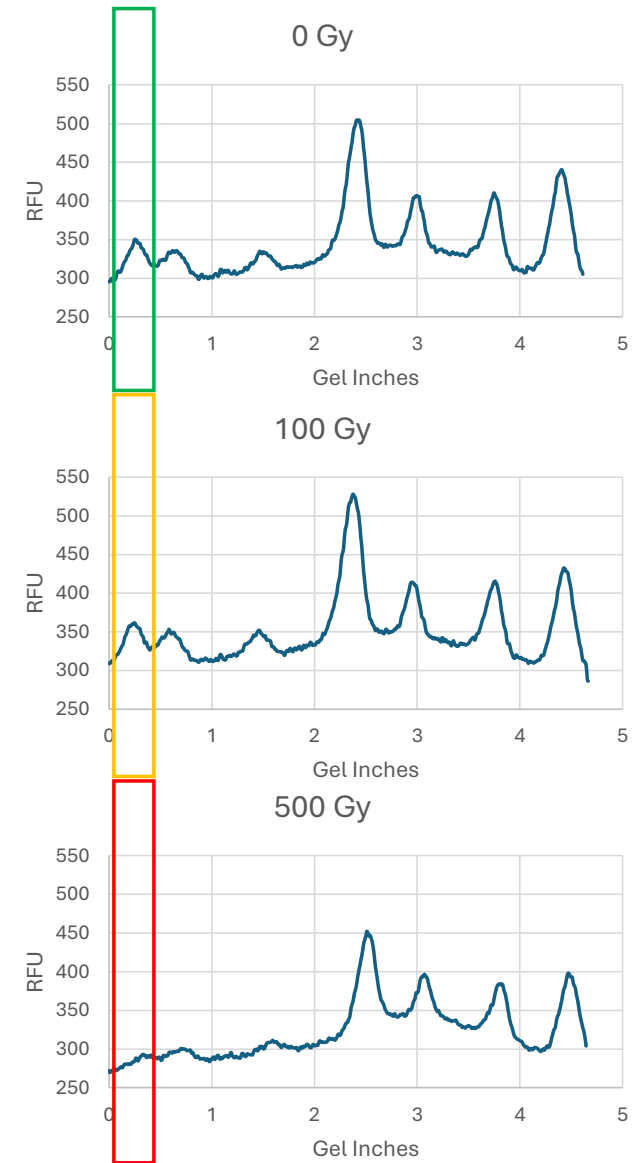
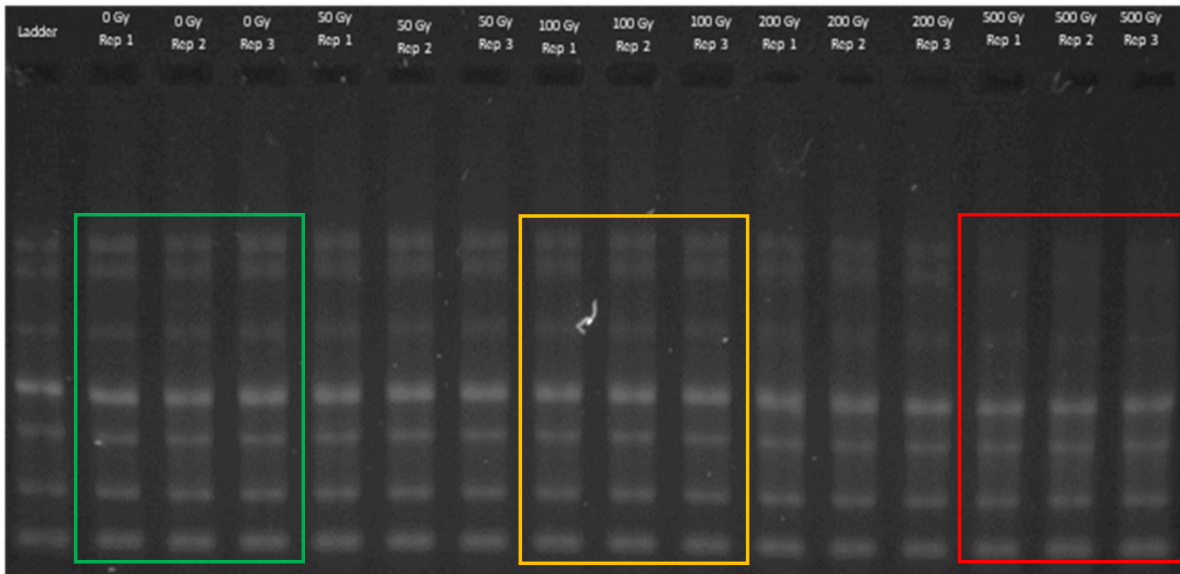
- The RNA and buffers are prepared individually and then mixed.
- The RNA samples are irradiated:



- After irradiation, the RNA samples are collected and prepared for gel electrophoresis.
- The images collected from gel electrophoresis are analysed via ImageJ and a bespoke Mathematica program.

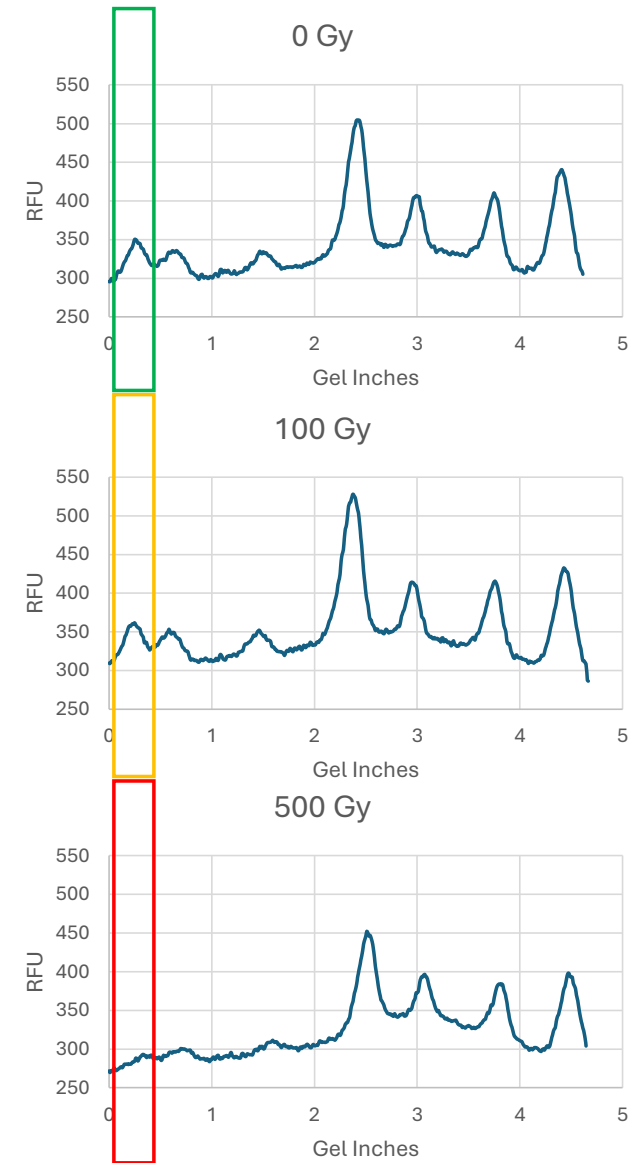
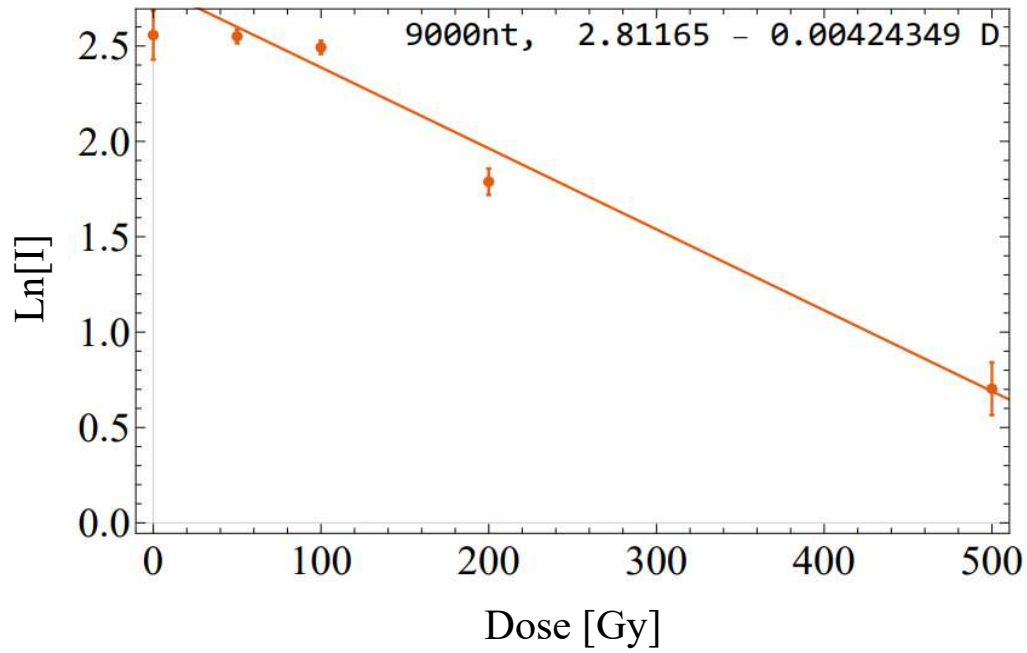
Method

Gel analysis



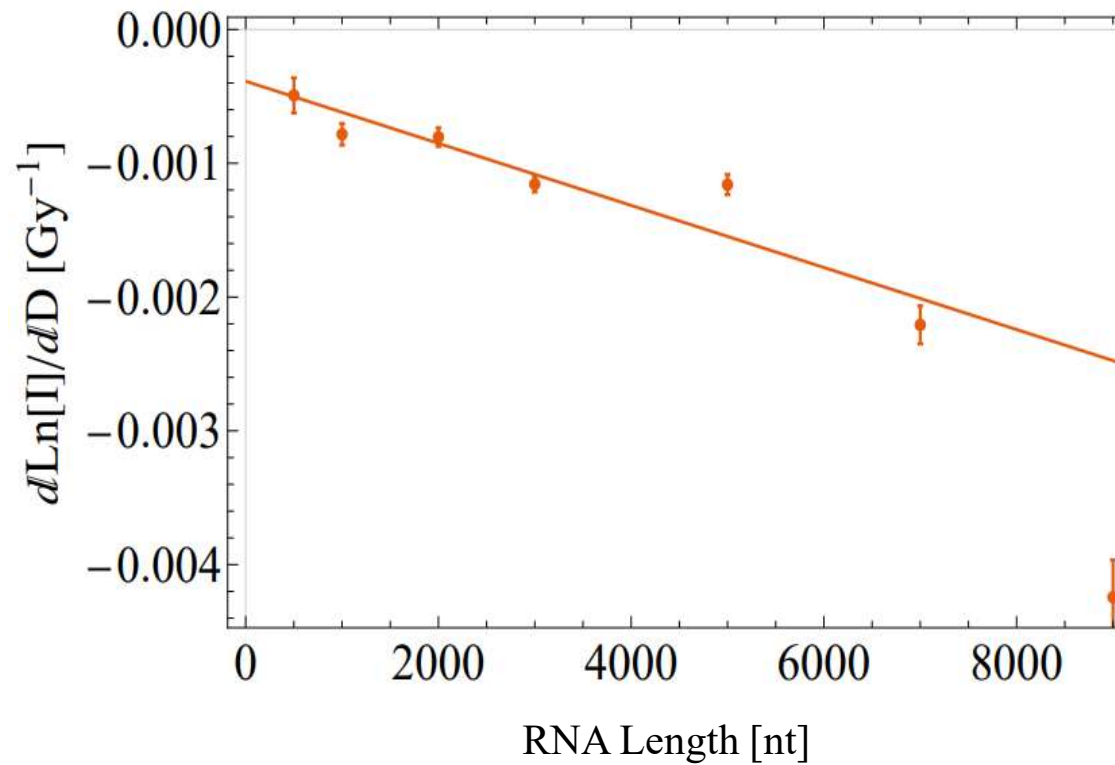
Method

Gel analysis



* Ongoing PhD project at the University of Manchester (PhD candidate - Jordan Elliot)

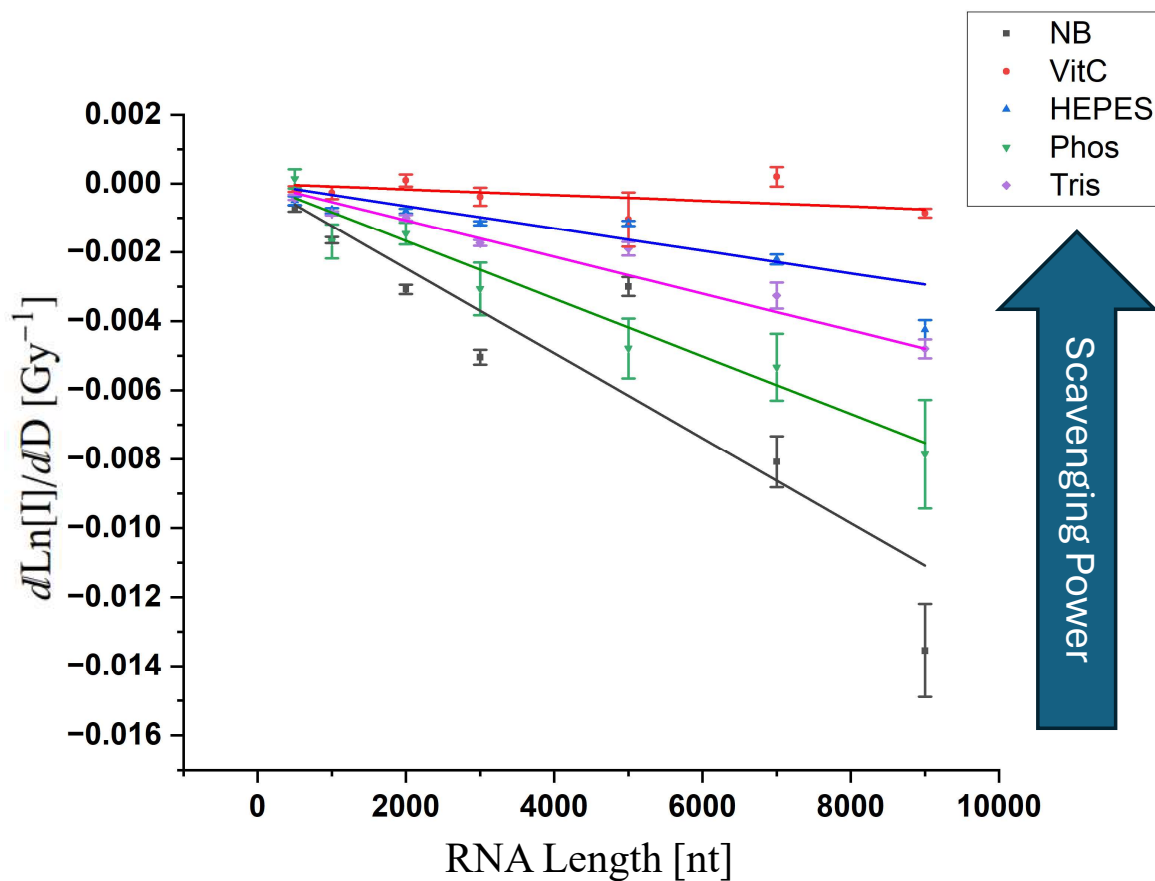
Gel analysis



Results

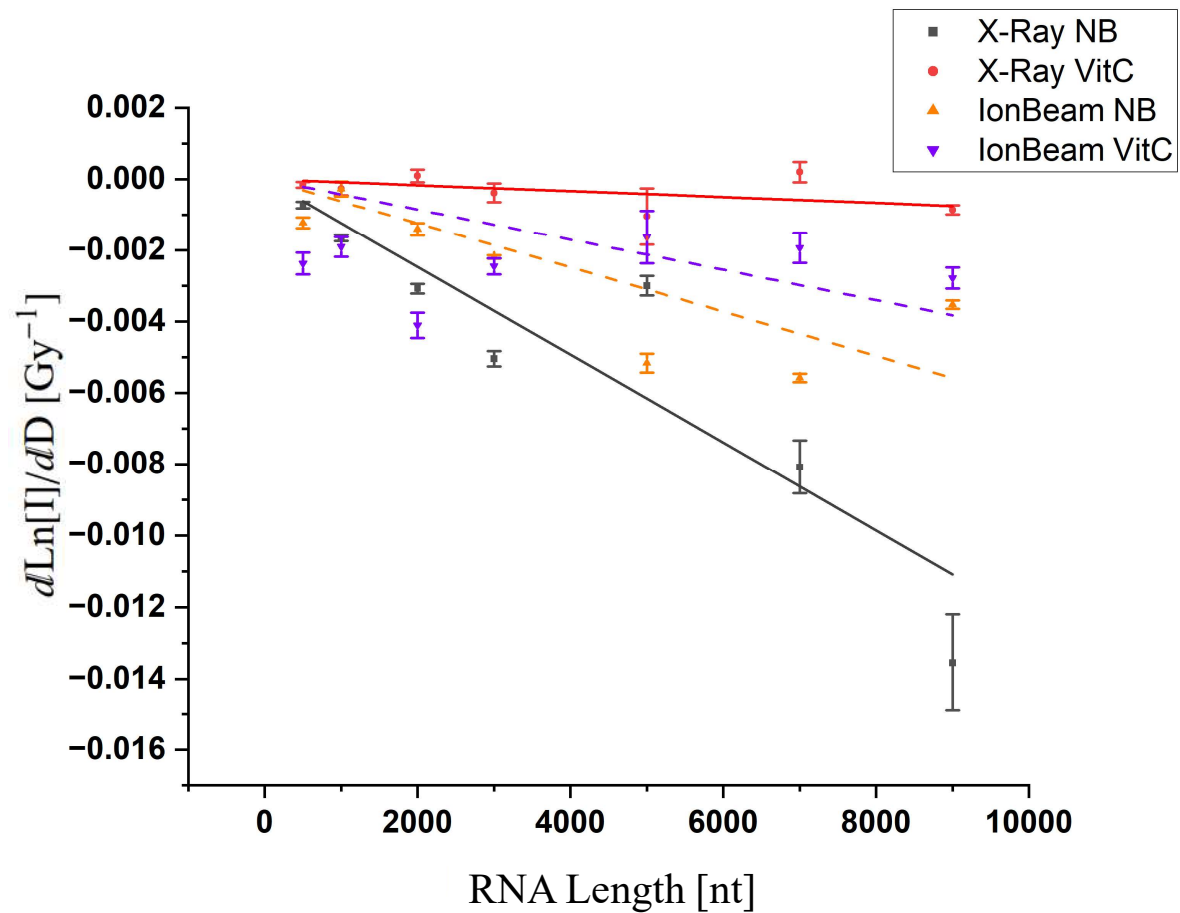
Comparison of RNA damage in different scavenger buffers

Buffer	Rate constant for OH· radicals ($L mol^{-1}s^{-1}$)
Ascorbic Acid	1×10^{10}
HEPES	6.9×10^9
Tris	1.5×10^9
Phosphate	$<1 \times 10^7$



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Comparison of RNA Damage: X-Ray vs Ion Beam

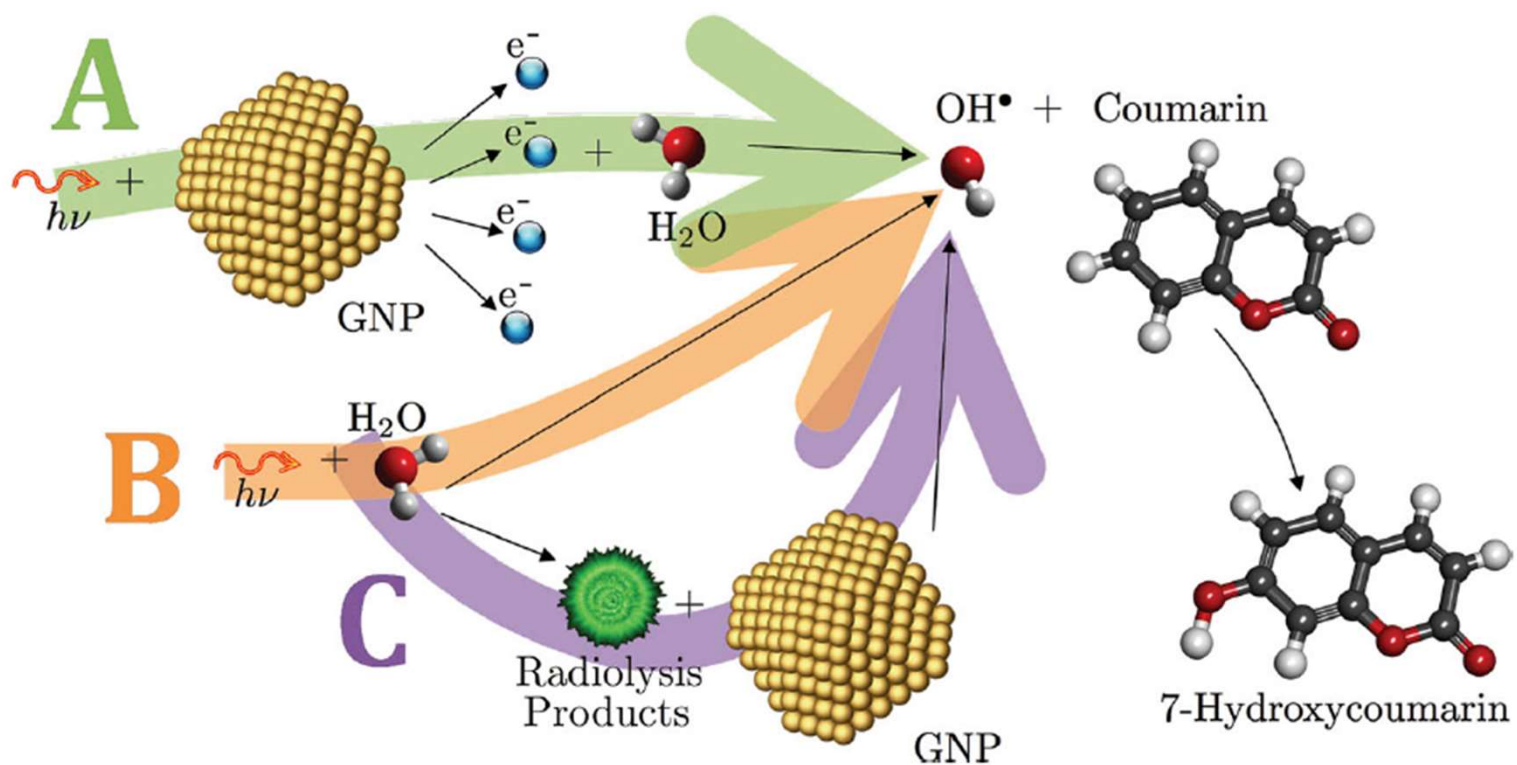


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Open Questions

- How do VHEE damage RNA/DNA in comparison to photons, ions and lower energy electrons?
- How does the efficiency of antioxidant protection against VHEE induced damage compares to that of other radiation types?
- Dose rate effects?
- Oxygen depletion during irradiation.

Potential Experiment 3: Radiolytic Processes on Nanoparticle Interfaces



Summary

- Radiolysis is complex, interesting and longstanding problem;
- Ultrafast experiments will allow to observe and control interspur dynamics;
- Insights into mechanisms in solution and at interfaces;
- Three potential radiation chemistry experiments proposed:
 - (a) Water radiolysis under extreme pH (strongly alkaline vs. acidic); **requires time-resolved detection**
 - (b) Radiation damage to DNA/RNA and evaluation of antioxidant protection; **doesn't need time-resolved detection**
 - (c) Radiation effects on NP interfaces; **doesn't need time-resolved detection but could benefit from it.**